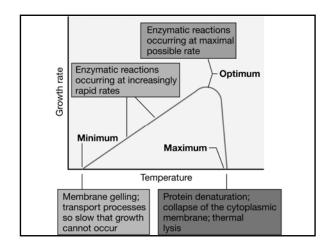
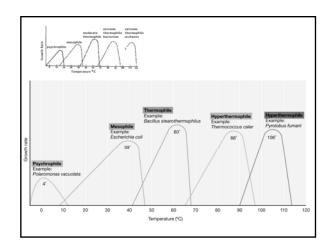
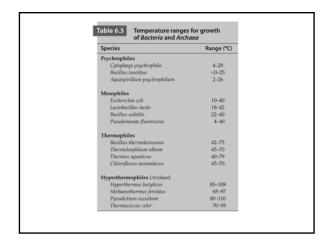
## Microbial Growth

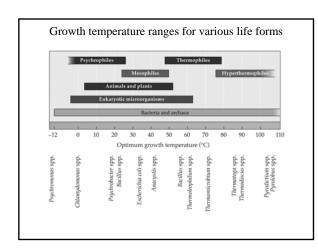
#### **Environmental Forcing Functions:**

- Temperature: Psychrophile, Mesophile, Thermophile, & Hyperthermophile
   Cardinal Temps: Min\*, Max, & Optimal\*
   Q<sub>10</sub> Rule: 10°C rise will double the growth rate\*
- Pressure: Barophiles (Most are also psychrophiles!)
  Found only in the deep ocean.....so far

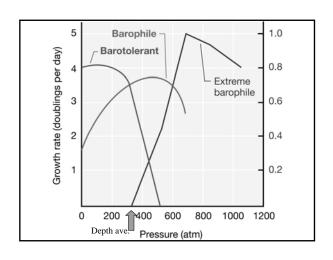








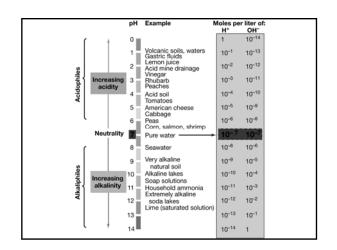


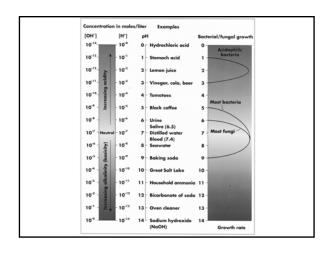


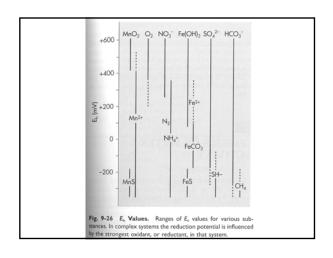
# Microbial Growth

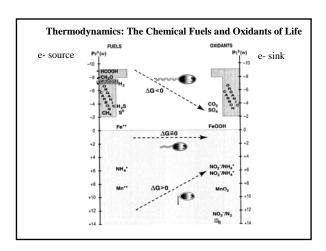
#### **Environmental Forcing Functions:**

- pH: acidophiles & alkaliphiles cytoplasm still near neutral
- eH: available electron donors & terminal electron acceptors
  - affects the chemistry of the environment





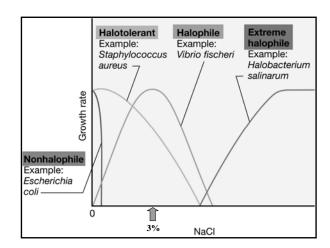


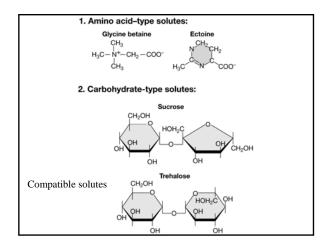


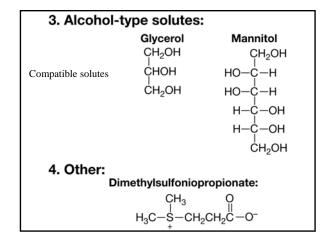
### Microbial Growth

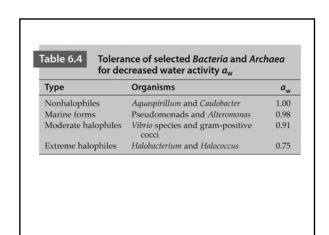
### **Environmental Forcing Functions:**

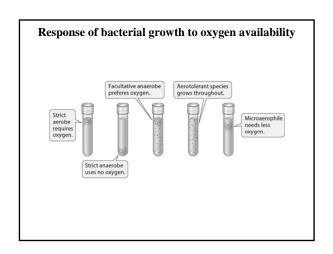
- Salt: Halophiles
   Compatible solutes: amino acid derivatives (e.g., proline & glycine)
- Water Activity: Xerophiles (live in very dry habitats)
   All microbes are osmotrophs, must use organic material in solution!
- Oxygen Usage: aerobe, facultative (an)aerobe, microaerophile, obligate anaerobe
   DeTox enzymes: Catalase, Peroxidase, SOD

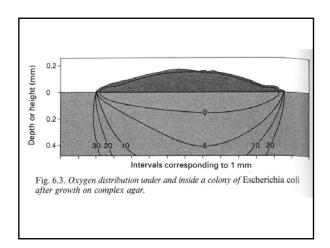








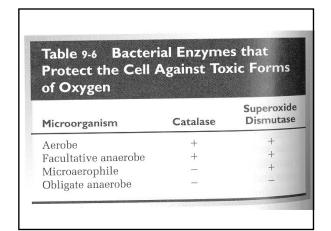






| Table 9-5 Electr                               | onic S           | states o                              | f Oxygen                      |
|--|------------------|---------------------------------------|-------------------------------|
| Form   | Formula          | Simplified<br>Electronic<br>Structure | Spin of<br>Outer<br>Electrons |
| Triplet oxygen<br>(normal<br>atmospheric form) | $^3\mathrm{O}_2$ | Ó—Ó                                   | 1 1                           |
| Singlet oxygen  Nasty!                         | $^{1}O_{2}$      | Ó—Ó                                   | ① ① ①                         |
| Superoxide free<br>radical                     | $O_2^-$          | Ö—Ö                                   | (I) (I)                       |
| Peroxide                                       | $O_2^{2-}$       | Ö—Ö                                   | (1) (1)                       |

$$O_2 + e^- \rightarrow O_2^-$$
 Superoxide  
 $O_2^- + e^- + 2 H^+ \rightarrow H_2O_2$  Hydrogen peroxide  
 $H_2O_2 + e^- + H^+ \rightarrow H_2O + OH^{\bullet}$  Hydroxyl radical  
 $OH^{\bullet} + e^- + H^+ \rightarrow H_2O$  Water  
Overall:  $O_2 + 4 e^- + 4 H^+ \rightarrow 2 H_2O$   
4 electron reduction of  $O_2$  to water



(a) Catalase:  

$$H_2O_2 + H_2O_2 \rightarrow 2 H_2O + O_2$$
  
(b) Peroxidase:  
 $H_2O_2 + NADH + H^+ \rightarrow 2 H_2O + NAD^+$   
(c) Superoxide dismutase:  
 $O_2^- + O_2^- + 2 H^+ \rightarrow H_2O_2 + O_2$   
(d) Superoxide dismutase/catalase in combination:  
 $4 O_2^- + 4 H^+ \rightarrow 2 H_2O + 3 O_2$   
(e) Superoxide reductase:  
 $O_2^- + 2 H^+ + cyt c_{reduced} \rightarrow H_2O_2 + cyt c_{oxidized}$ 

