

**Table 1. Essential nutrient elements showing element, symbol and primary forms used by plants.**

Element	Symbol	Primary Forms Used by Plants
<b>NON-MINERAL ELEMENTS</b>		
Carbon	C	CO <sub>2</sub> (g)
Hydrogen	H	H <sub>2</sub> O (l), H <sup>+</sup>
Oxygen	O	H <sub>2</sub> O (l), O <sub>2</sub> (g)
<b>MINERAL ELEMENTS</b>		
<b>Major Nutrients</b>		
Nitrogen	N	NH <sub>4</sub> <sup>+</sup> , NO <sub>3</sub> <sup>-</sup>
Phosphorous	P	HPO <sub>4</sub> <sup>2-</sup> , H <sub>2</sub> PO <sub>4</sub> <sup>-</sup>
Potassium	K	K <sup>+</sup>
<b>Secondary Nutrients</b>		
Calcium	Ca	Ca <sup>2+</sup>
Magnesium	Mg	Mg <sup>2+</sup>
Sulfur	S	SO <sub>4</sub> <sup>2-</sup>
<b>Micronutrients</b>		
Iron	Fe	Fe <sup>3+</sup> , Fe <sup>2+</sup>
Manganese	Mn	Mn <sup>2+</sup>
Zinc	Zn	Zn <sup>2+</sup>
Copper	Cu	Cu <sup>2+</sup>
Boron	B	B(OH) <sub>3</sub> (Boric acid)
Molybdenum	Mo	MoO <sub>4</sub> <sup>2-</sup>
Chlorine	Cl	Cl <sup>-</sup>

Productivity depends on soil microbes

Decomposition depends on soil microbes

Plants require at least 16 elements for normal growth and for completion of their life cycle.

Those used in the largest amounts, carbon, hydrogen and oxygen, are non-mineral elements supplied by air and water.

The other 13 elements are taken up by plants only in mineral form from the soil or must be added as fertilizers.

Minerals must be available, continuously, and in balanced proportions to support photosynthesis and other metabolic processes of plants.

If **any** one of these essential elements is missing, plant productivity will be limited or will cease entirely.

**Principle of limiting factors:**  
The level of production can be no greater than that allowed by the most limiting of the essential plant growth factors.

Applies in both cropping systems and in natural ecosystems.

Common limiting factors:

N,P,K: Primary nutrients - plants need relatively large amounts. Frequently supplied in fertilizers.

Ca, Mg, S: Secondary nutrients - required in smaller amounts

Plants absorb the essential elements through their root systems or their leaves in various forms. In general, the soil contains large amounts of all the elements, but only a very small percentage of these total amounts are actually **plant-available**.

For example, the actual total iron content of a soil may exceed 50,000 parts per million (ppm), however the portion available to plants may be less than 5 ppm.

The availability of nutrients to plants is determined by:

- the form and chemical properties of the element
- the soil pH
- interactions with soil colloids
- soil physical conditions such as aeration, compaction, temperature, and moisture --

**microbial activity**

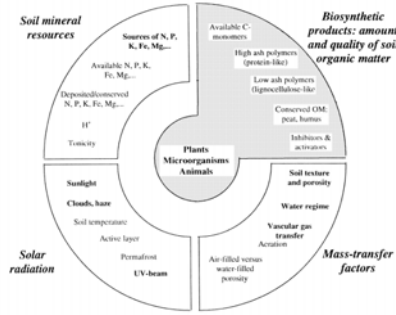
- Carbon cycle
- Nitrogen cycle
- Sulfur cycle
- Phosphorus availability
- Iron availability
- Etc.

**Two perspectives on soil microbial communities:**

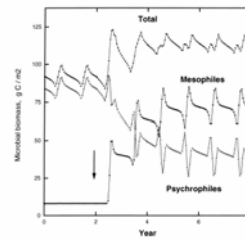
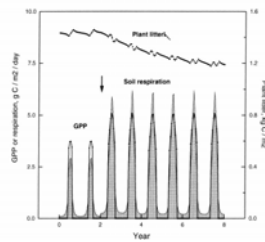
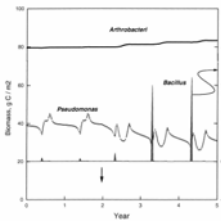
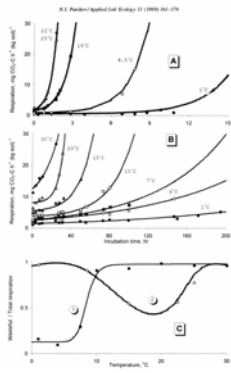
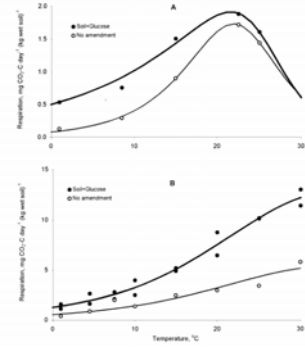
1. Systems view (nutrient cycling as a whole)
2. Single-process view (suppressive soil due to single antibiotic)

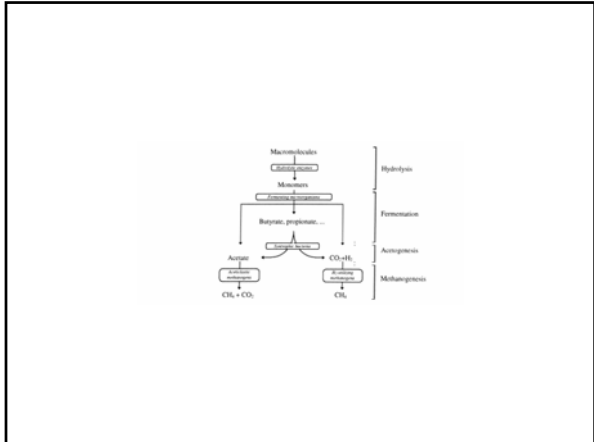
**Both are important!**

# Environmental factors affecting soil communities



N.S. Panikov / Applied Soil Ecology 11 (1999) 161-176



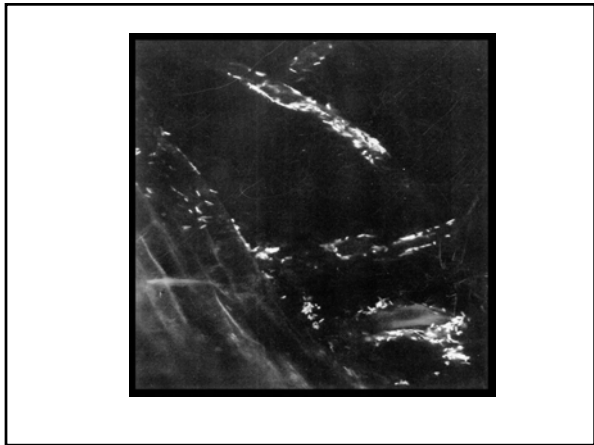


### Suppressive soils

Fluorescent pseudomonads suppress *Fusarium oxysporum* (Fusarium wilt of many plants); iron competition, ISR, and direct inhibition of fungal growth


Fluorescent pseudomonads suppress *Gaeumannomyces graminis* var. *tritici* (take all of wheat) - 2,4-DAPG

*Trichoderma* inhibits *Rhizoctonia solani* root rots - parasitism




### Biological control of *Pythium* damping-off of cotton by *Pseudomonas fluorescens*


### Known interacting antibiotics produced by *Pseudomonas*



**Pyoluteorin (PLT)**  
Oomycete pathogens – *P. ultimum*



**2,4-diacetylphloroglucinol (2,4-DAPG)**  
Broad spectrum: *Fusarium*, *Gaeumannomyces*, etc.



**Pyrrolnitrin (PRN)**  
*Rhizoctonia*, *Pyrenophora*, *Sclerotinia*

