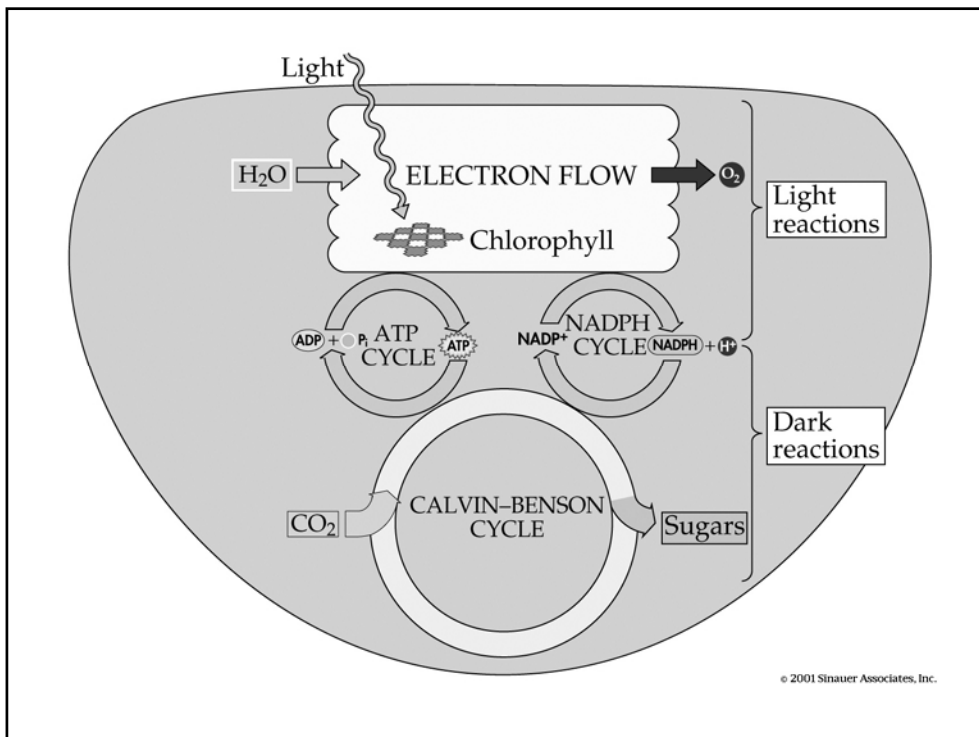


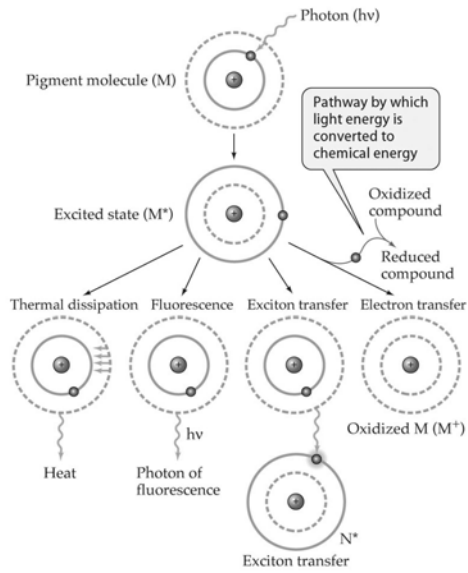
Steven J. Schmitt and M.T. Madigan

(b)



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## The possible fates of an excited electron

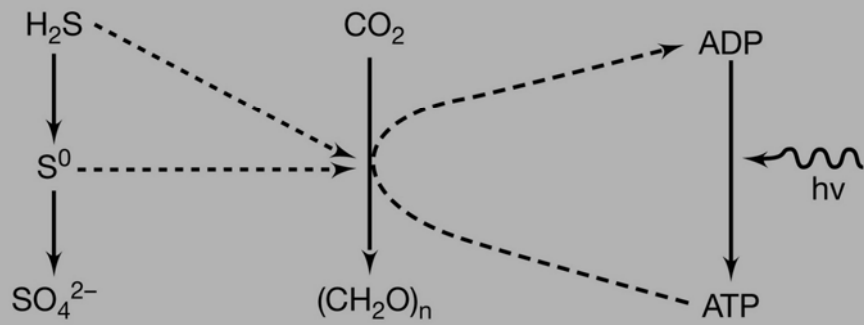


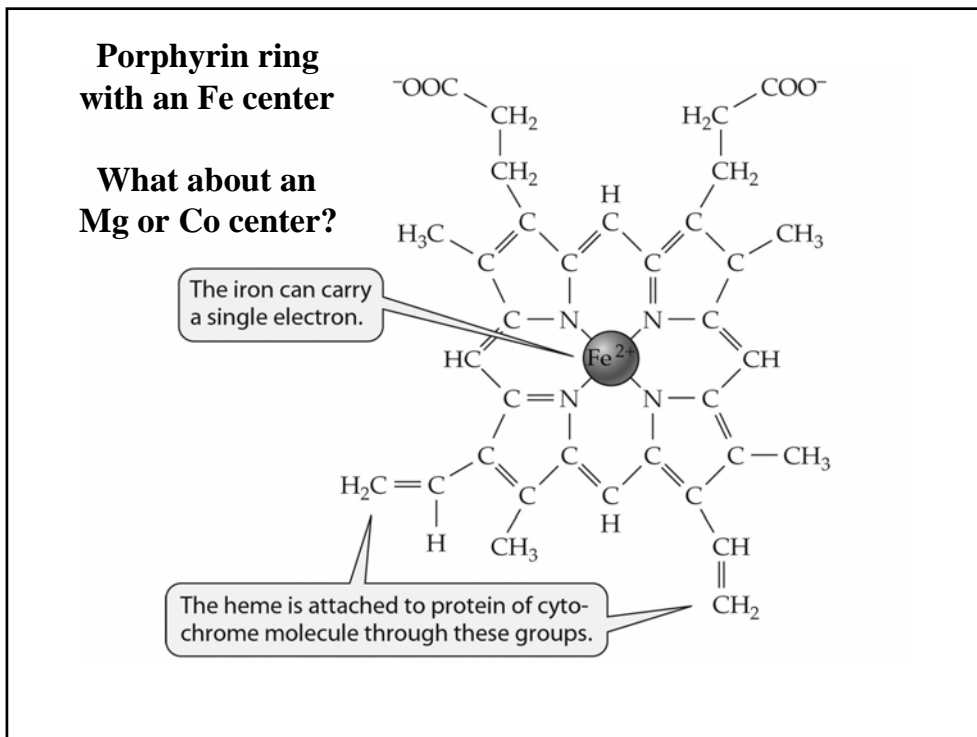
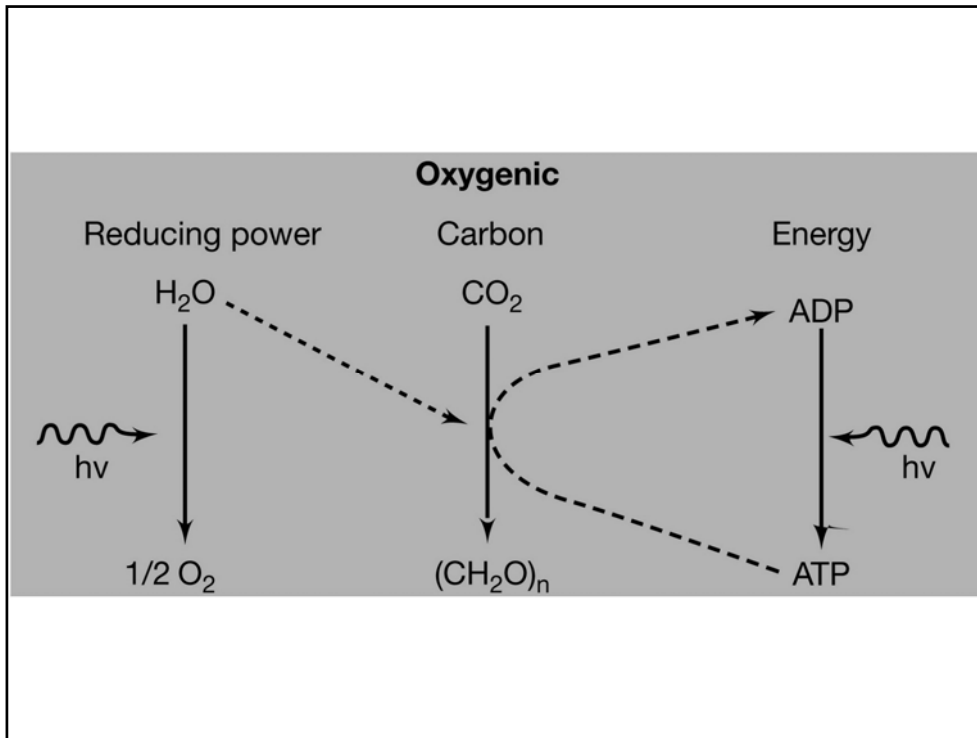
## Anoxygenic

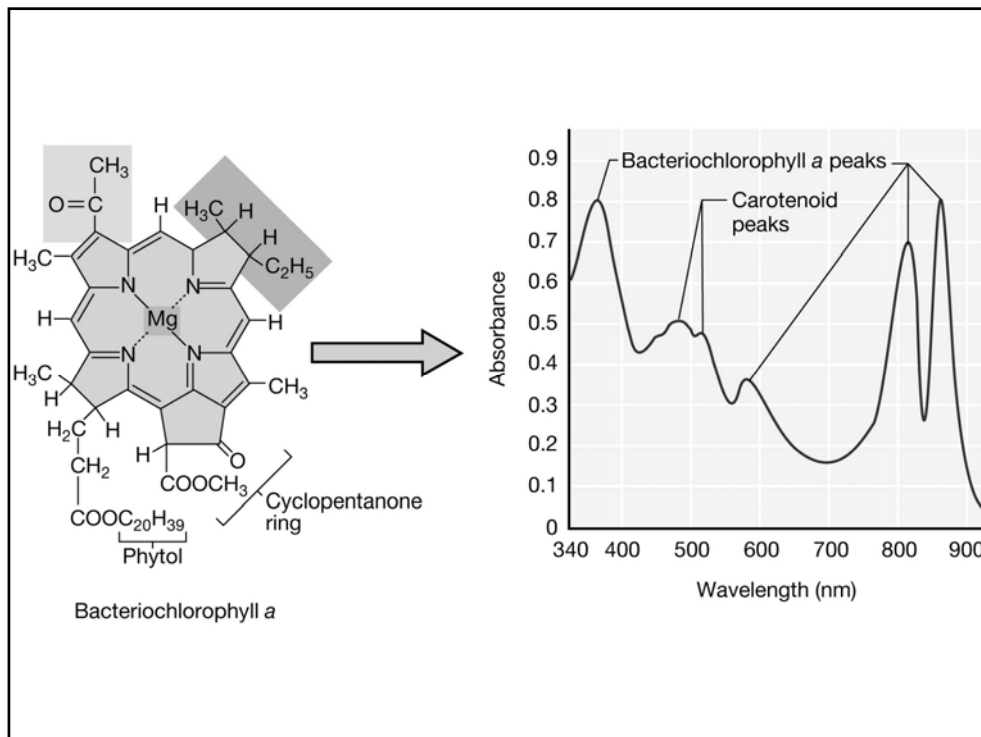
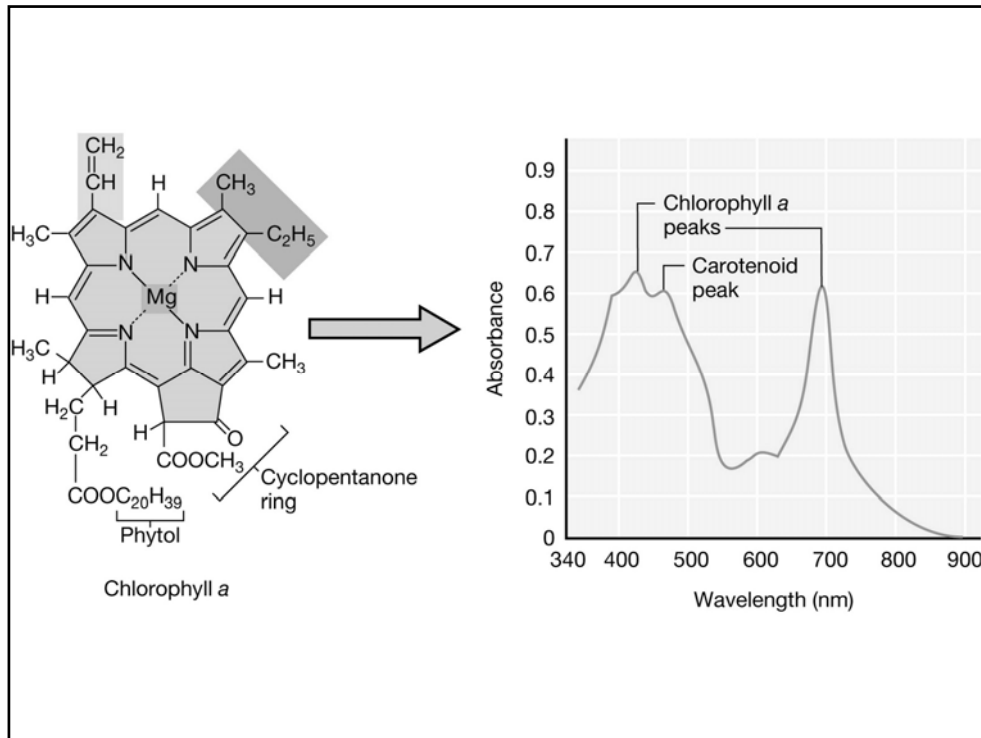
Reducing power

Carbon

Energy

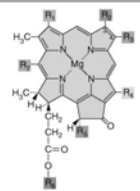






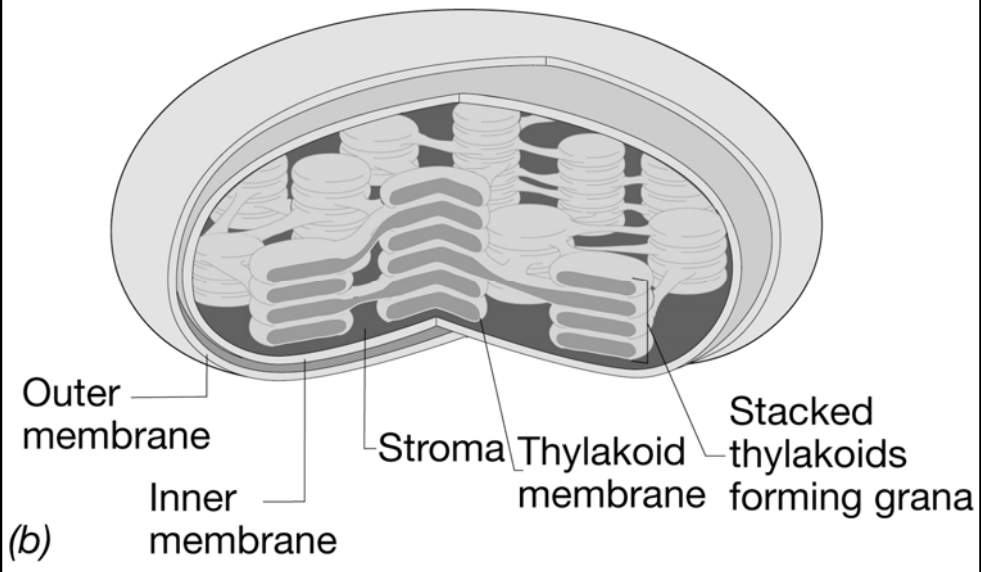
Pigment	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	R <sub>7</sub>	Absorption maxima (nm)	
								In vivo	Extract (methanol)
Bacteriochlorophyll <i>a</i> (purple bacteria)	$\begin{array}{c} \text{—C—CH}_3 \\   \\ \text{O} \end{array}$	$\text{—CH}_3^b$	$\text{—CH}_2\text{—CH}_3$	$\text{—CH}_3$	$\begin{array}{c} \text{—C—O—CH}_3 \\   \\ \text{O} \end{array}$	P/Cg <sup>a</sup> —H		805 830-890	771
Bacteriochlorophyll <i>b</i> (purple bacteria)	$\begin{array}{c} \text{—C—CH}_3 \\   \\ \text{O} \end{array}$	$\text{—CH}_3^b$	$\begin{array}{c} \text{=C—CH}_3 \\   \\ \text{H} \end{array}$	$\text{—CH}_3$	$\begin{array}{c} \text{—C—O—CH}_3 \\   \\ \text{O} \end{array}$	P	—H	835-850 1020-1040	794
Bacteriochlorophyll <i>c</i> (green sulfur bacteria)	$\begin{array}{c} \text{H} \\   \\ \text{—C—CH}_3 \\   \\ \text{OH} \end{array}$	$\text{—CH}_3$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\   \\ \text{—C}_2\text{H}_5^d \\   \\ \text{—C}_4\text{H}_9 \end{array}$	$\text{—C}_2\text{H}_5$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\   \\ \text{—C}_2\text{H}_5 \\   \\ \text{—CH}_3 \end{array}$	F	—CH <sub>3</sub>	745-755	660-669
Bacteriochlorophyll <i>c</i> <sub>2</sub> (green nonsulfur bacteria)	$\begin{array}{c} \text{H} \\   \\ \text{—C—CH}_3 \\   \\ \text{OH} \end{array}$	$\text{—CH}_3$	$\text{—C}_2\text{H}_5$	$\text{—CH}_3$	—H	S	—CH <sub>3</sub>	740	667
Bacteriochlorophyll <i>d</i> (green sulfur bacteria)	$\begin{array}{c} \text{H} \\   \\ \text{—C—CH}_3 \\   \\ \text{OH} \end{array}$	$\text{—CH}_3$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\   \\ \text{—C}_2\text{H}_5 \\   \\ \text{—C}_4\text{H}_9 \end{array}$	$\text{—C}_2\text{H}_5$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\   \\ \text{—C}_2\text{H}_5 \\   \\ \text{—CH}_3 \end{array}$	F	—H	705-740	654
Bacteriochlorophyll <i>e</i> (green sulfur bacteria)	$\begin{array}{c} \text{H} \\   \\ \text{—C—CH}_3 \\   \\ \text{OH} \end{array}$	$\begin{array}{c} \text{—C—H} \\   \\ \text{O} \end{array}$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\   \\ \text{—C}_2\text{H}_5 \\   \\ \text{—C}_4\text{H}_9 \end{array}$	$\text{—C}_2\text{H}_5$	—H	F	—CH <sub>3</sub>	719-726	646
Bacteriochlorophyll <i>g</i> (helicobacteria)	$\begin{array}{c} \text{H} \\   \\ \text{—C=CH}_2 \end{array}$	$\text{—CH}_3^b$	$\text{—C}_2\text{H}_5$	$\text{—CH}_3$	$\begin{array}{c} \text{—C—O—CH}_3 \\   \\ \text{O} \end{array}$	F	—H	670, 788	765

<sup>a</sup>P, Phytol ester (C<sub>20</sub>H<sub>39</sub>O—); F, farnesyl ester (C<sub>15</sub>H<sub>27</sub>O—); Gg, geranylgeraniol ester (C<sub>20</sub>H<sub>37</sub>O—); S, stearyl alcohol (C<sub>18</sub>H<sub>37</sub>O—).  
<sup>b</sup>No double bond between C<sub>5</sub> and C<sub>6</sub>; additional H atoms are in positions C<sub>7</sub> and C<sub>8</sub>.  
<sup>c</sup>No double bond between C<sub>5</sub> and C<sub>6</sub>; an additional H atom is in position C<sub>7</sub>.  
<sup>d</sup>Bacteriochlorophylls *c*, *d*, and *e* consist of isomeric mixtures with the different substituents on R<sub>3</sub> as shown.



Bacteriochlorophyll Structures

### Chloroplast Structure



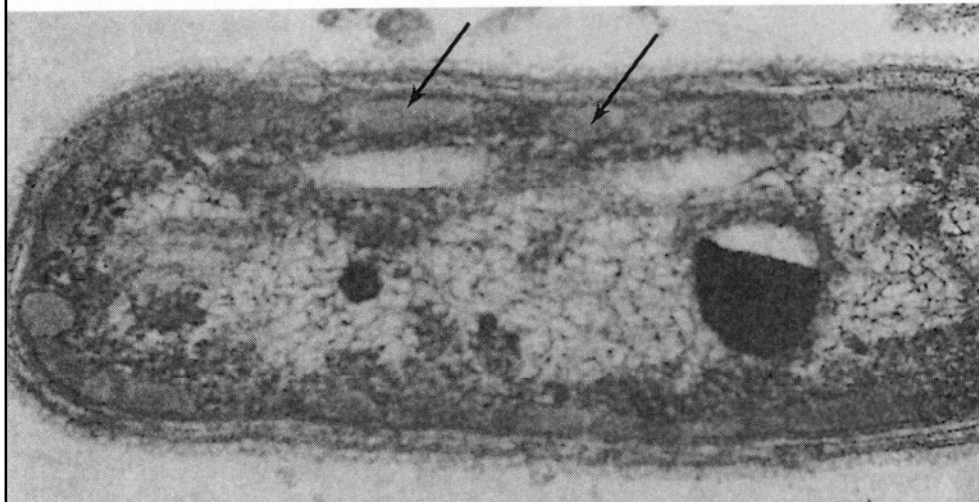
**Table 9.1** Some general properties of the various photosynthetic bacteria

	Nonsulfur Purple Bacteria	Purple Sulfur Bacteria	Green Sulfur Bacteria	Cyano-bacteria	Helio-bacteria
Source of reducing power ( $e^-$ )	$H_2$ , reduced organic	$H_2S$	$H_2S$	$H_2O$	Lactate, organic
Oxidized product	Oxidized organic	$SO_4^{2-}$	$SO_4^{2-}$	$O_2$	Oxidized organic
Source of carbon	$CO_2$ or organic	$CO_2$	$CO_2$	$CO_2$	Lactate pyruvate
Heterotrophic growth	Common	Limited <sup>a</sup>	Limited <sup>a</sup>	Limited <sup>a</sup>	Required

<sup>a</sup>Generally limited to assimilation of low molecular weight organics during autotrophic growth.

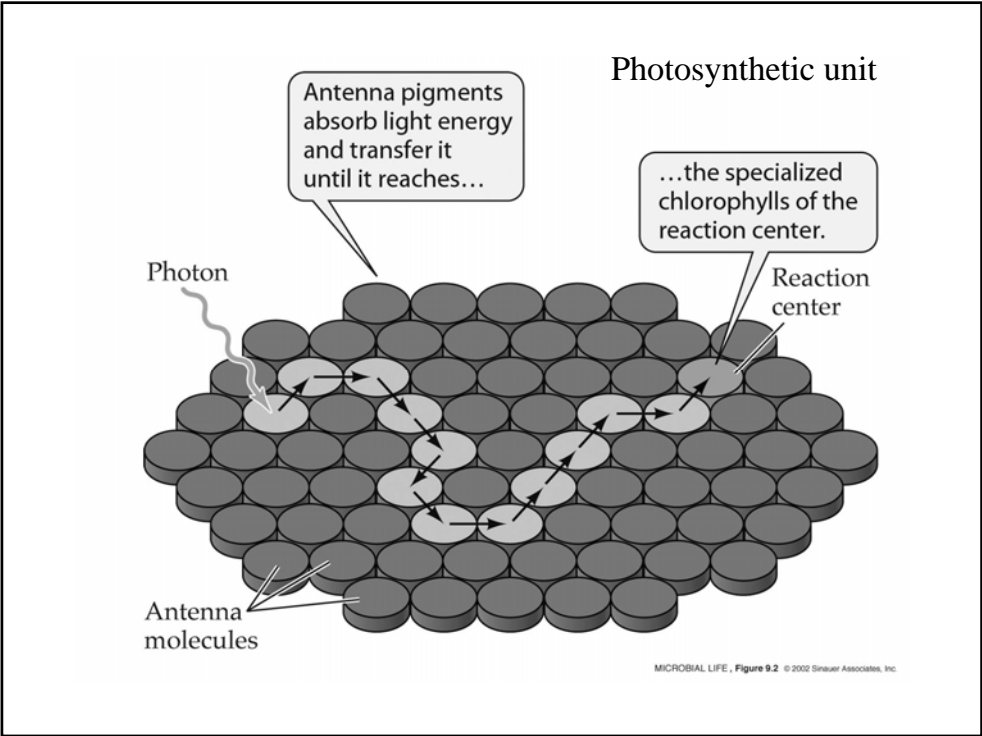
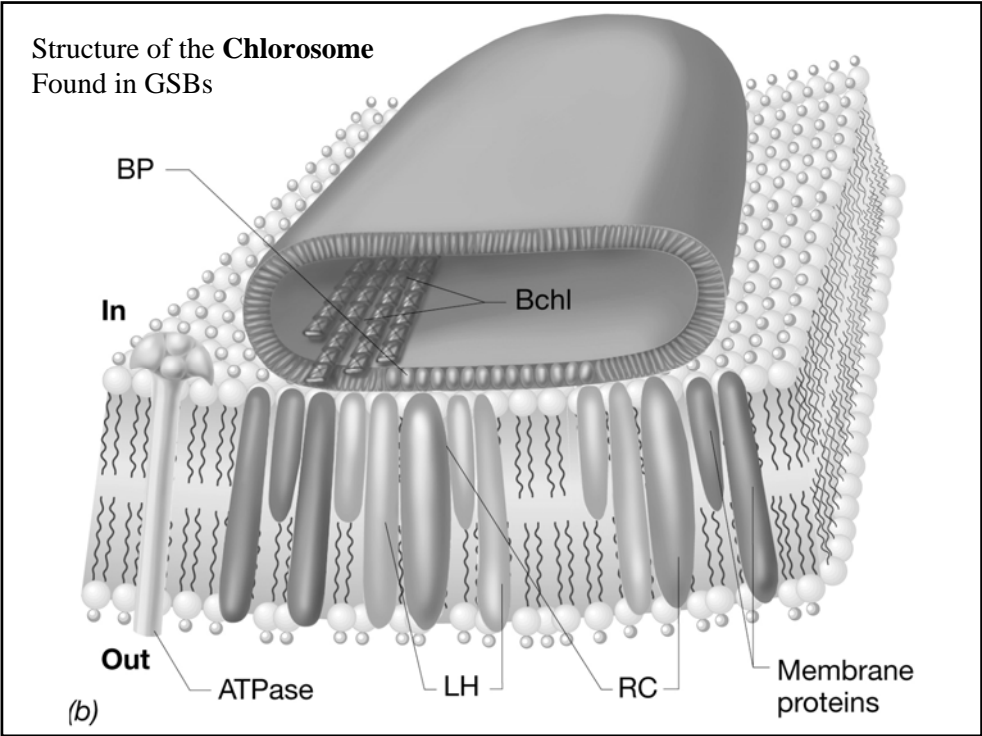
MICROBIAL LIFE, Table 9.1 © 2002 Sinauer Associates, Inc.

### Structure and Location of the Chlorosome



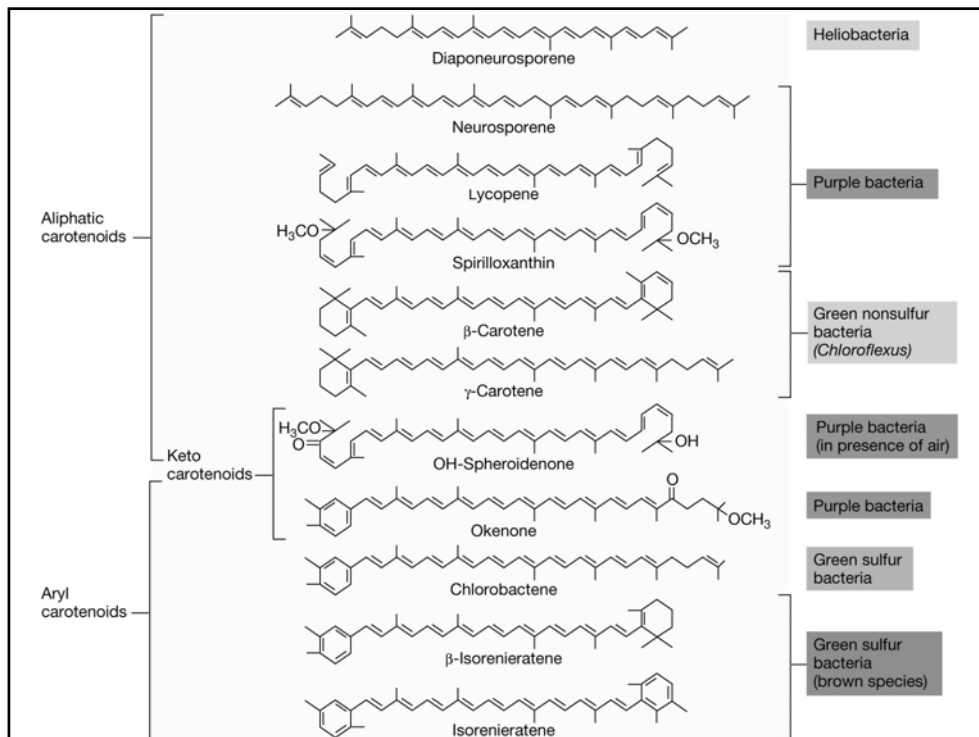
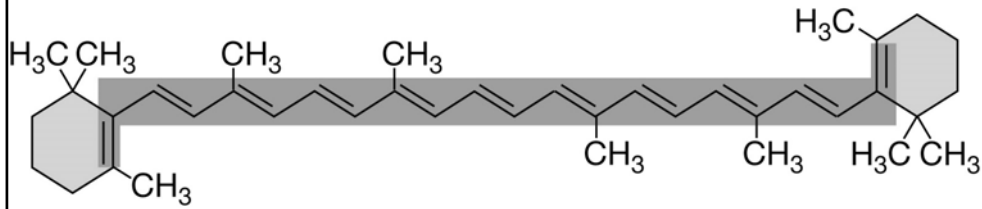
(a)

Found in GSBs





## Beta-Carotene, a typical carotenoid

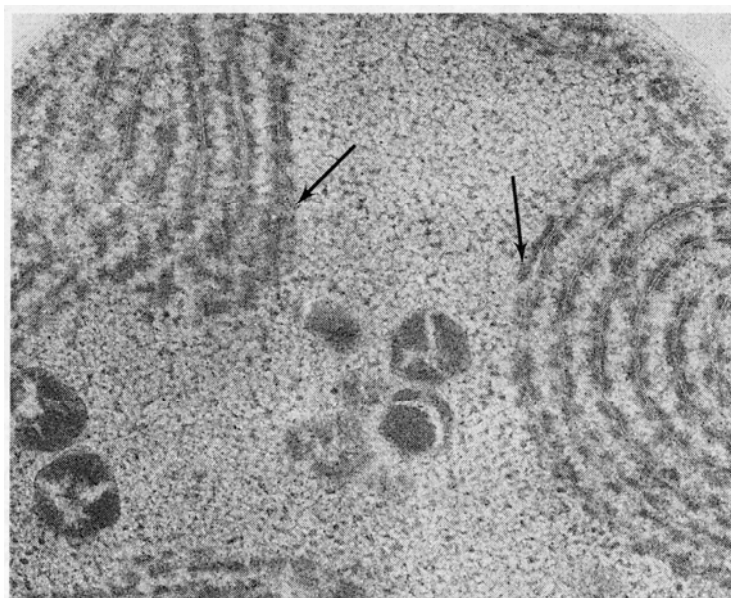


**Table 9.2** The bacteriochlorophyll present in photosynthetic bacteria and primary acceptors involved in energy conserving reactions

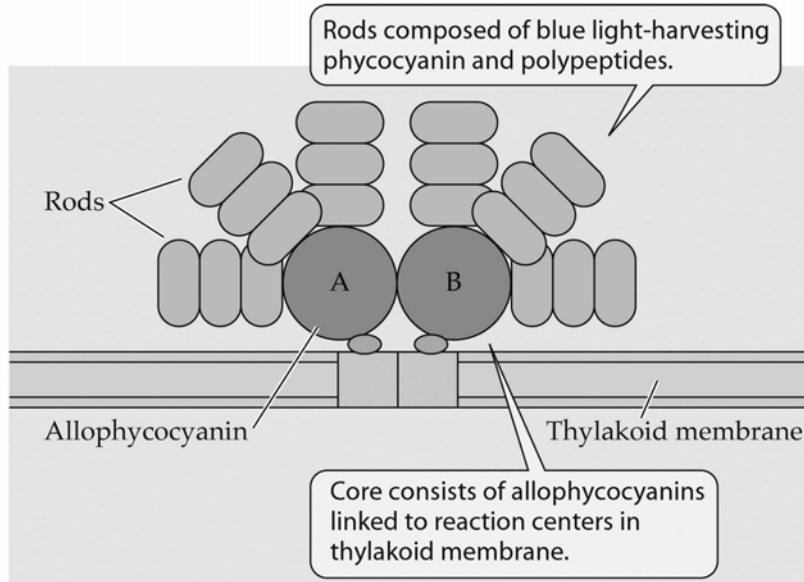
	Electron Donor	Electron Acceptor
Purple nonsulfur bacteria	Bacteriochlorophyll <i>a</i> and <i>b</i>	Bacteriopheophytin <i>a</i> , $Q_{A'}$ , and $Q_B$
Green sulfur bacteria	Bacteriochlorophyll <i>c</i> , <i>d</i> , and <i>e</i>	Bacteriopheophytin <i>a</i> and FeS-protein
Cyanobacteria photosystem I	Chlorophyll <i>a</i>	Chlorophyll <i>a</i> and FeS-protein
Cyanobacteria photosystem II	Chlorophyll <i>a</i>	Pheophytin <i>a</i> , $Q_{A'}$ , $Q_{B'}$ , and plastoquinones
<i>Heliobacteria</i>	Bacteriochlorophyll <i>g</i>	Bacteriochlorophyll <i>c</i> and FeS-protein

MICROBIAL LIFE, Table 9.2 © 2002 Sinauer Associates, Inc.

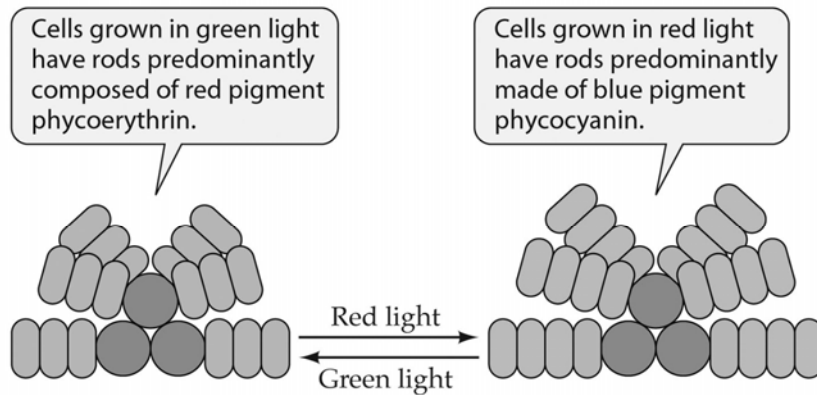
### Structure and Location of **Phycobilisomes**



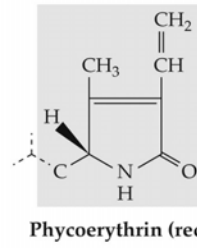
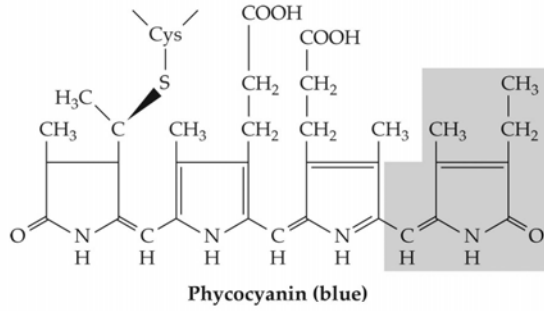
### Phycobilisome of cyanobacteria



### Chromatic adaptation of a phycobilisome

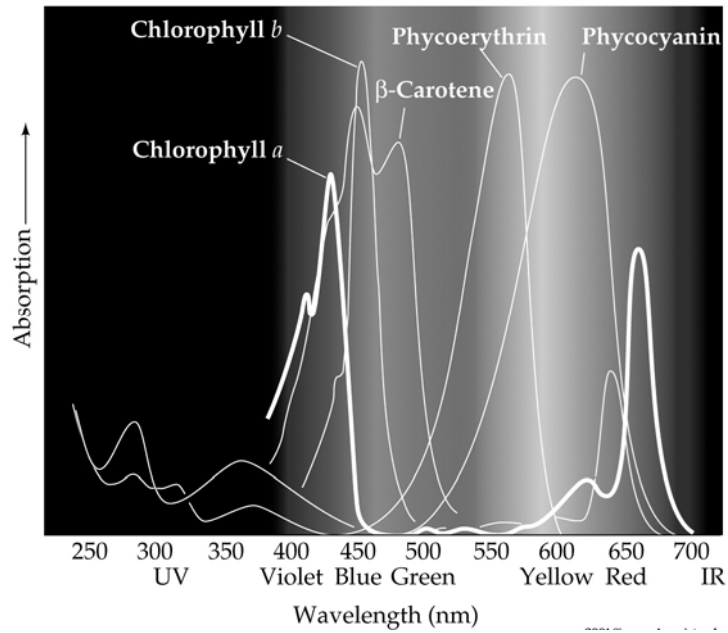


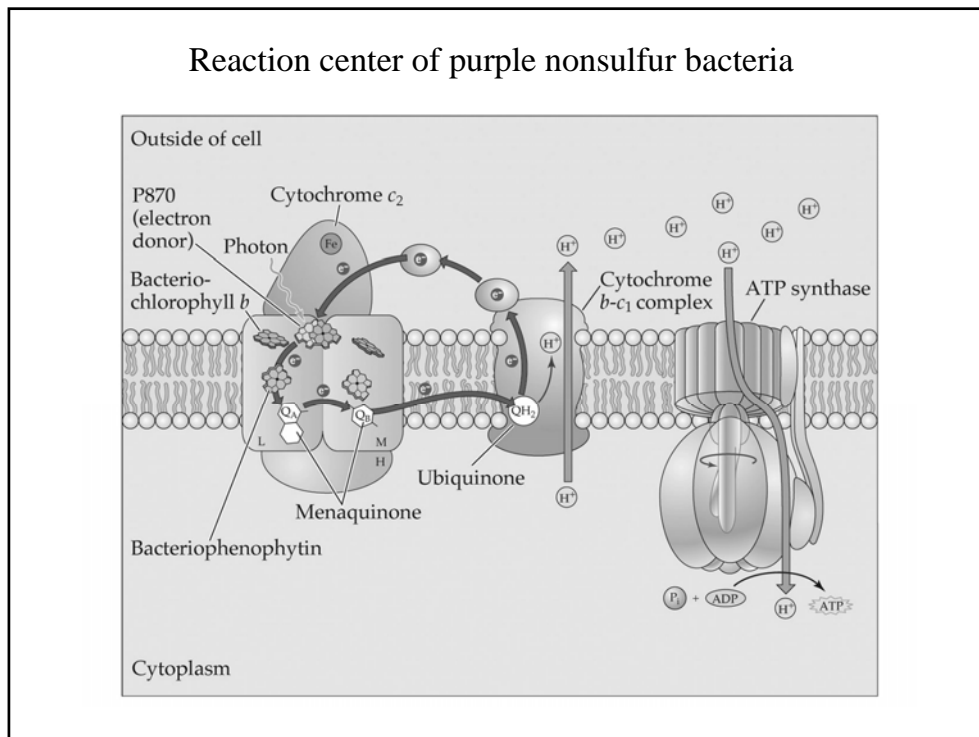
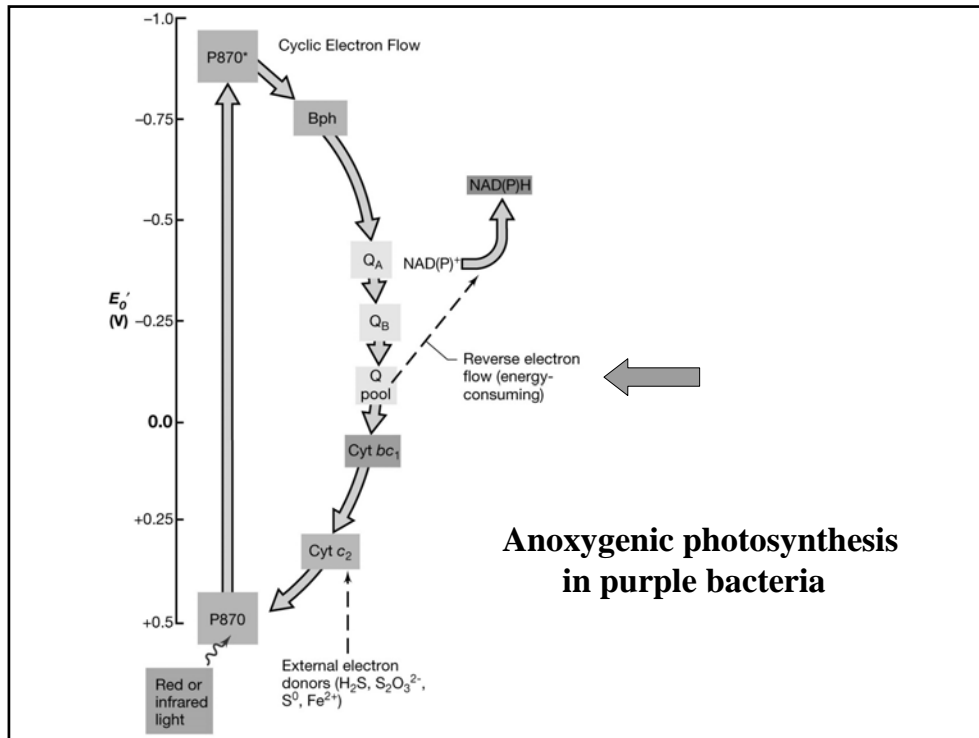
## Chromophores of phycobilisomes



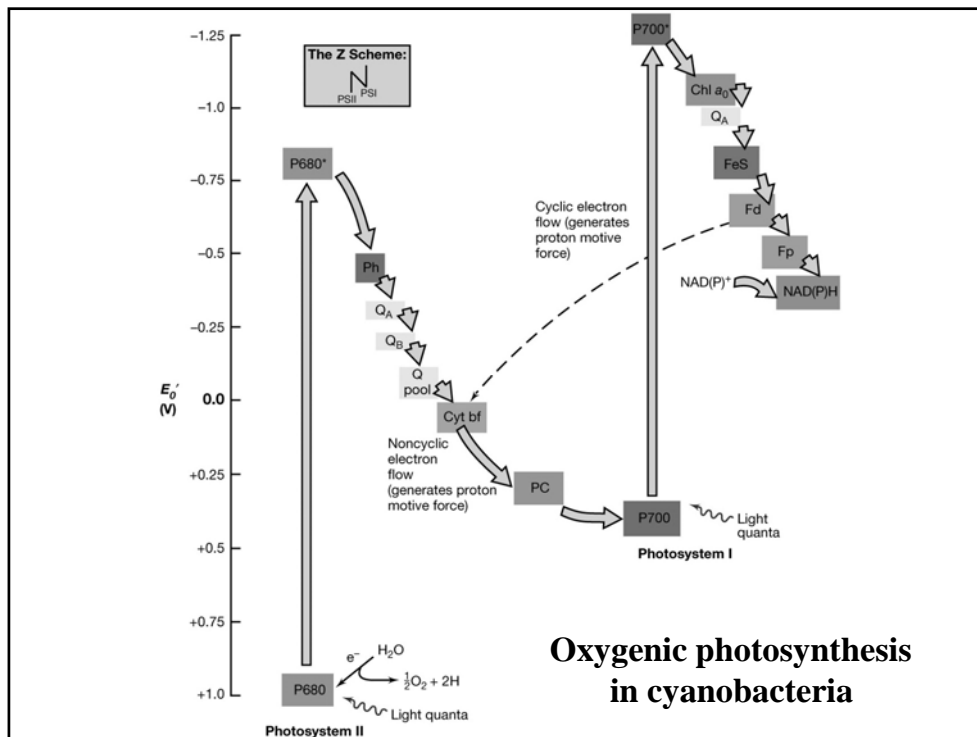
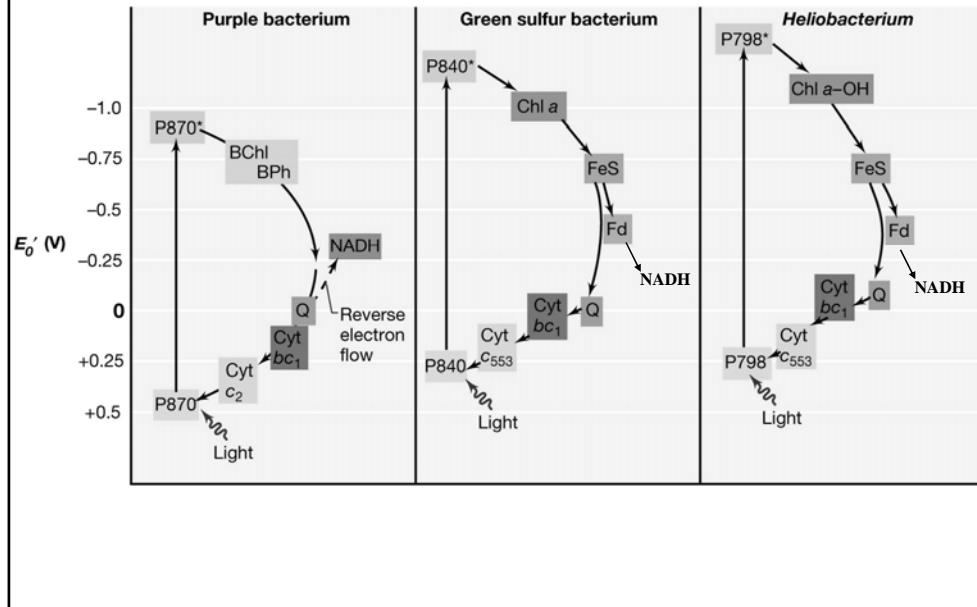
MICROBIAL LIFE, Figure 9.10 © 2002 Sinauer Associates, Inc.

## Absorption Spectra

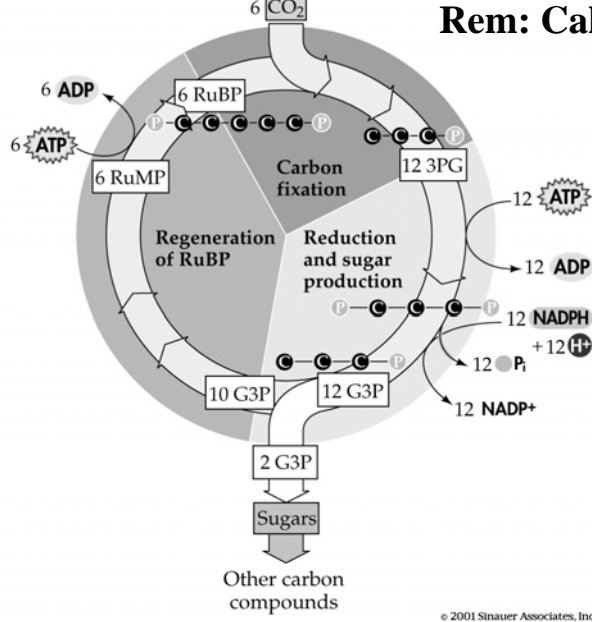




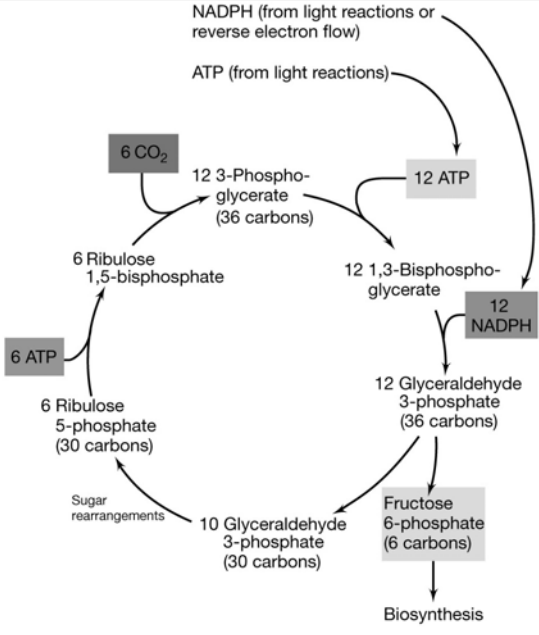
## Electron flow in phototrophs



# Rem: Calvin Cycle

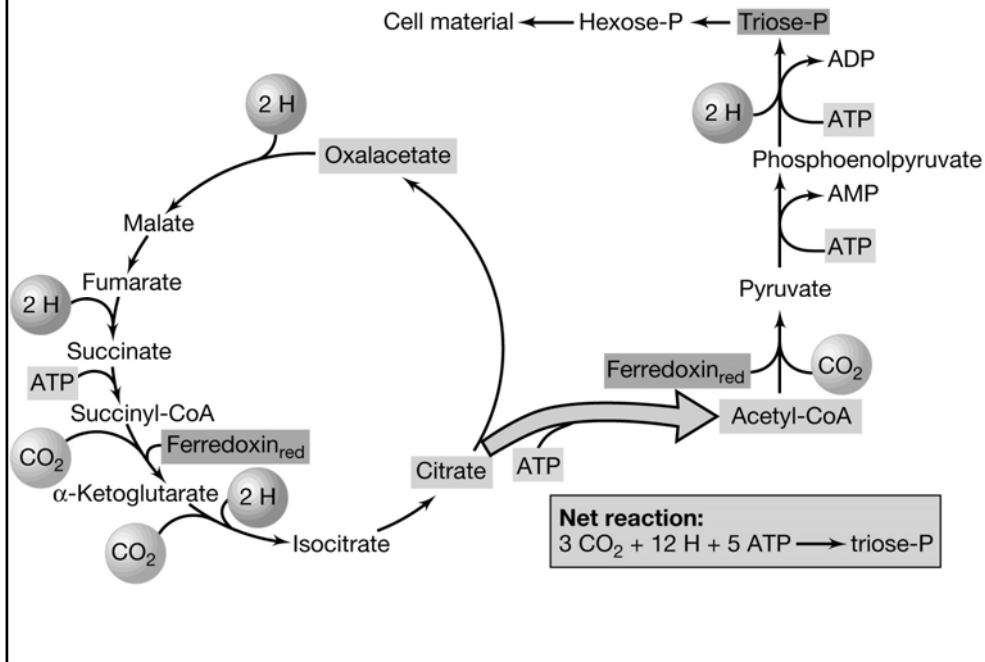


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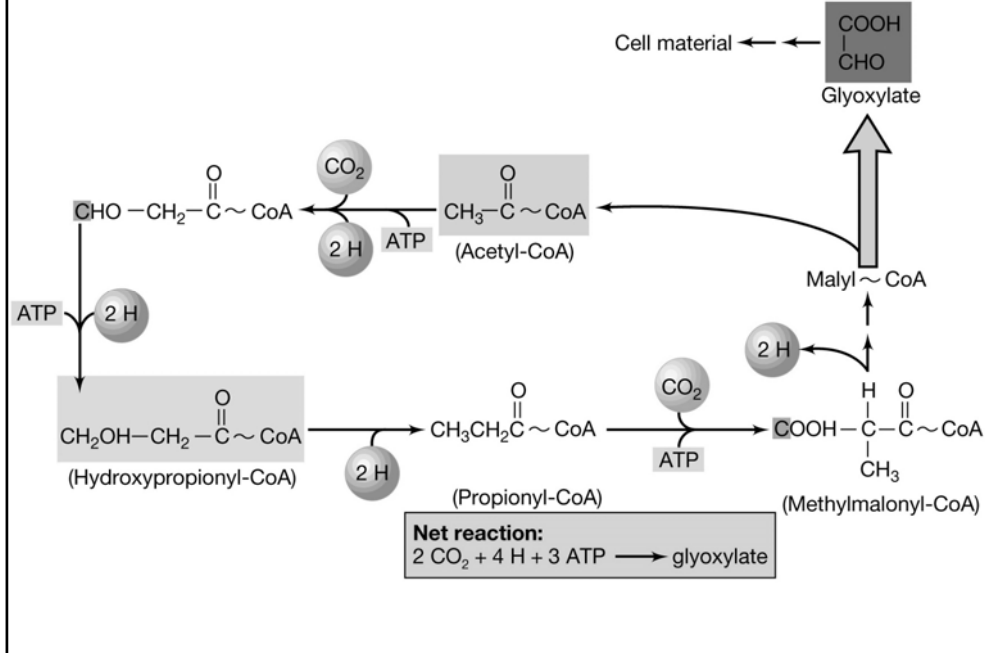


**Overall stoichiometry:**  
 $6 \text{ CO}_2 + 12 \text{ NADPH} + 18 \text{ ATP} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6(\text{PO}_3\text{H}_2) + 12 \text{ NADP}^+ + 18 \text{ ADP} + 17 \text{ P}_i$

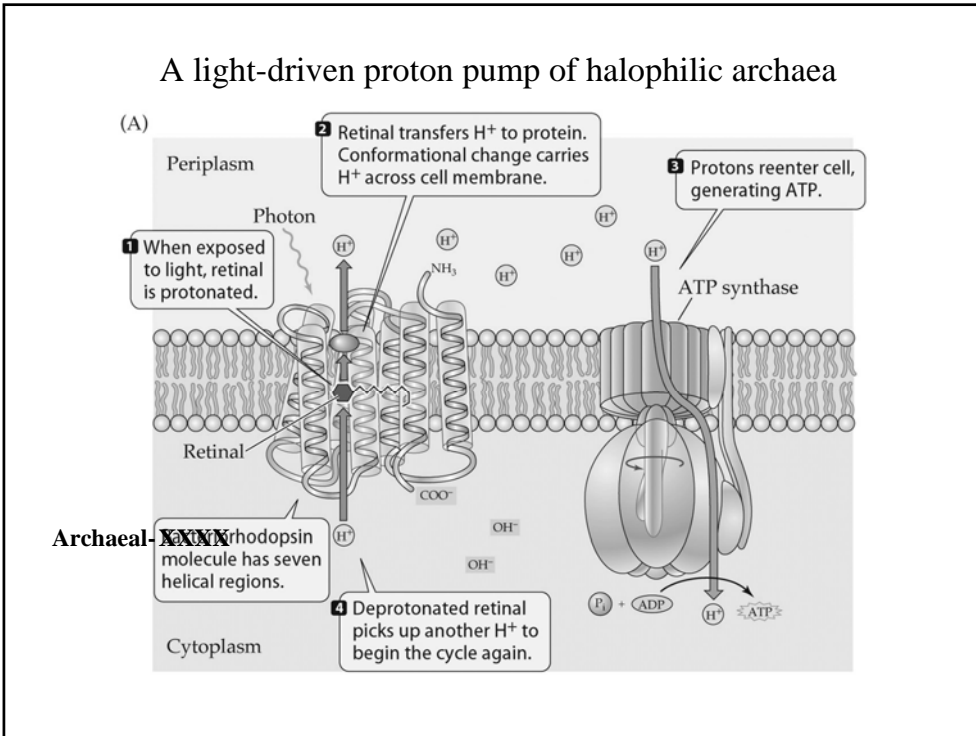
## Reverse TCA in GSBs



## Hydroxypropionate in GNBs

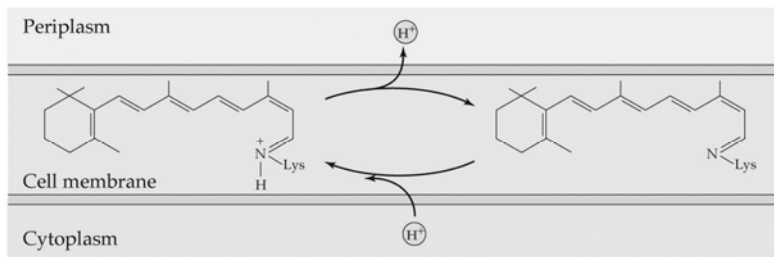






## Light-driven proton pump of halophilic archaea

(B)



Archaeal rhodopsin: retinal structure

## Banded Iron Formations ~2.5 Bya



Oxygenic or anoxygenic photosynthesis:  $\text{Fe}^{2+}_{\text{sol}}$  to  $\text{Fe}^{3+}_{\text{insol}}$