

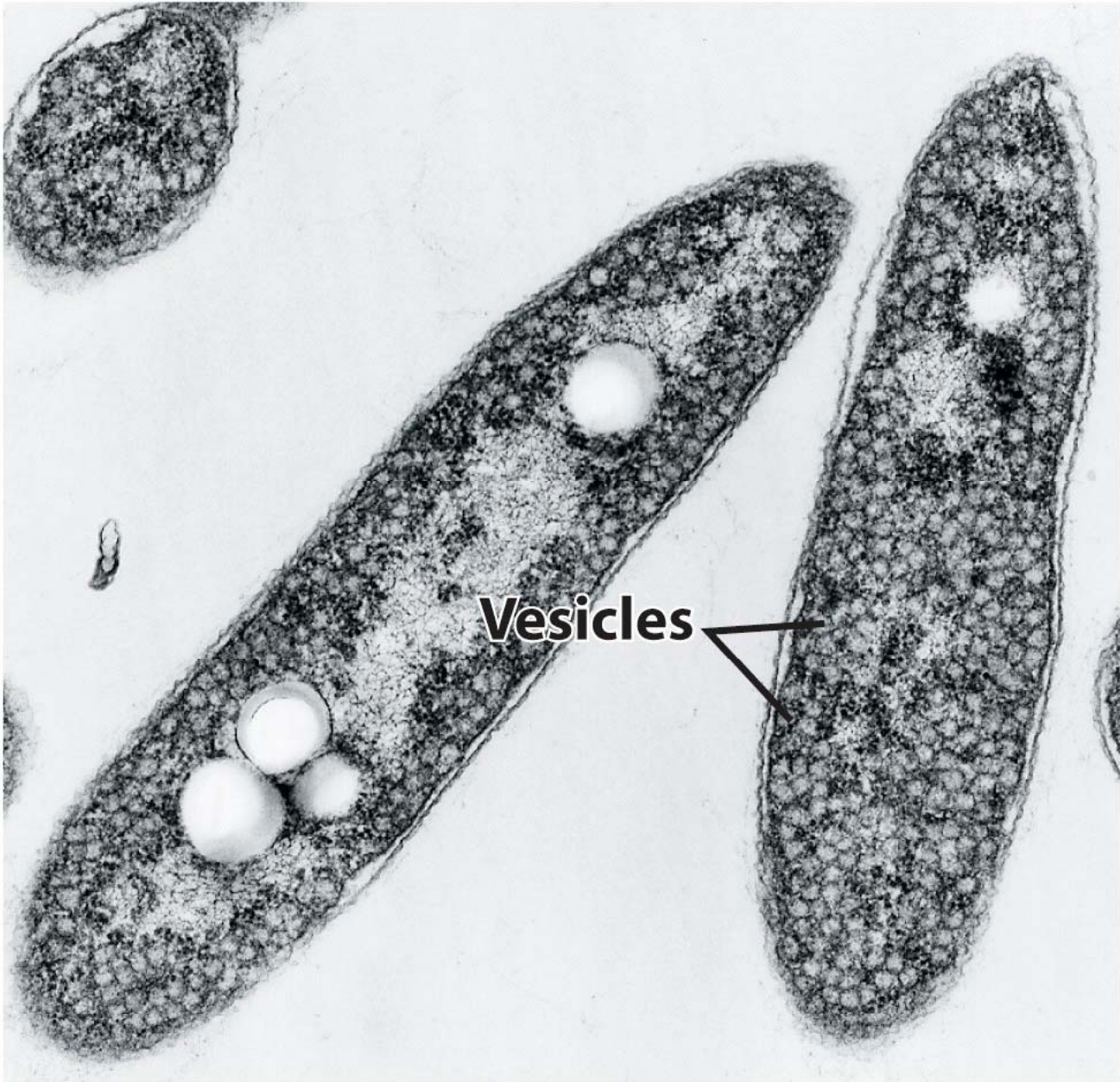
**More on
Phototrophic Potential**

PHOTOTROPHS
(use light
as energy source)

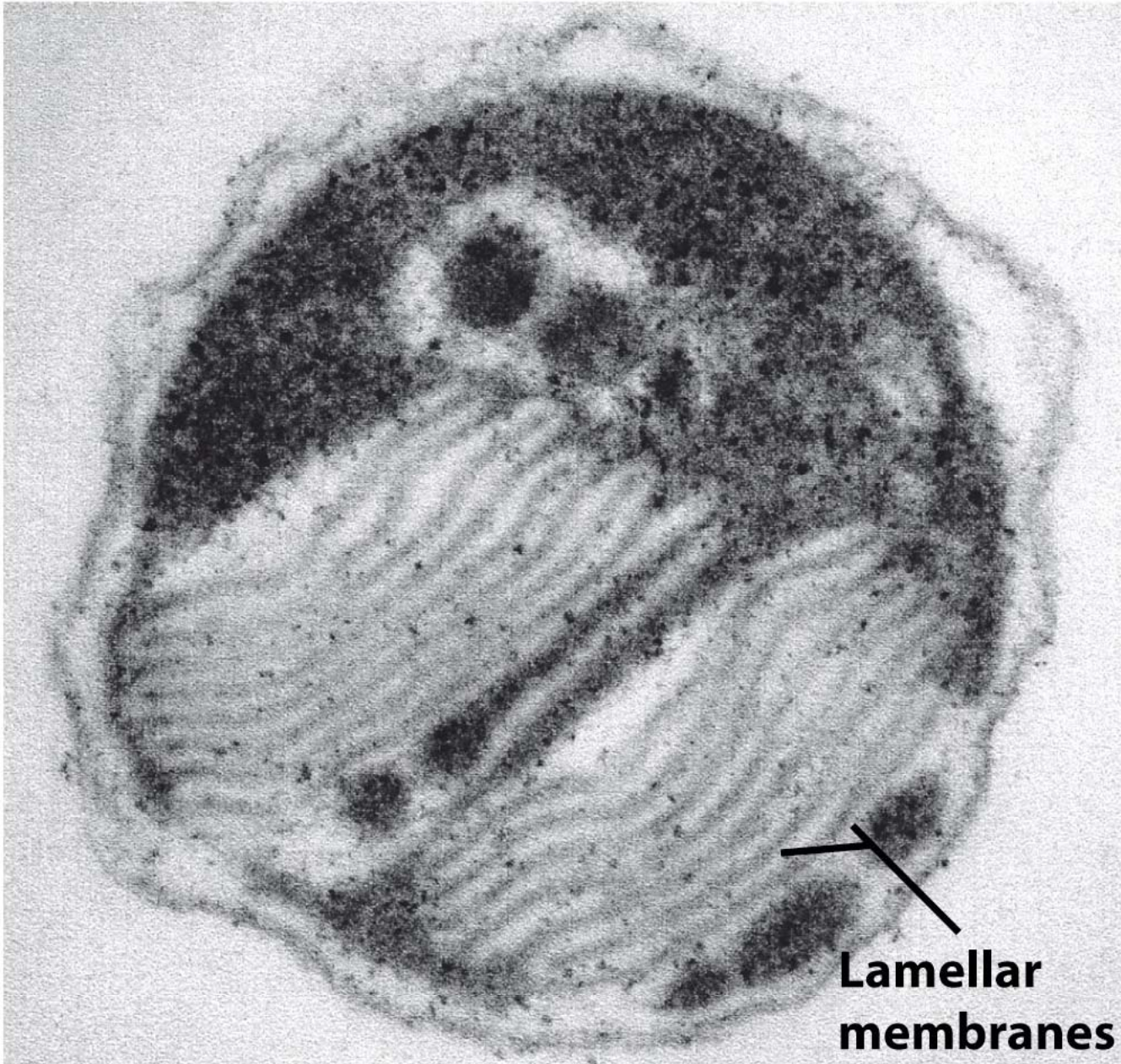
```
graph TD; A["PHOTOTROPHS  
(use light  
as energy source)"] --- B["Photoautotrophs  
(Use CO2)"]; A --- C["Photoheterotrophs  
(Use Organic Carbon)"]
```

Photoautotrophs
(Use CO₂)

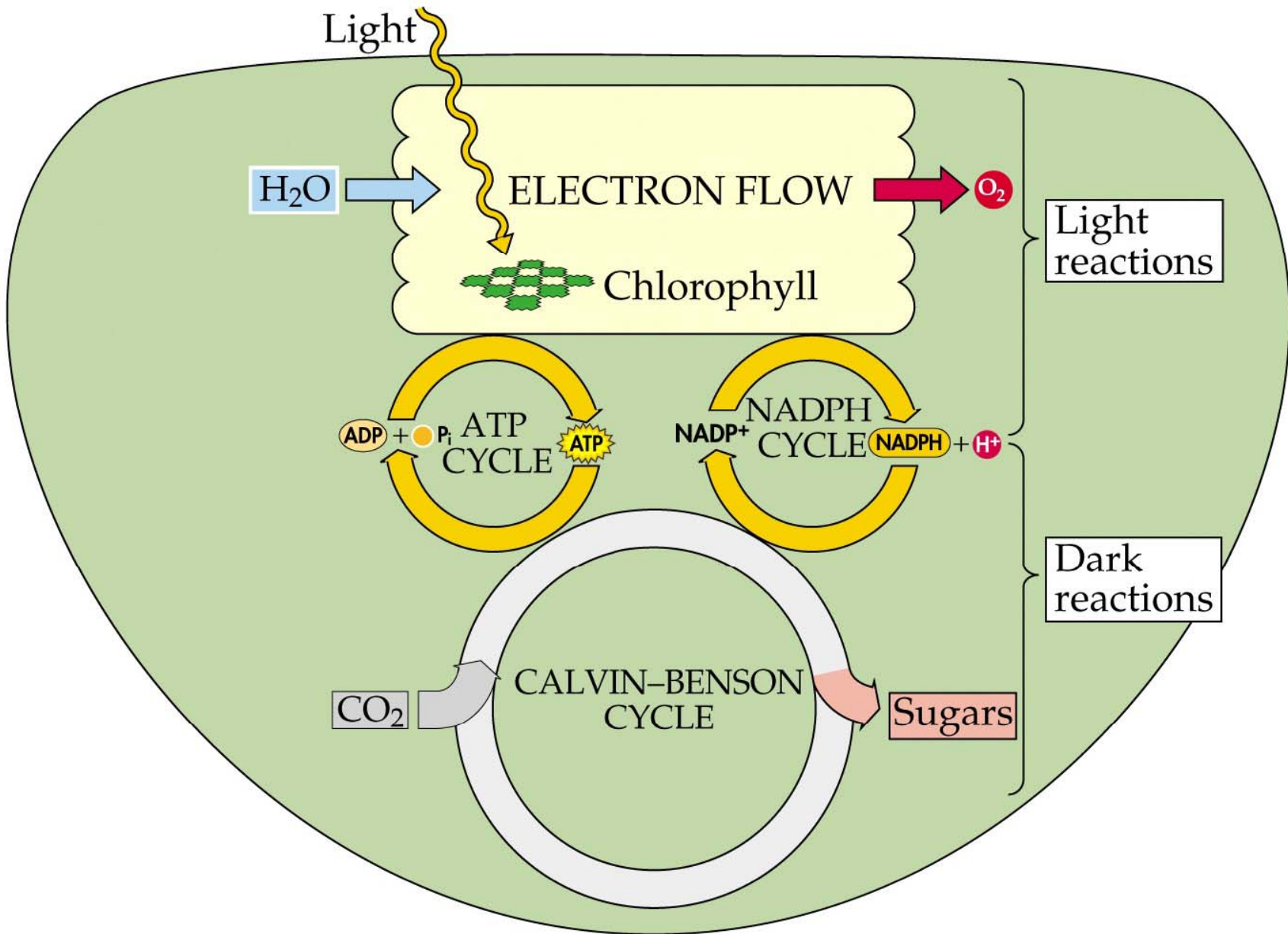
Photoheterotrophs
(Use Organic Carbon)



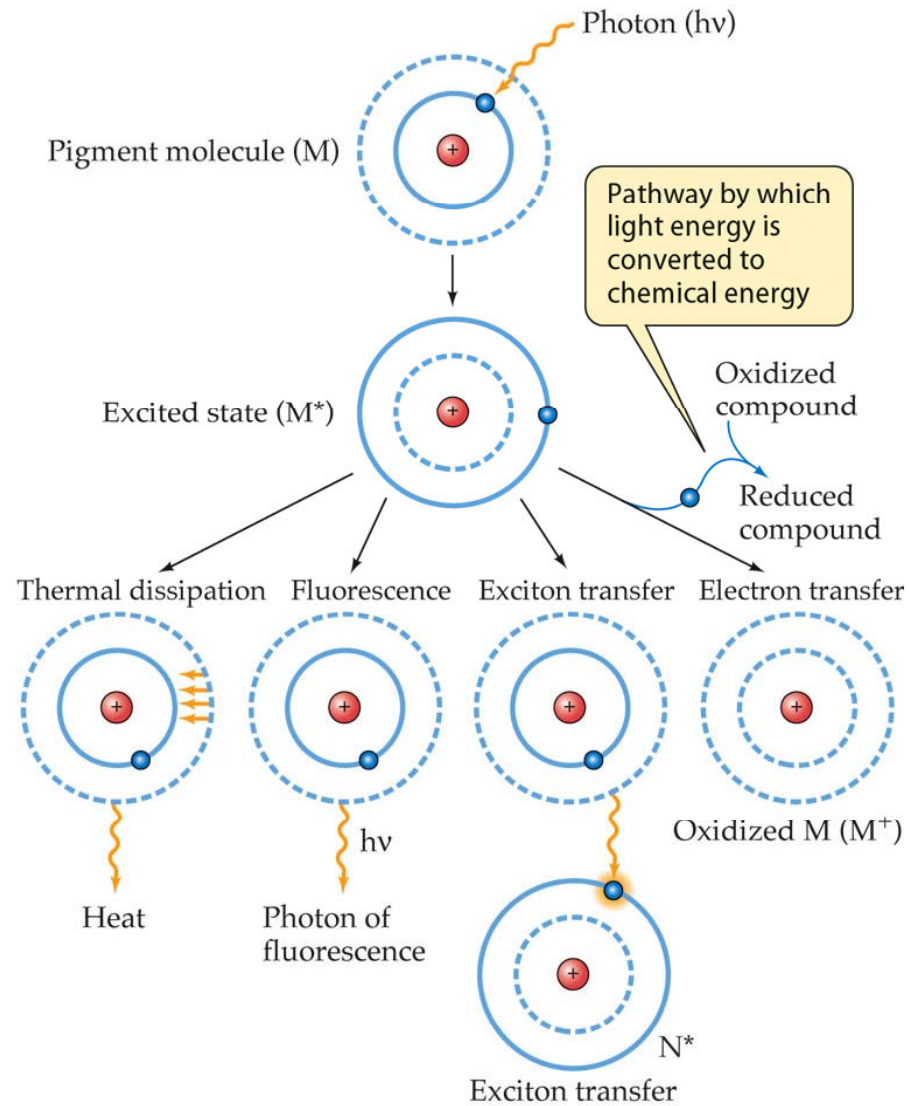
Vesicles



**Lamellar
membranes**



The possible fates of an excited electron

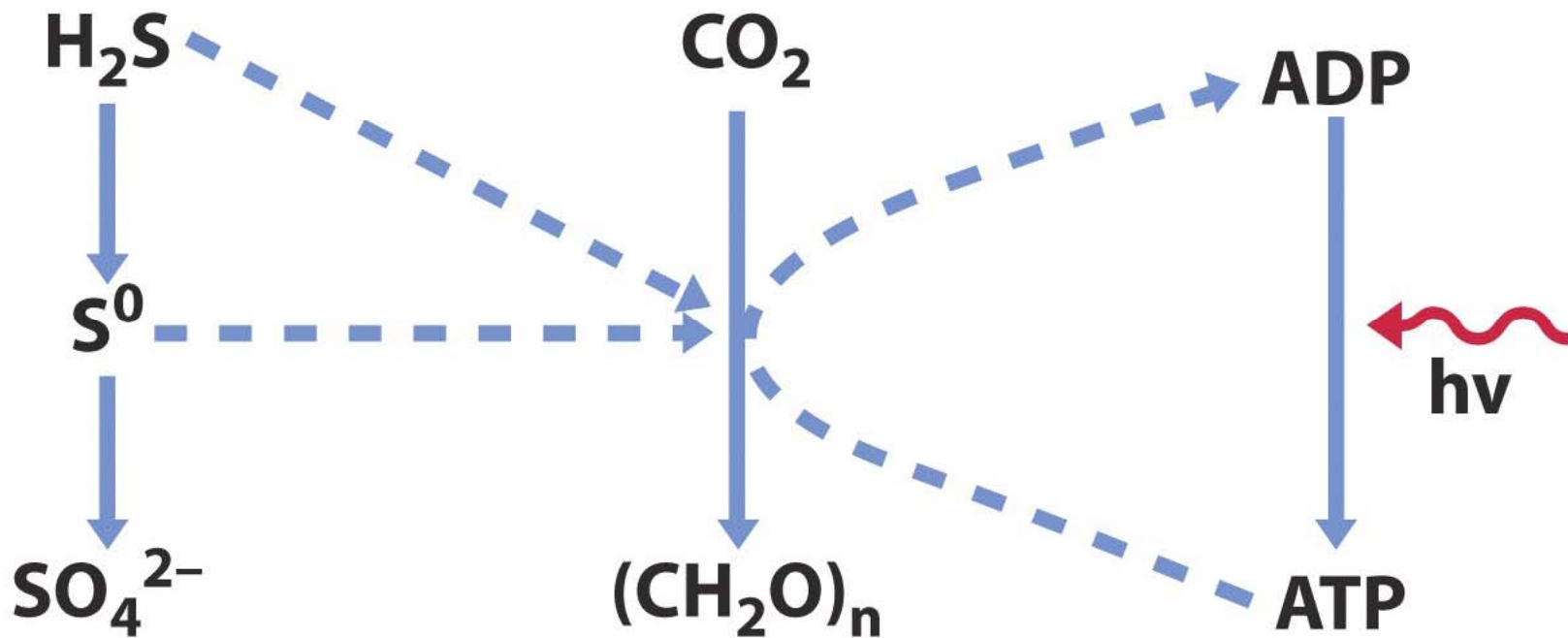


Anoxygenic

Reducing power

Carbon

Energy

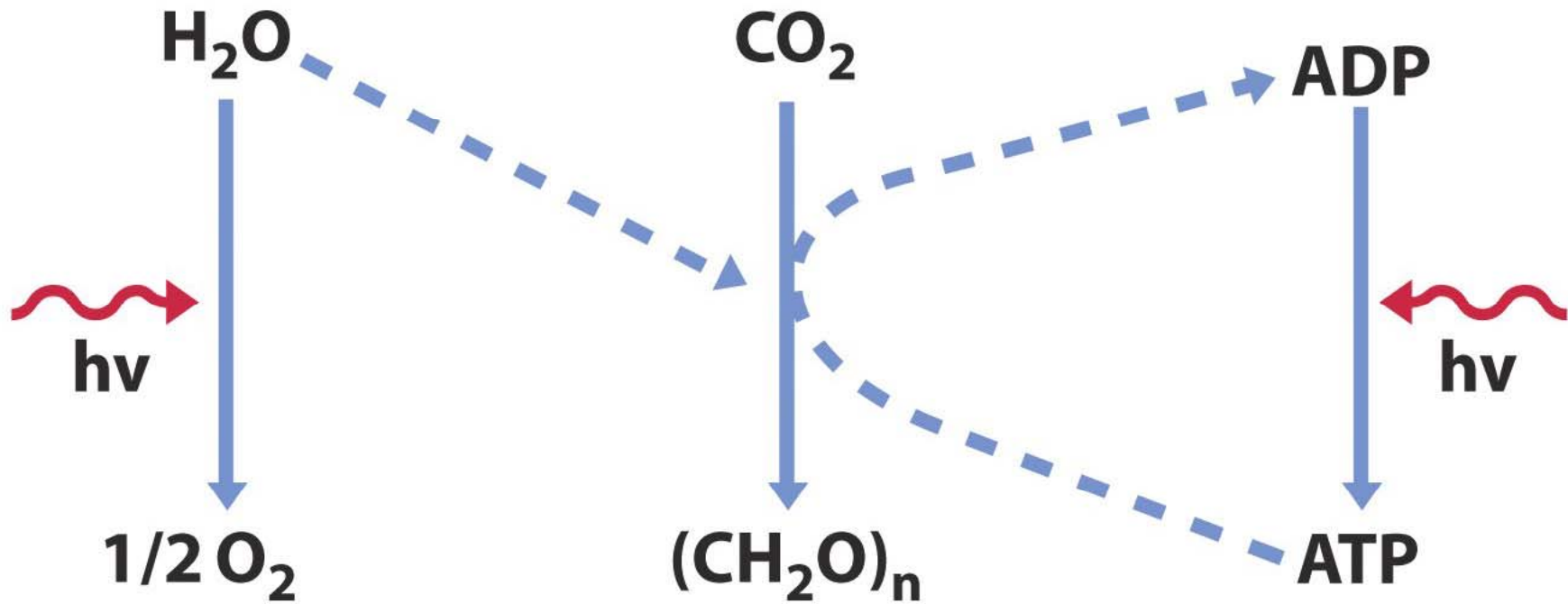


Oxygenic

Reducing power

Carbon

Energy

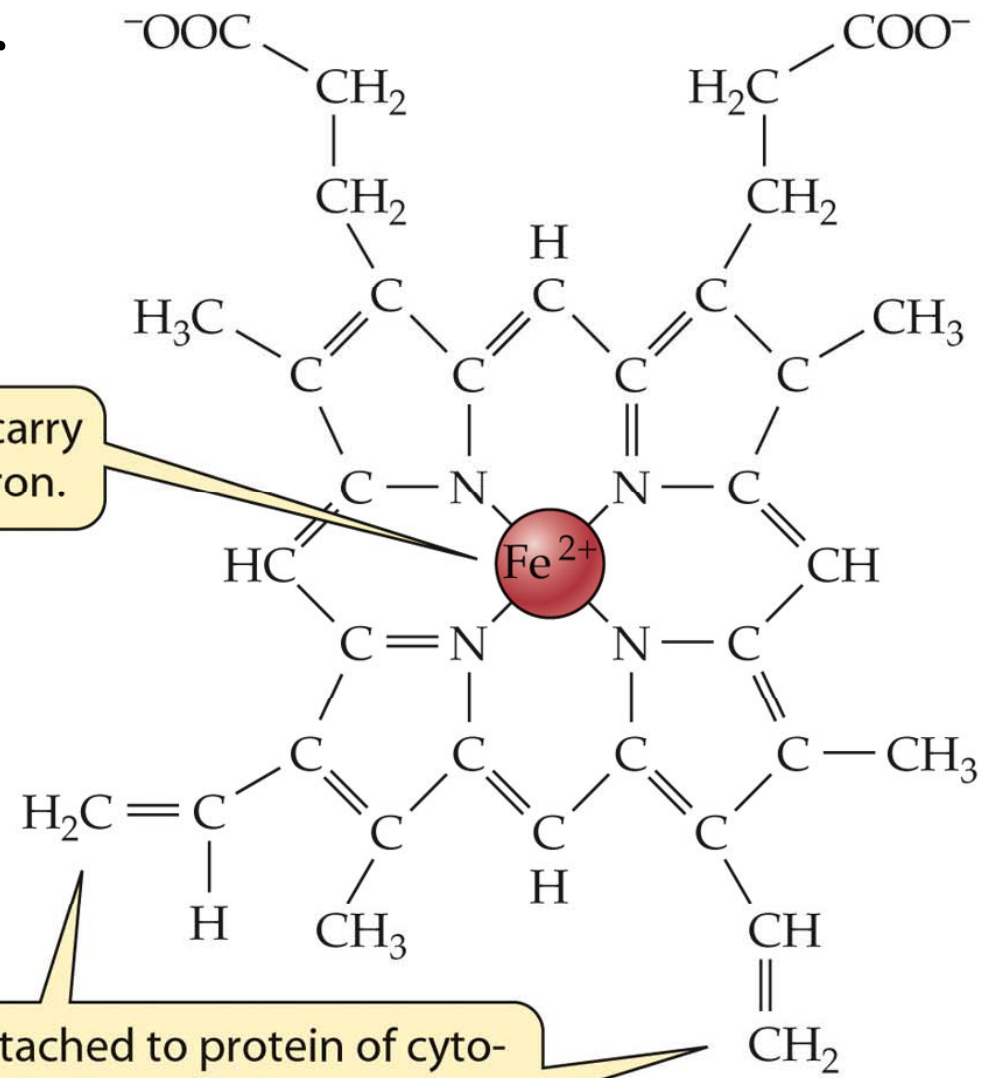


Wow!

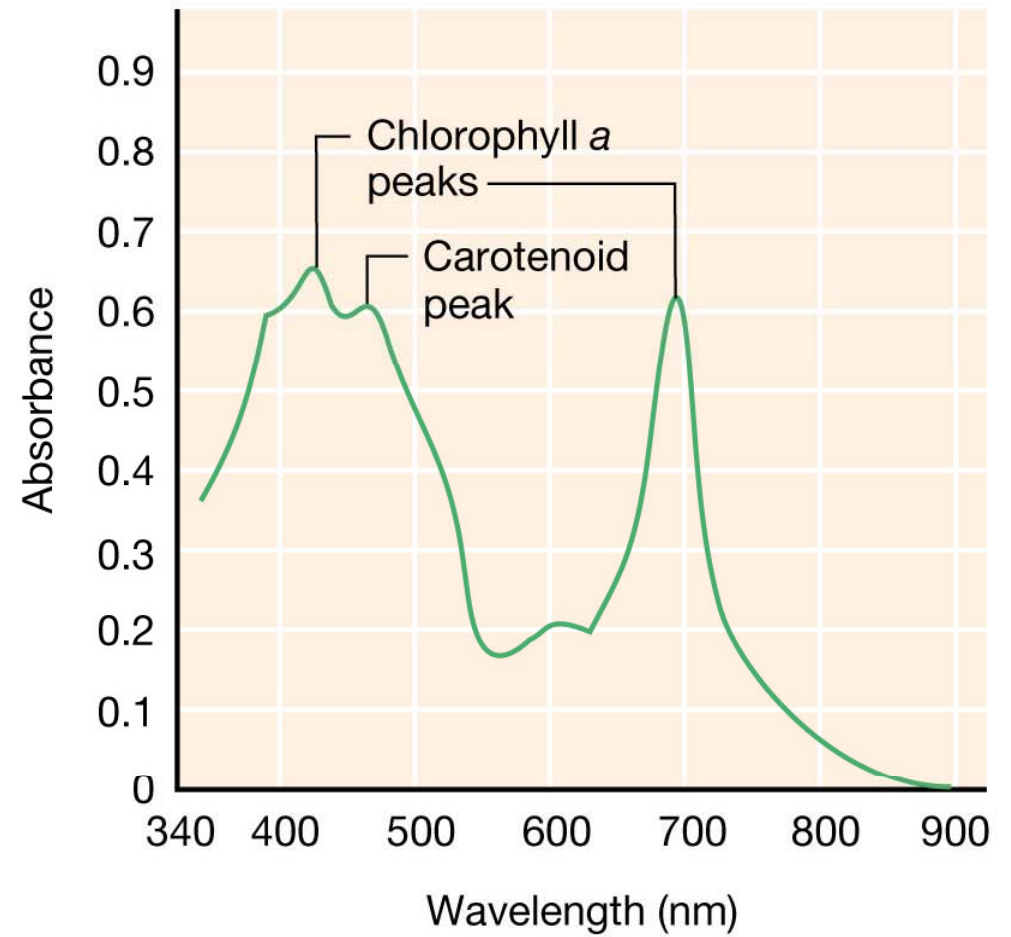
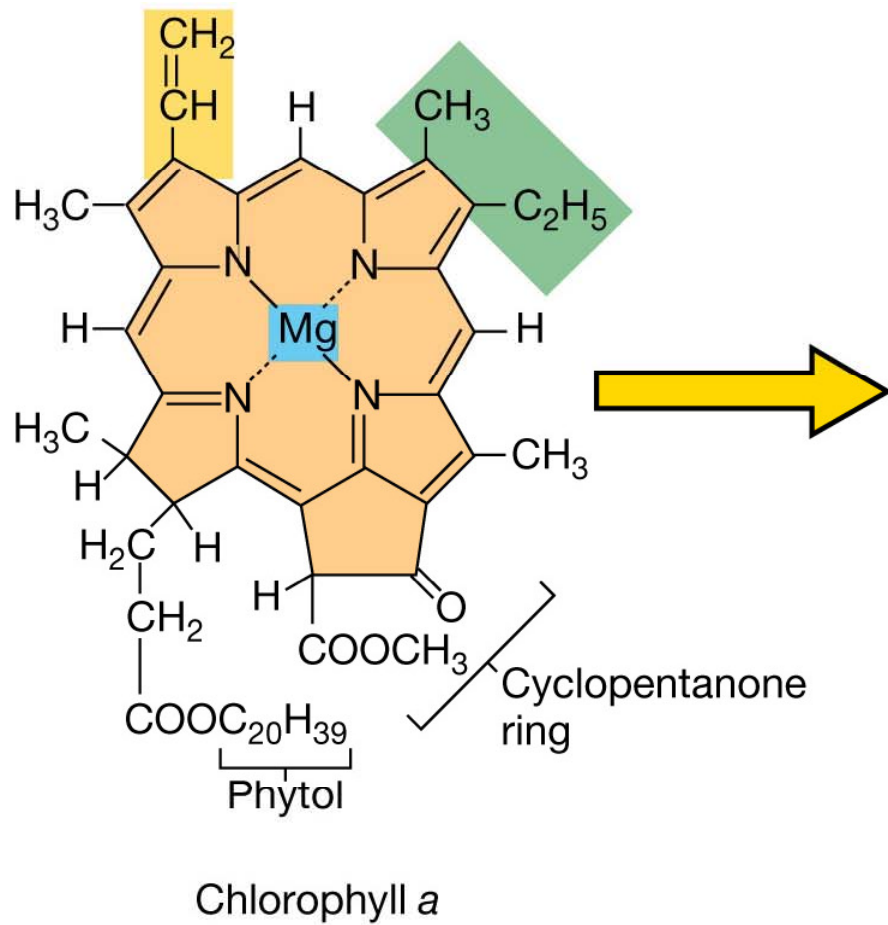
Porphyrin ring with an Fe center

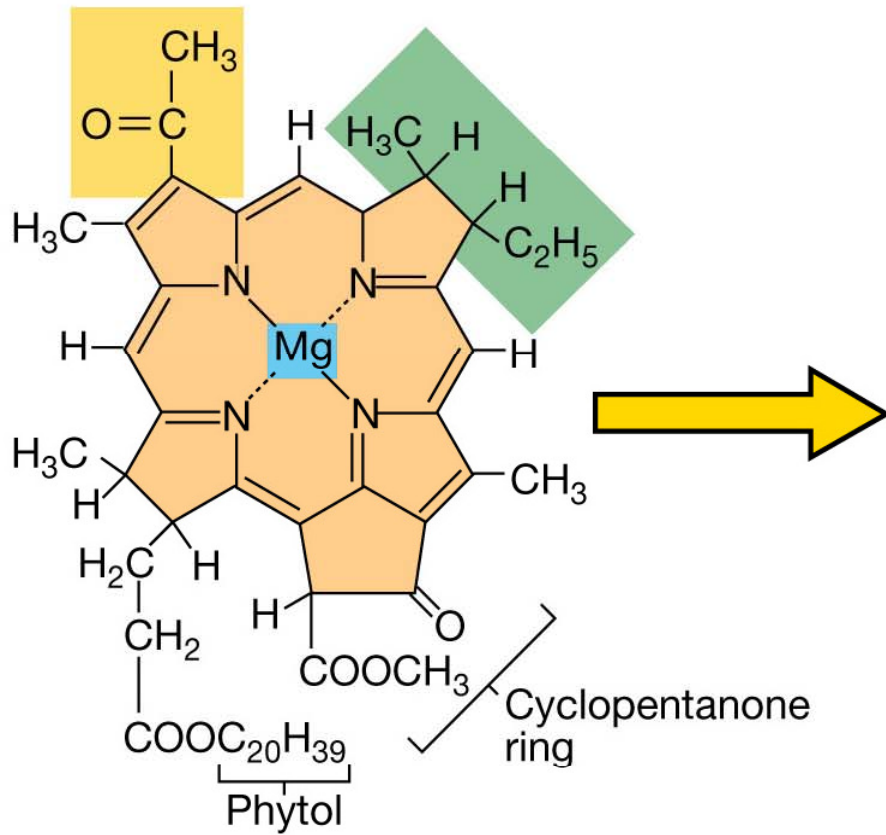
What about a
Co center?

The iron can carry
a single electron.

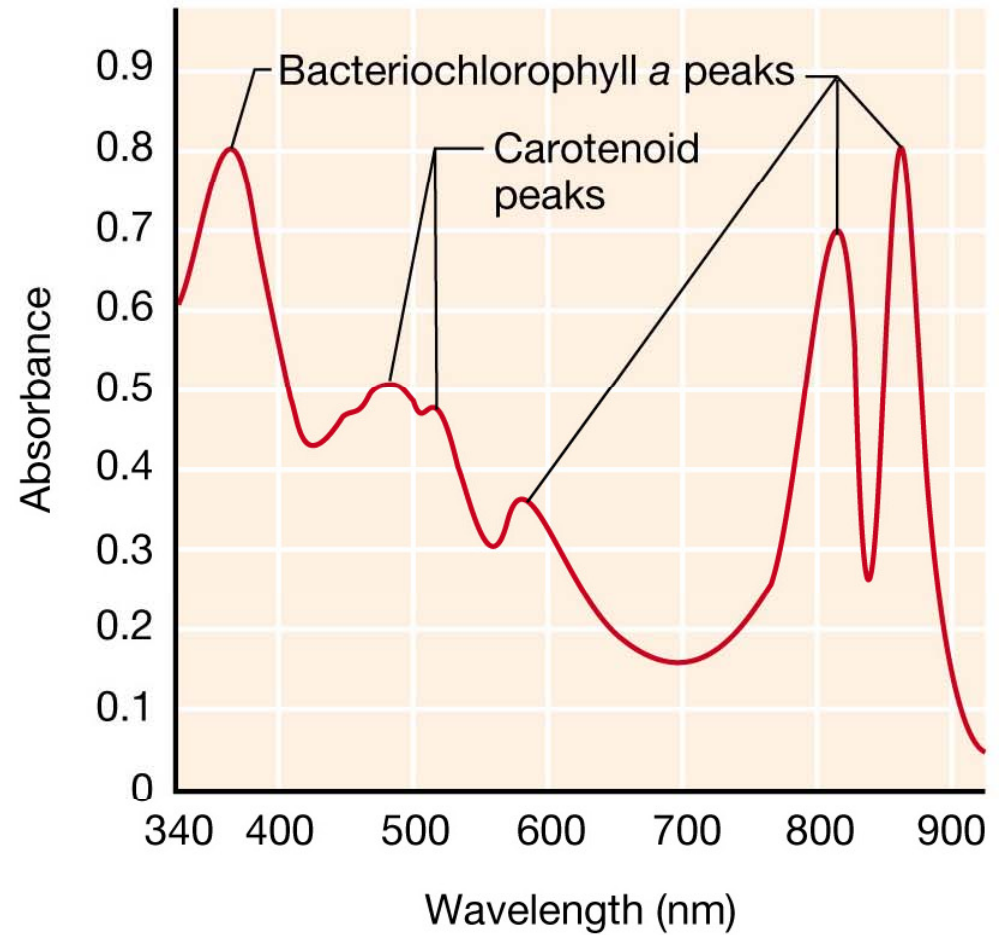


The heme is attached to protein of cytochrome molecule through these groups.



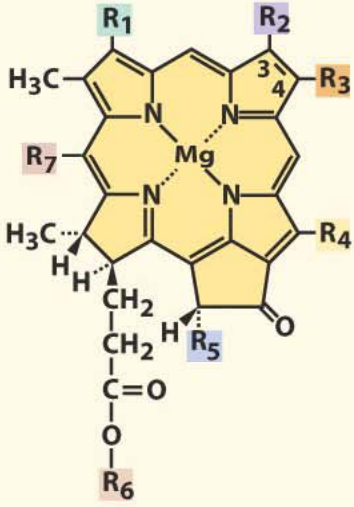


Bacteriochlorophyll a



Bacteriochlorophyll Structures

Pigment / Absorption maxima (<i>in vivo</i>)	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆	R ₇
Bchl a (purple bacteria)/ 805, 830–890	$\begin{array}{c} \text{—C—CH}_3 \\ \\ \text{O} \end{array}$	—CH_3^b	$\text{—CH}_2\text{—CH}_3$	—CH_3	$\begin{array}{c} \text{—C—O—CH}_3 \\ \\ \text{O} \end{array}$	P/Gg ^a —H	
Bchl b (purple bacteria)/ 835–850, 1020–1040	$\begin{array}{c} \text{—C—CH}_3 \\ \\ \text{O} \end{array}$	—CH_3^c	$\begin{array}{c} =\text{C—CH}_3 \\ \\ \text{H} \end{array}$	—CH_3	$\begin{array}{c} \text{—C—O—CH}_3 \\ \\ \text{O} \end{array}$	P	—H
Bchl c (green sulfur bacteria)/745–755	$\begin{array}{c} \text{H} \\ \\ \text{—C—CH}_3 \\ \\ \text{OH} \end{array}$	—CH_3	$\begin{array}{c} \text{—C}_2\text{H}_5 \\ \text{—C}_3\text{H}_7^d \\ \text{—C}_4\text{H}_9 \end{array}$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\ \text{—CH}_3 \end{array}$	—H	F	—CH ₃
Bchl c_s (green nonsulfur bacteria)/740	$\begin{array}{c} \text{H} \\ \\ \text{—C—CH}_3 \\ \\ \text{OH} \end{array}$	—CH_3	$\text{—C}_2\text{H}_5$	—CH_3	—H	S	—CH ₃
Bchl d (green sulfur bacteria)/705–740	$\begin{array}{c} \text{H} \\ \\ \text{—C—CH}_3 \\ \\ \text{OH} \end{array}$	—CH_3	$\begin{array}{c} \text{—C}_2\text{H}_5 \\ \text{—C}_3\text{H}_7 \\ \text{—C}_4\text{H}_9 \end{array}$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\ \text{—CH}_3 \end{array}$	—H	F	—H
Bchl e (green sulfur bacteria)/719–726	$\begin{array}{c} \text{H} \\ \\ \text{—C—CH}_3 \\ \\ \text{OH} \end{array}$	$\begin{array}{c} \text{—C—H} \\ \\ \text{O} \end{array}$	$\begin{array}{c} \text{—C}_2\text{H}_5 \\ \text{—C}_3\text{H}_7 \\ \text{—C}_4\text{H}_9 \end{array}$	$\text{—C}_2\text{H}_5$	—H	F	—CH ₃
Bchl g (heliobacteria)/ 670, 788	$\begin{array}{c} \text{H} \\ \\ \text{—C=CH}_2 \end{array}$	—CH_3^b	$\text{—C}_2\text{H}_5$	—CH_3	$\begin{array}{c} \text{—C—O—CH}_3 \\ \\ \text{O} \end{array}$	F	—H



^aP, Phytol ester (C₂₀H₃₉O—); F, farnesyl ester (C₁₅H₂₅O—); Gg, geranylgeraniol ester (C₁₀H₁₇O—); S, stearyl alcohol (C₁₈H₃₇O—).

^bNo double bond between C₃ and C₄; additional H atoms are in positions C₃ and C₄.

^cNo double bond between C₃ and C₄; an additional H atom is in position C₃.

^dBacteriochlorophylls c, d, and e consist of isomeric mixtures with the different substituents on R₃ as shown.

Chloroplast Structure

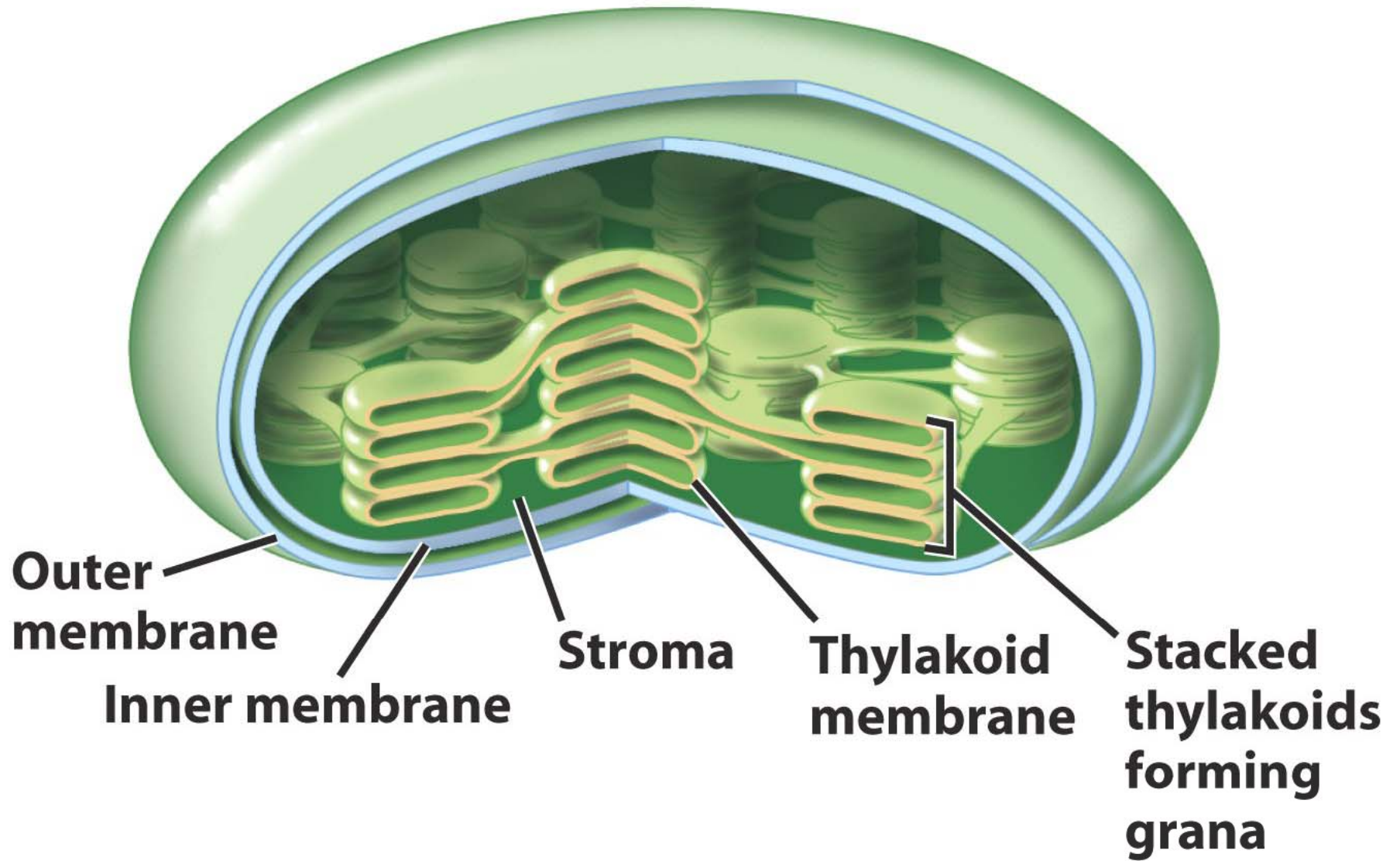
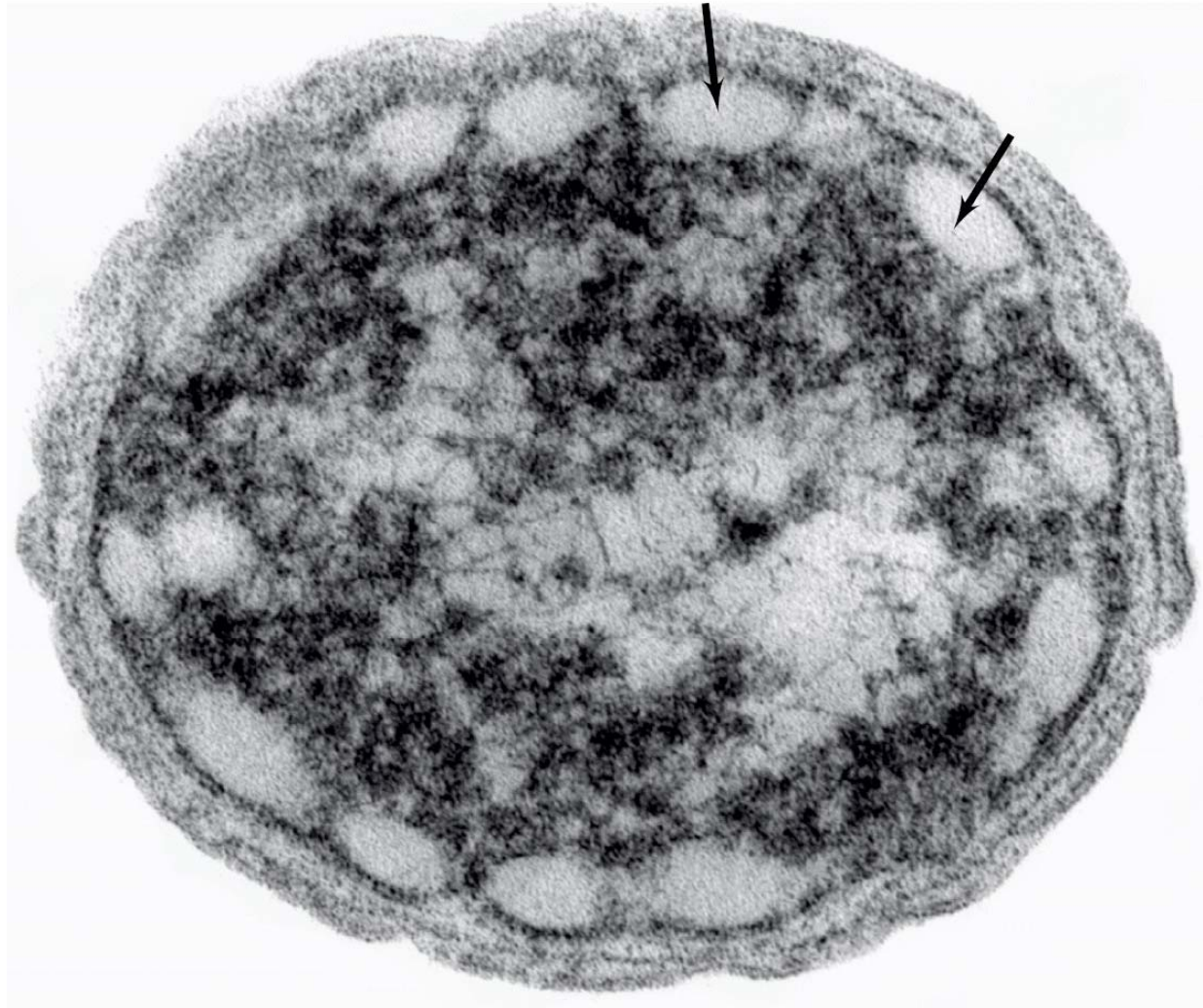


Table 9.1**Some general properties of the various photosynthetic bacteria**

	Nonsulfur Purple Bacteria	Purple Sulfur Bacteria	Green Sulfur Bacteria	Cyano- bacteria	Helio- bacteria
Source of reducing power (e^-)	H_2 , reduced organic	H_2S	H_2S	H_2O	Lactate, organic
Oxidized product	Oxidized organic	SO_4^{2-}	SO_4^{2-}	O_2	Oxidized organic
Source of carbon	CO_2 or organic	CO_2	CO_2	CO_2	Lactate pyruvate
Heterotrophic growth	Common	Limited ^a	Limited ^a	Limited ^a	Required

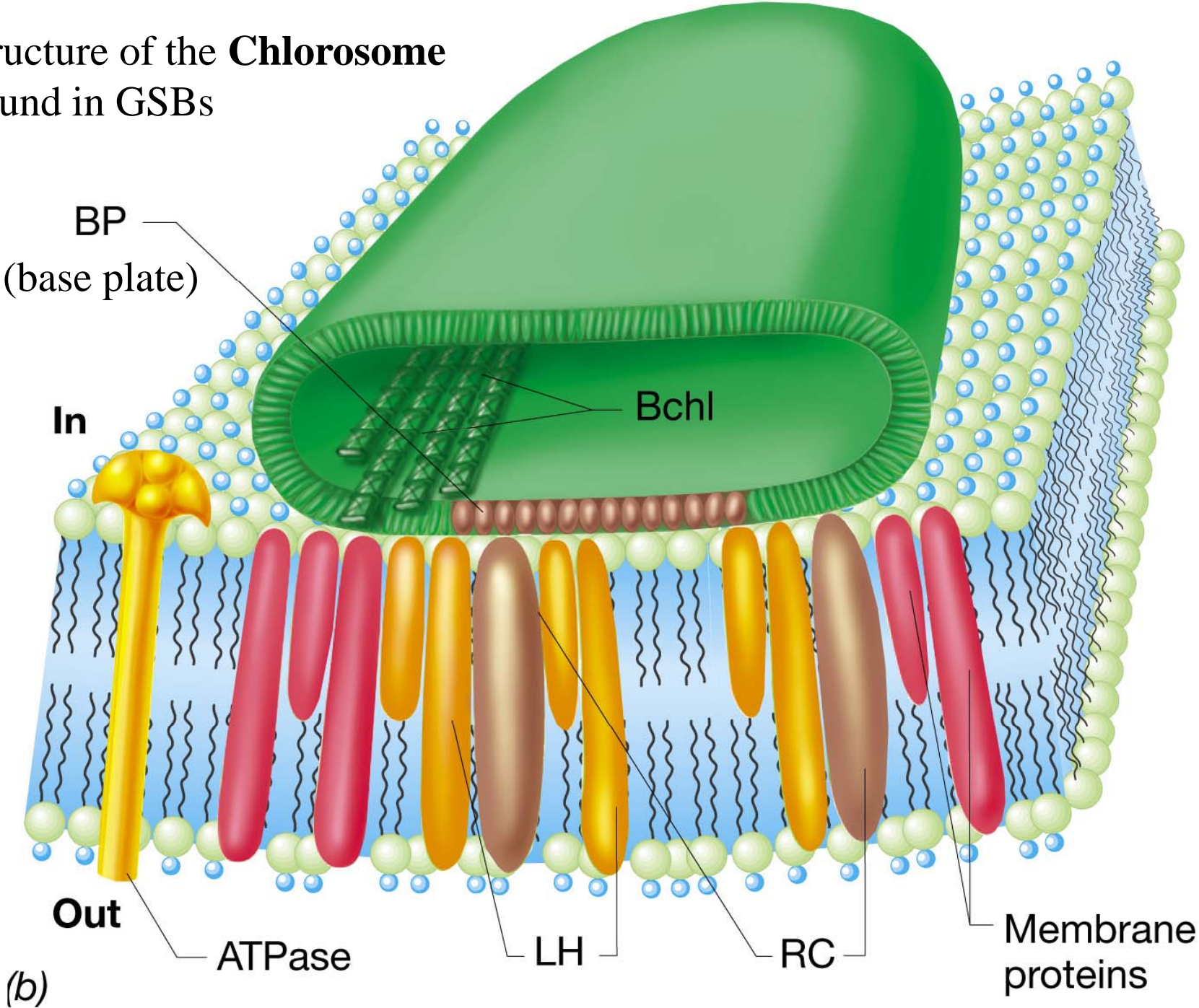
^aGenerally limited to assimilation of low molecular weight organics during autotrophic growth.

Structure and Location of the **Chlorosome**

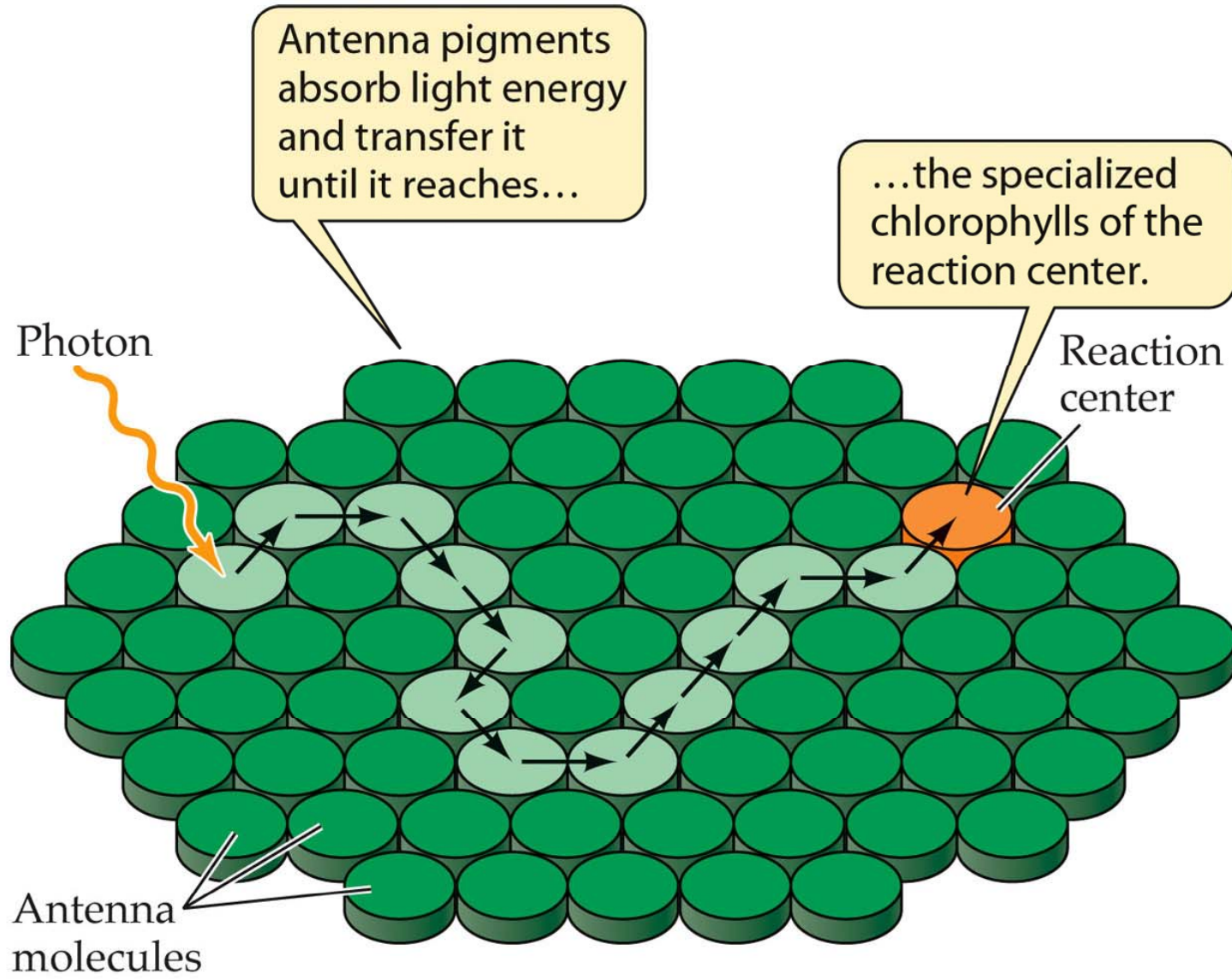


Found in *GSBs*

Structure of the **Chlorosome**
Found in GSBs



Photosynthetic unit



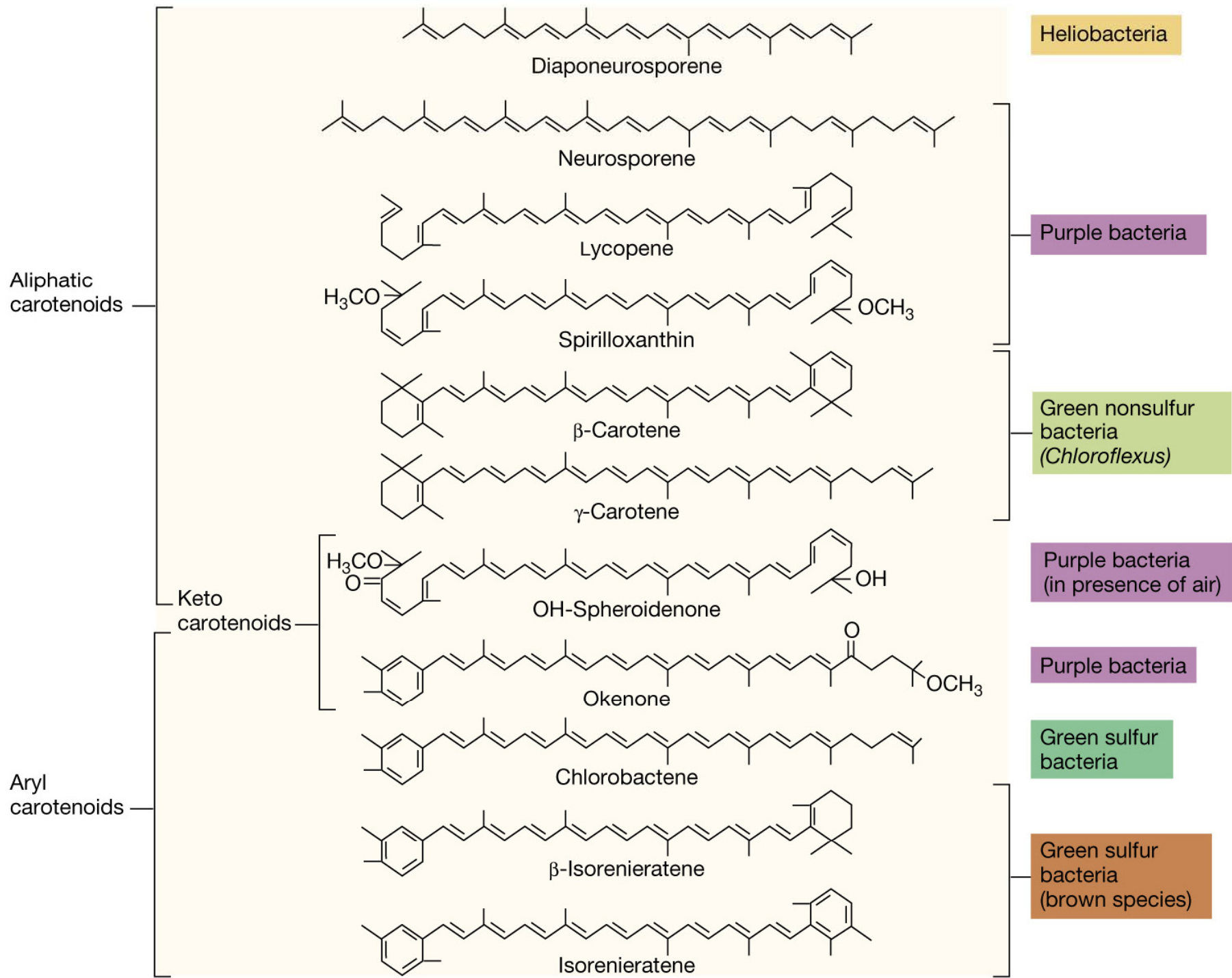
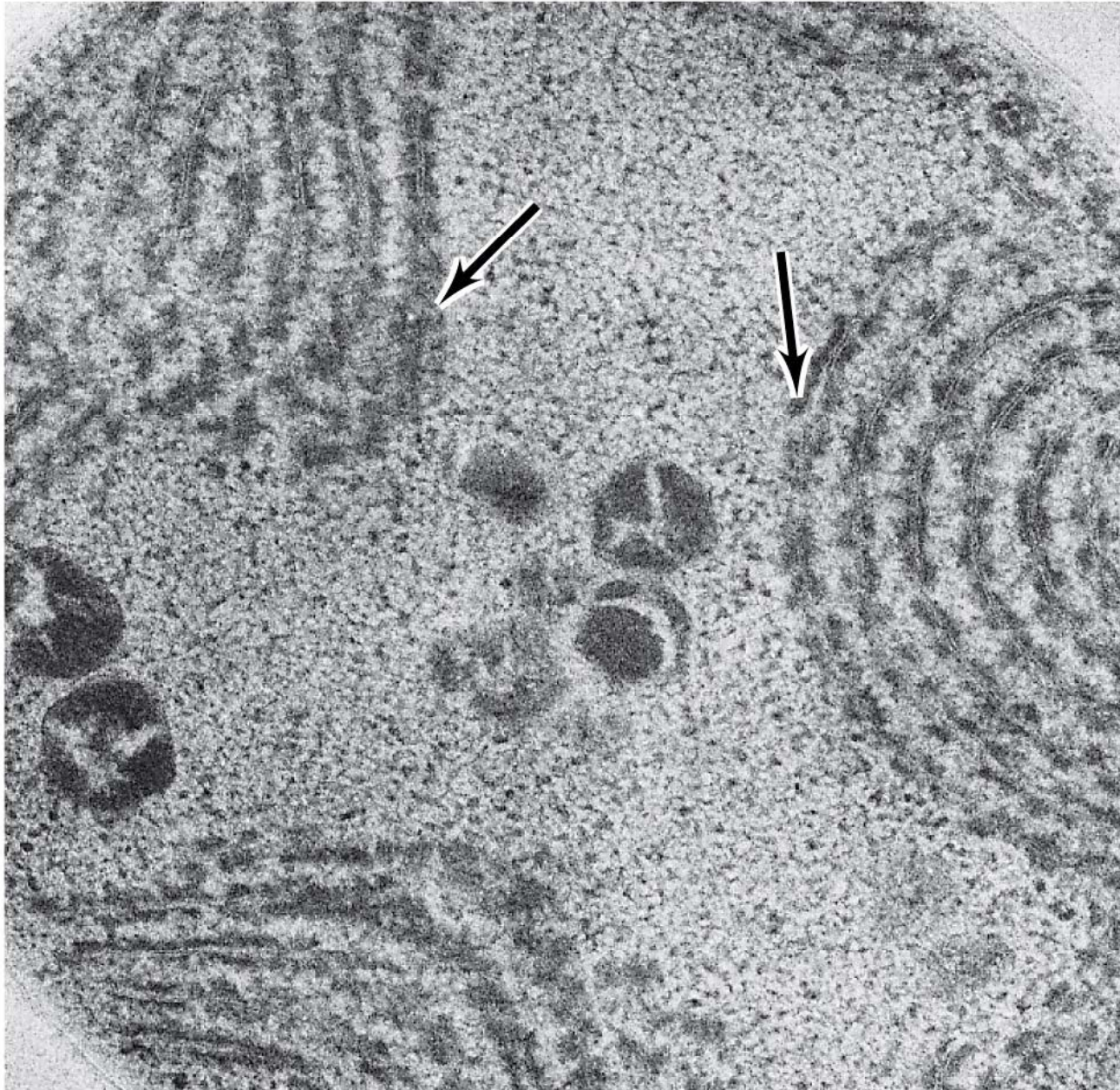


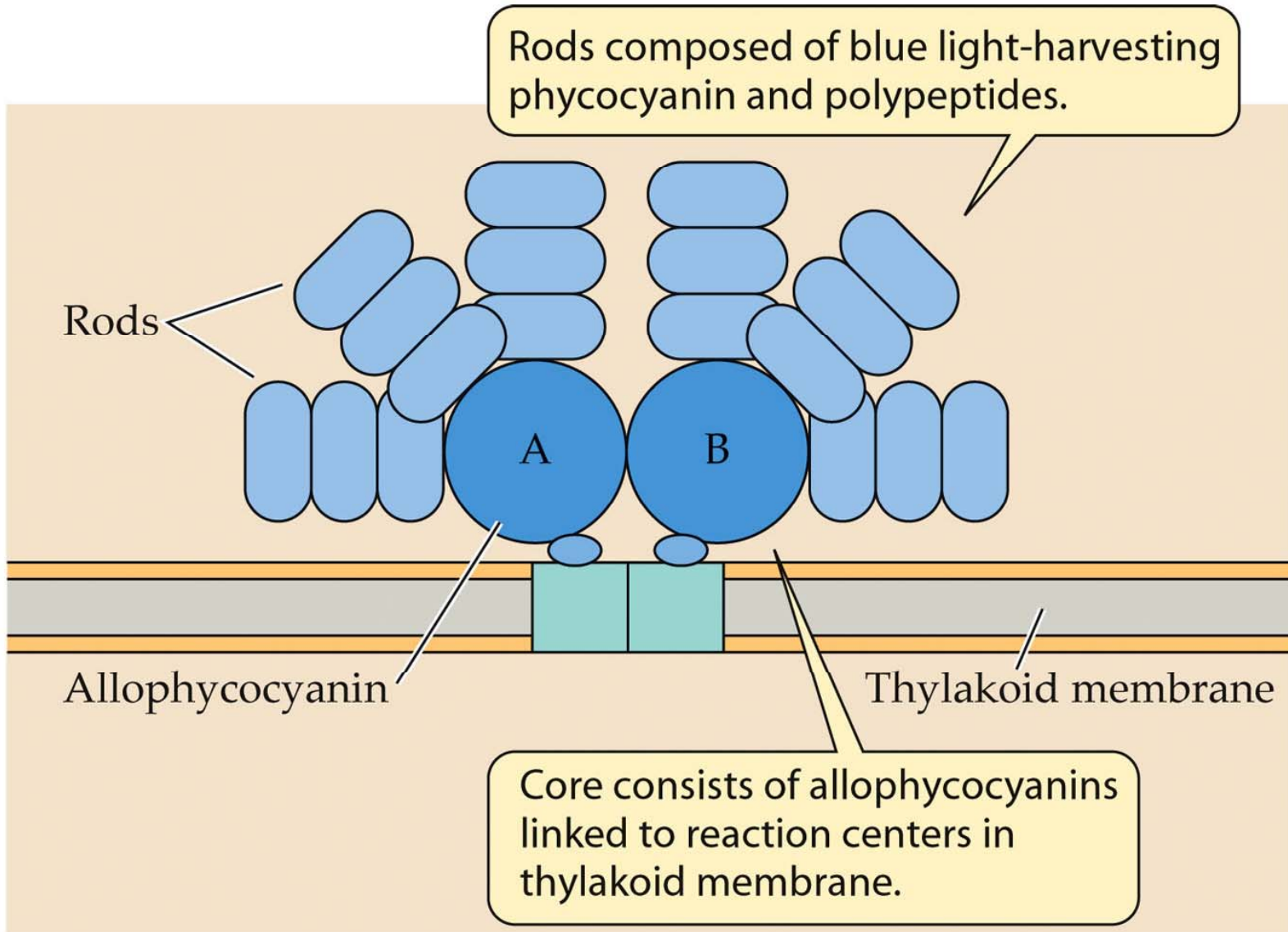
Table 9.2**The bacteriochlorophyll present in photosynthetic bacteria and primary acceptors involved in energy conserving reactions**

	Electron Donor	Electron Acceptor
Purple nonsulfur bacteria	Bacteriochlorophyll <i>a</i> and <i>b</i>	Bacteriopheophytin <i>a</i> , Q _A , and Q _B
Green sulfur bacteria	Bacteriochlorophyll <i>c</i> , <i>d</i> , and <i>e</i>	Bacteriopheophytin <i>a</i> and FeS-protein
Cyanobacteria photosystem I	Chlorophyll <i>a</i>	Chlorophyll <i>a</i> and FeS-protein
Cyanobacteria photosystem II	Chlorophyll <i>a</i>	Pheophytin <i>a</i> , Q _A , Q _B , and plastoquinones
<i>Heliobacteria</i>	Bacteriochlorophyll <i>g</i>	Bacteriochlorophyll <i>c</i> and FeS-protein

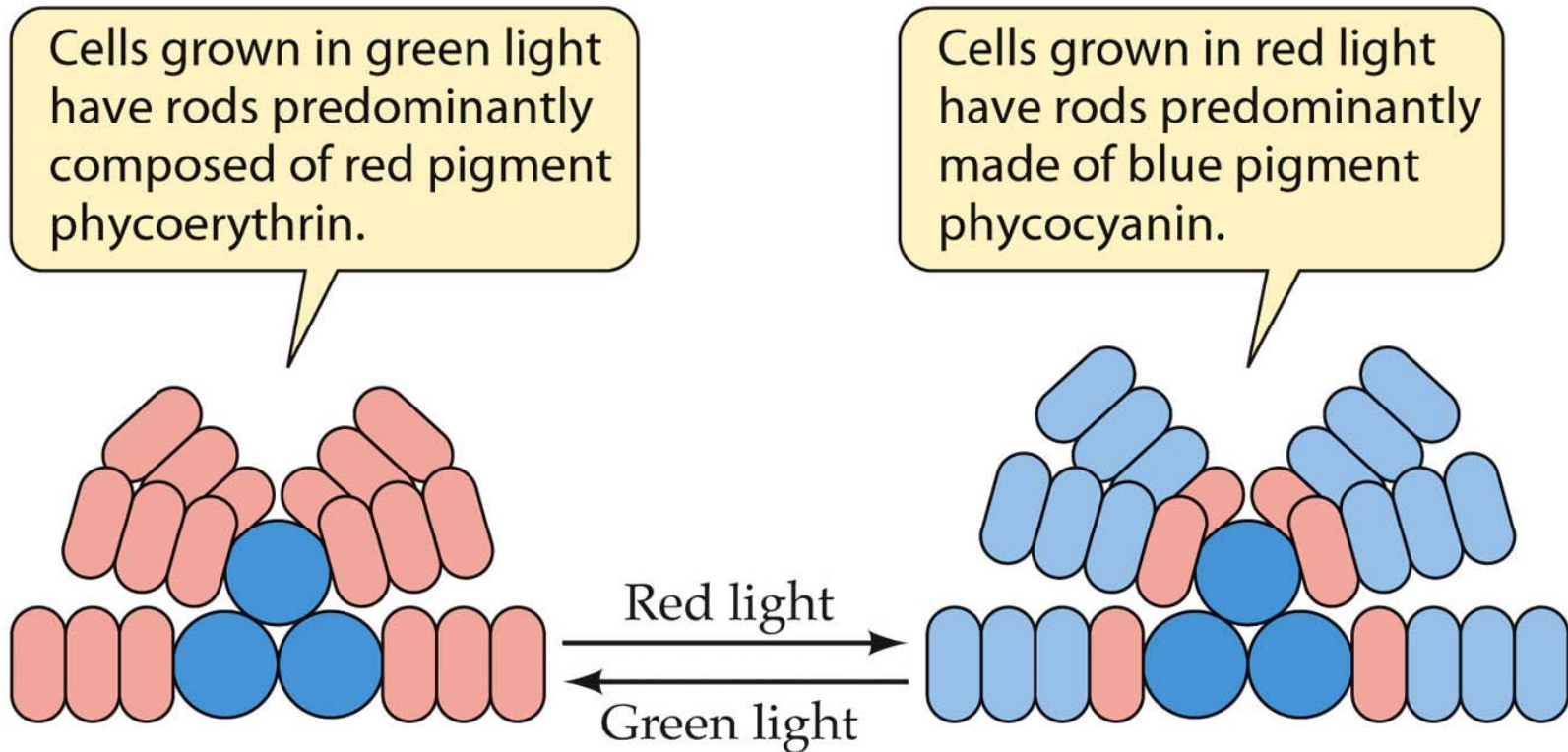
Structure and Location of Phycobilisomes



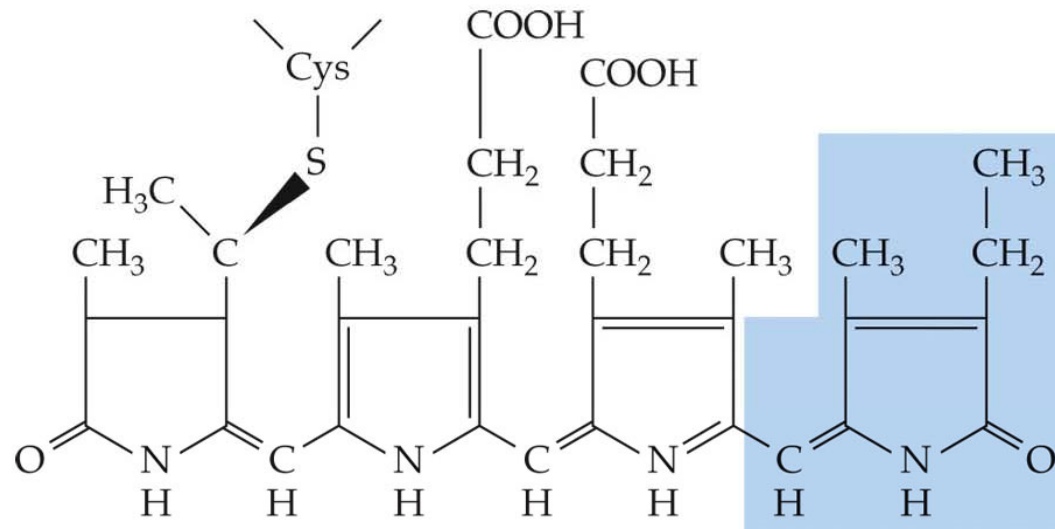
Phycobilisome of cyanobacteria



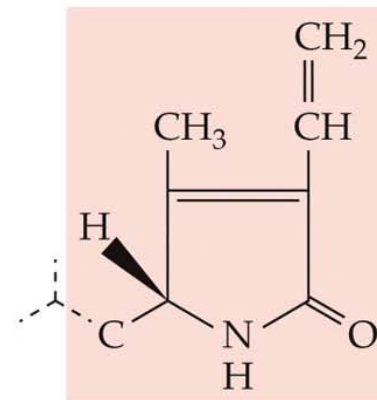
Chromatic adaptation of a phycobilisome



Chromophores of phycobilisomes

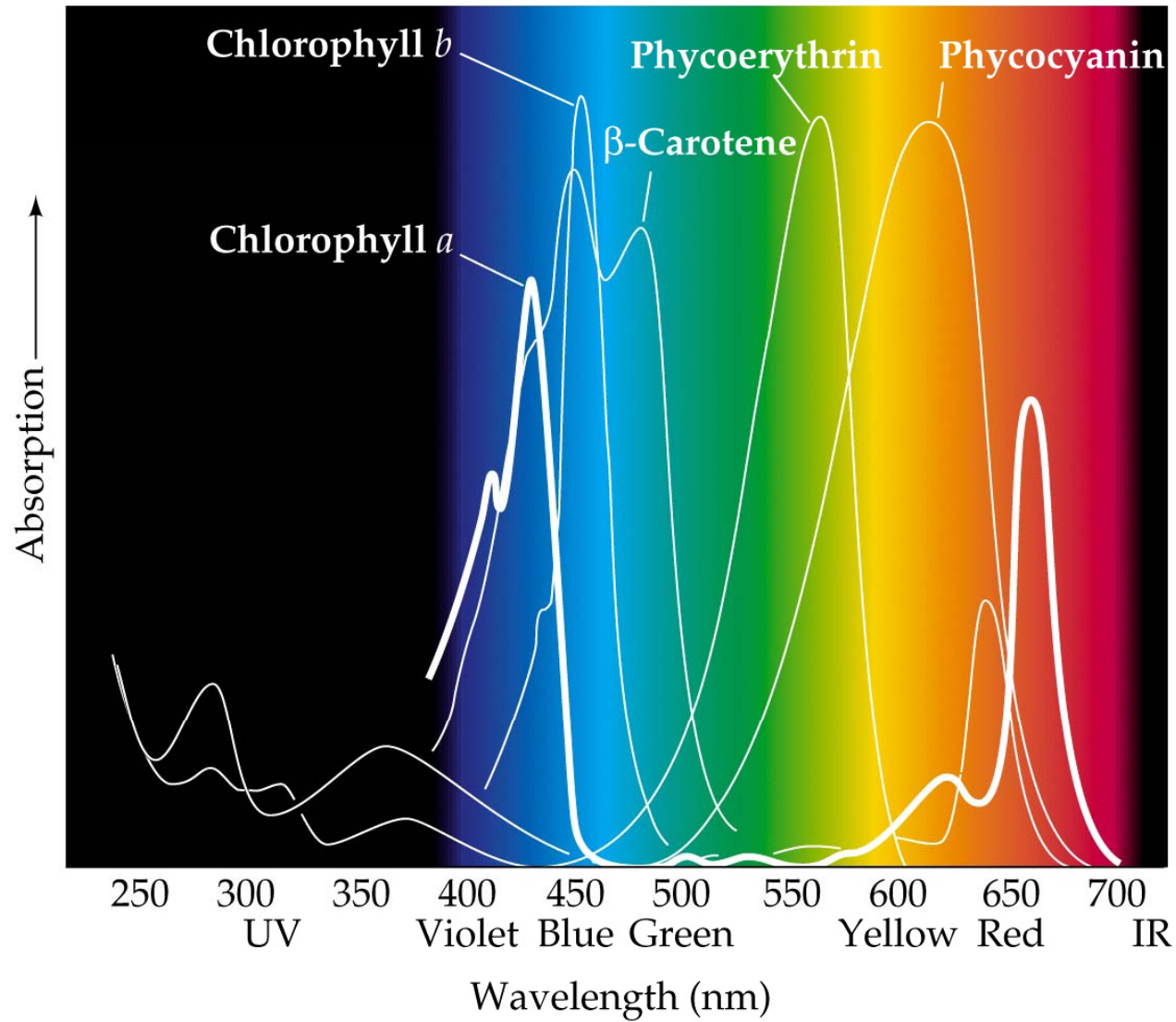


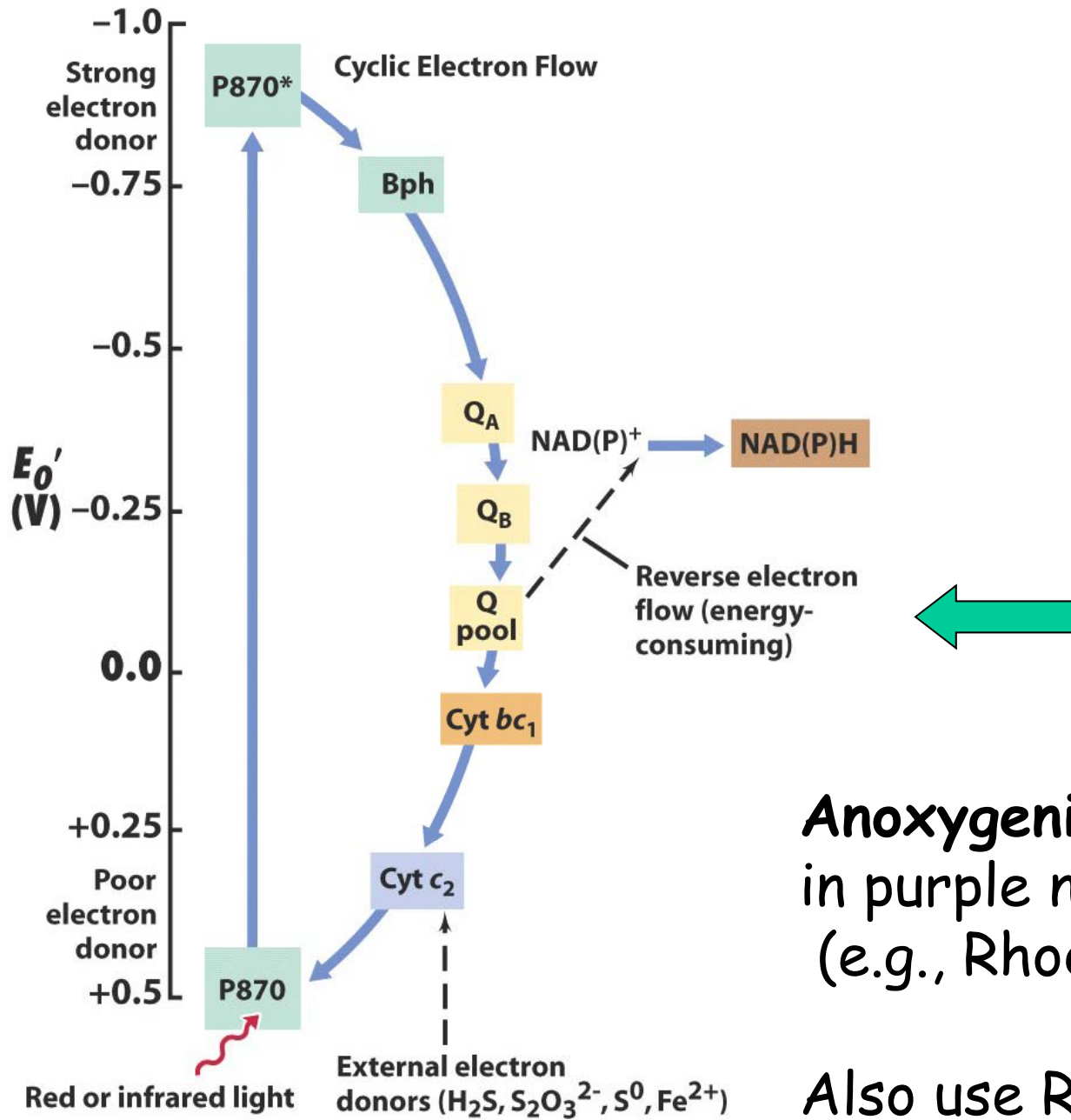
Phycocyanin (blue)



Phycoerythrin (red)

Absorption Spectra for Cyanobacteria

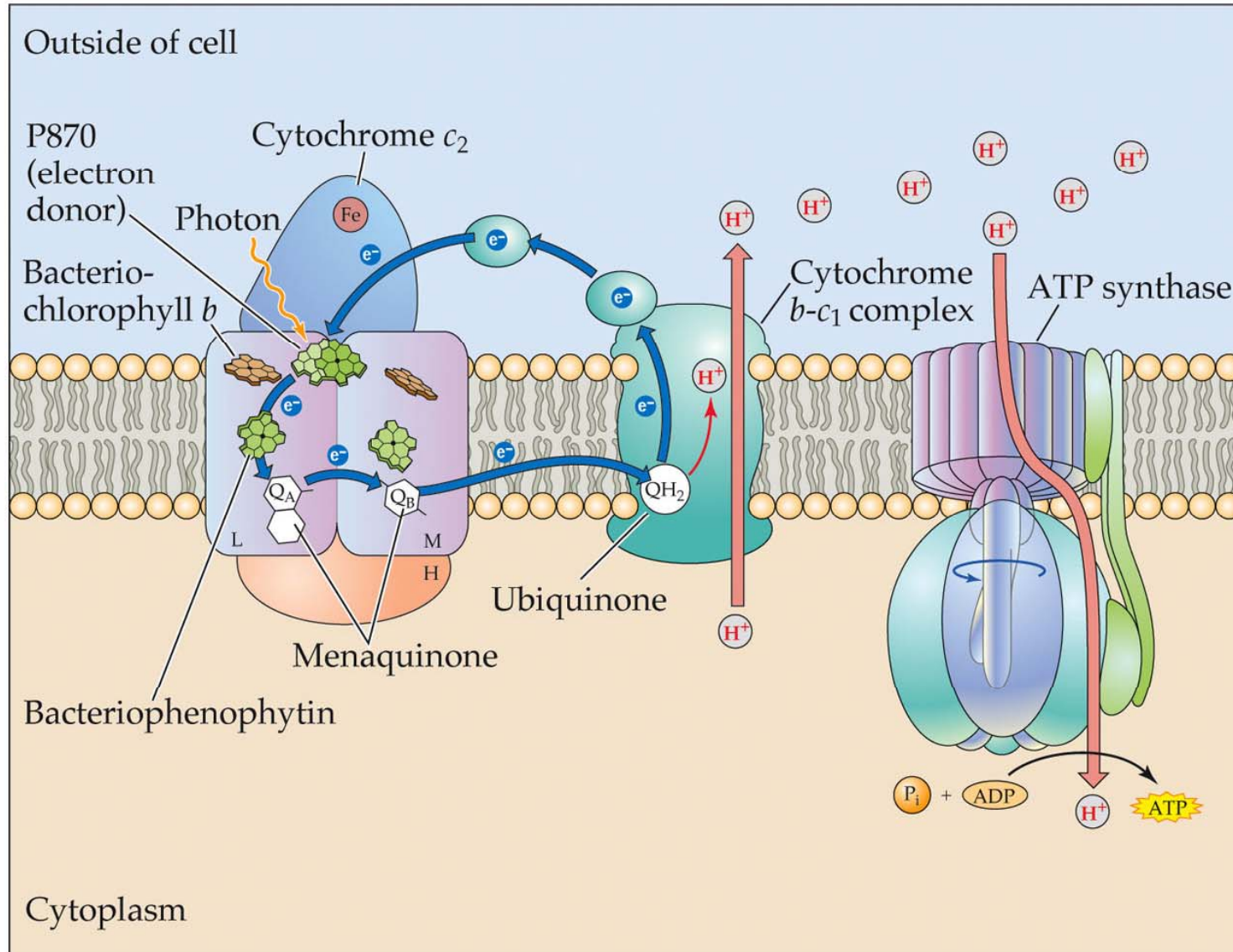




Anoxygenic photosynthesis
 in purple nonsulfur bacteria
 (e.g., Rhodobacter)

Also use REF: GNBS & PSBs

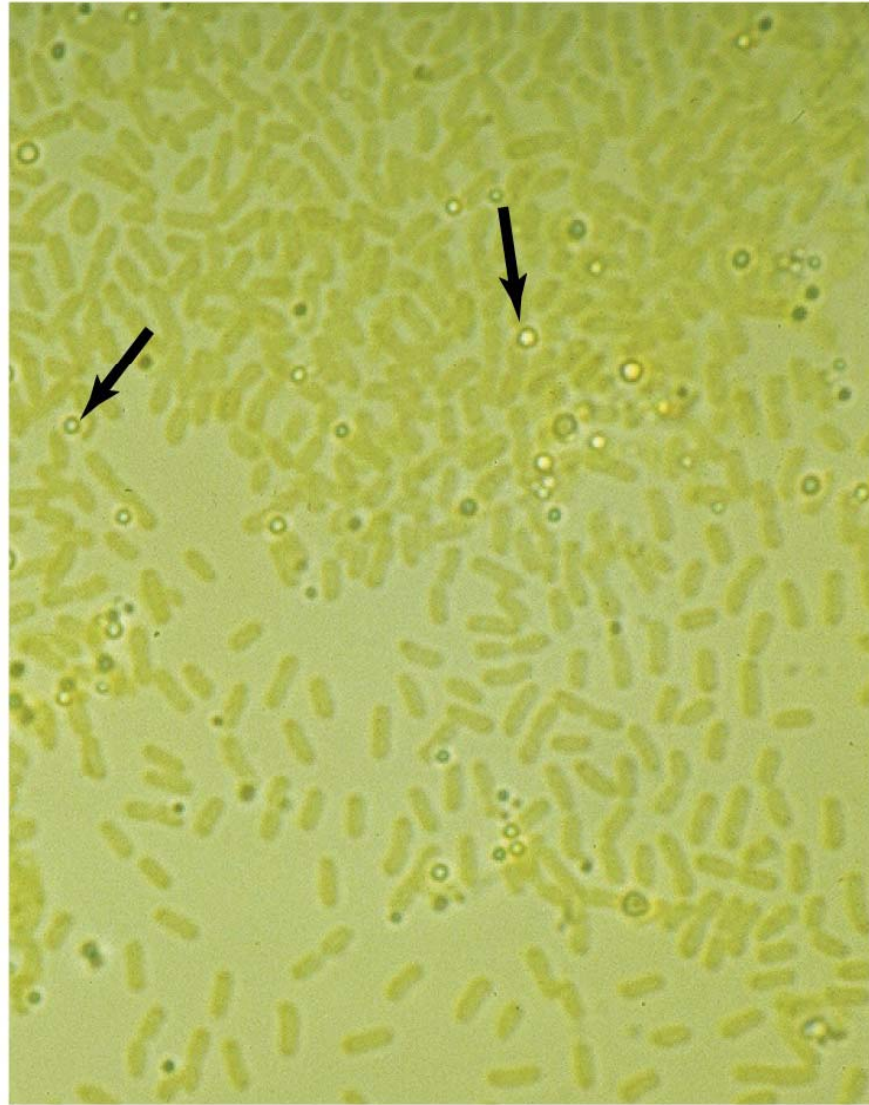
Reaction center of purple nonsulfur bacteria



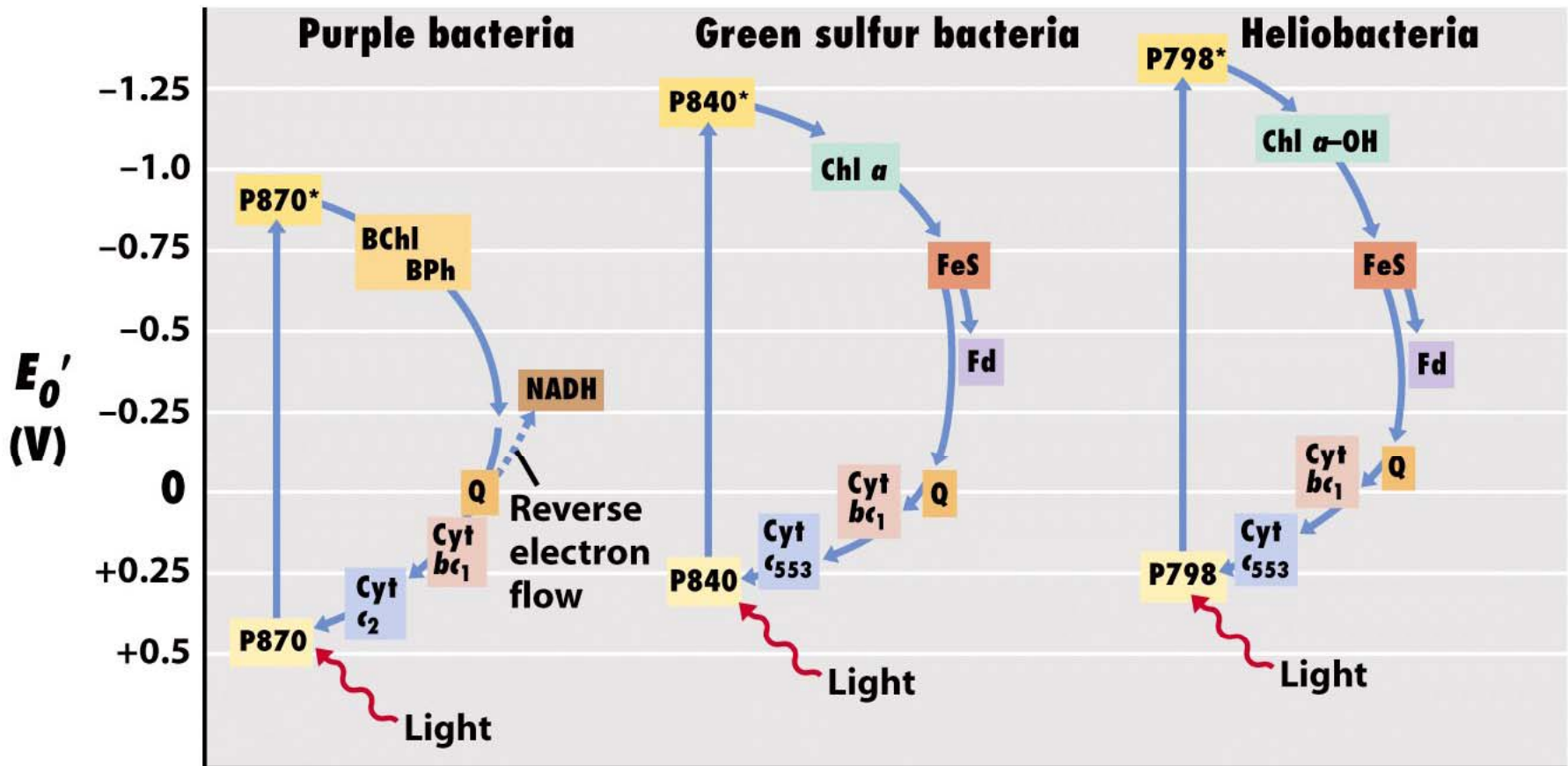
Sulfur granules in purple sulfur bacteria
e.g., *Chromatium*

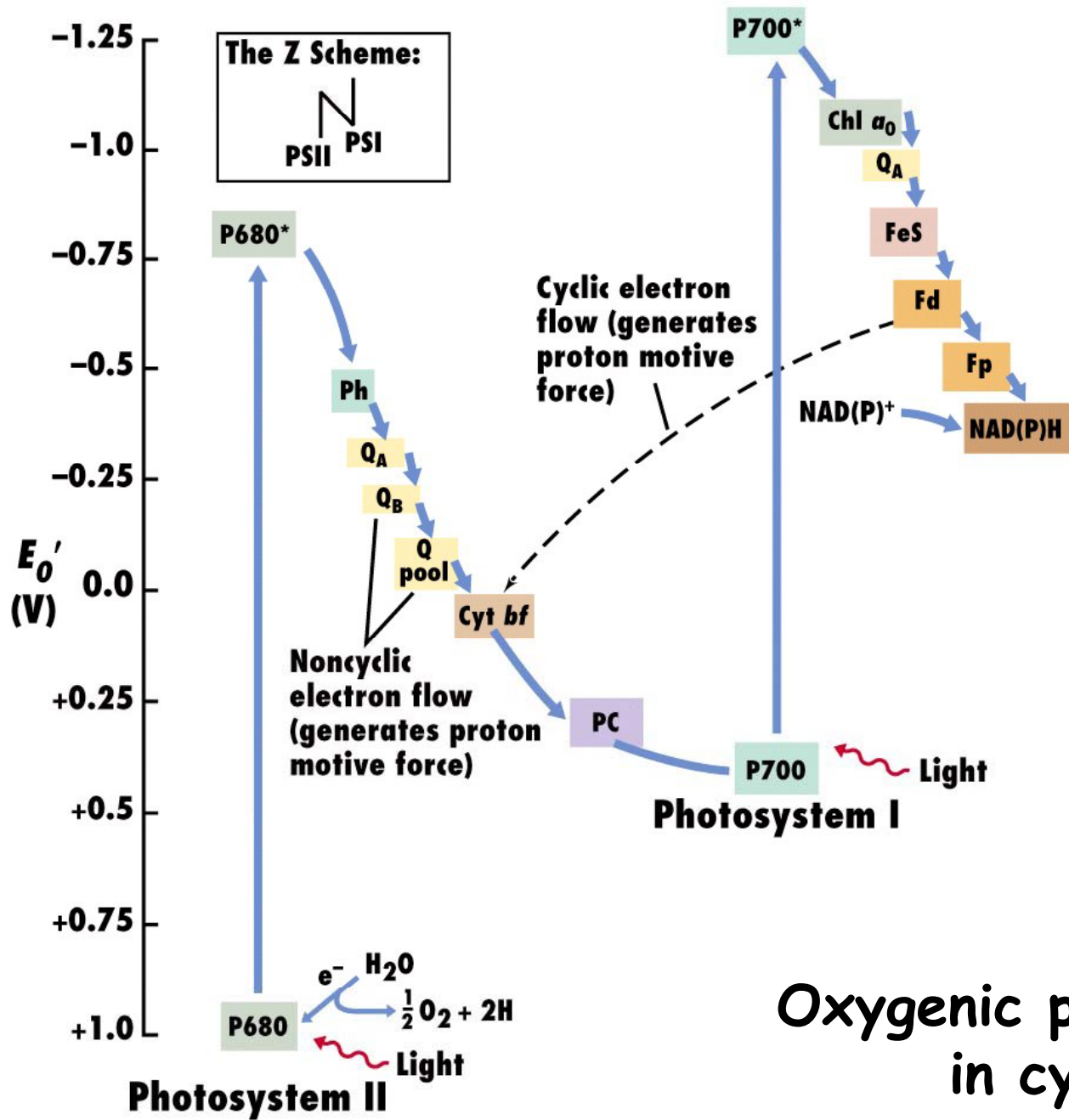


Sulfur granules in green sulfur bacteria
e.g., Chlorobium



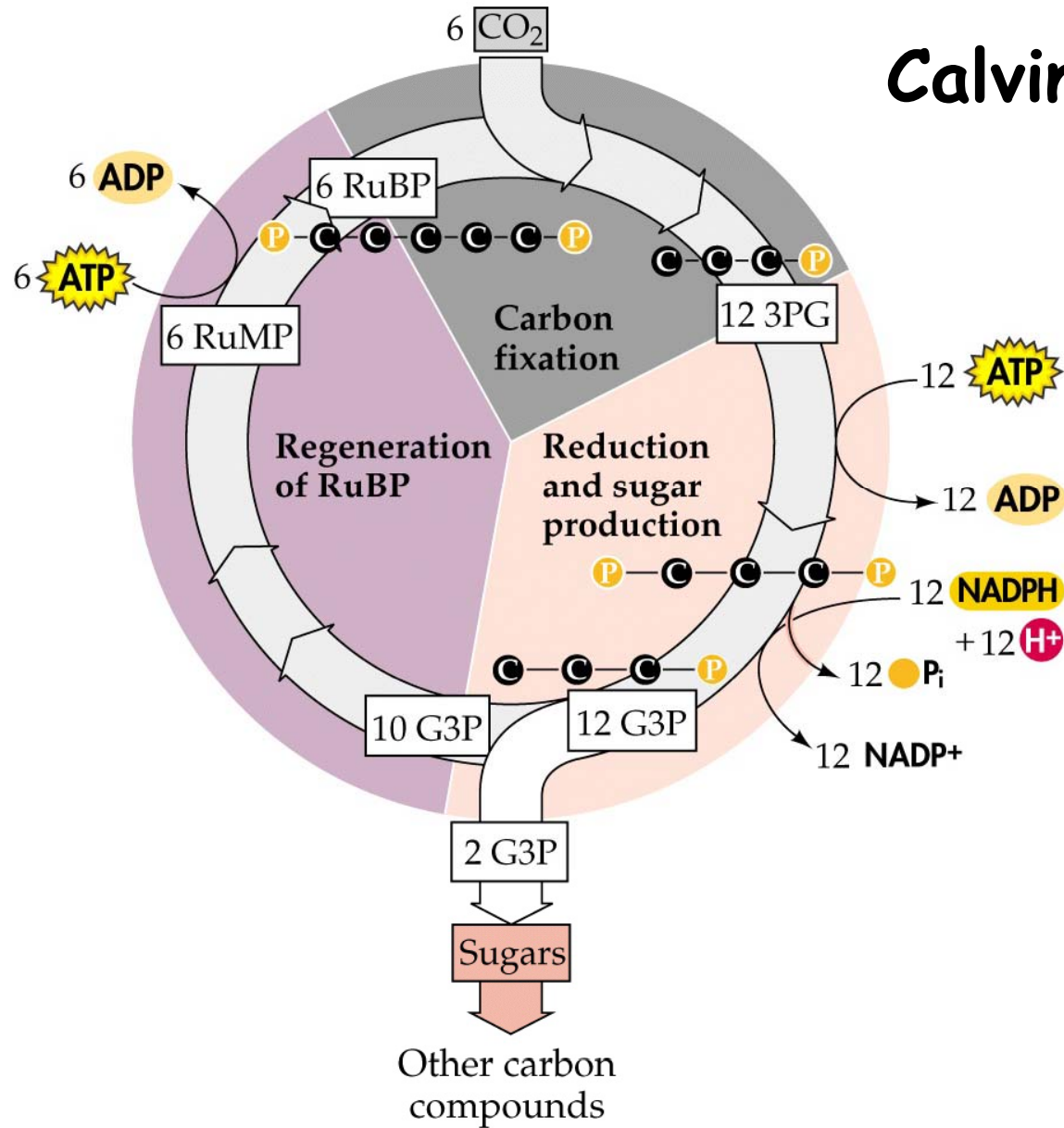
Electron flow in phototrophs



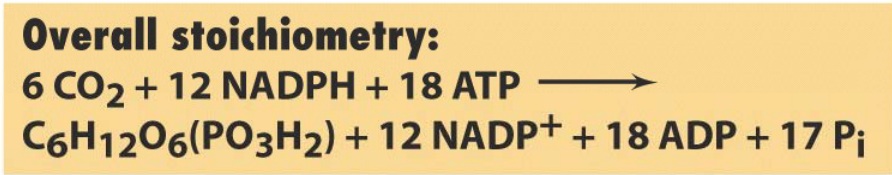
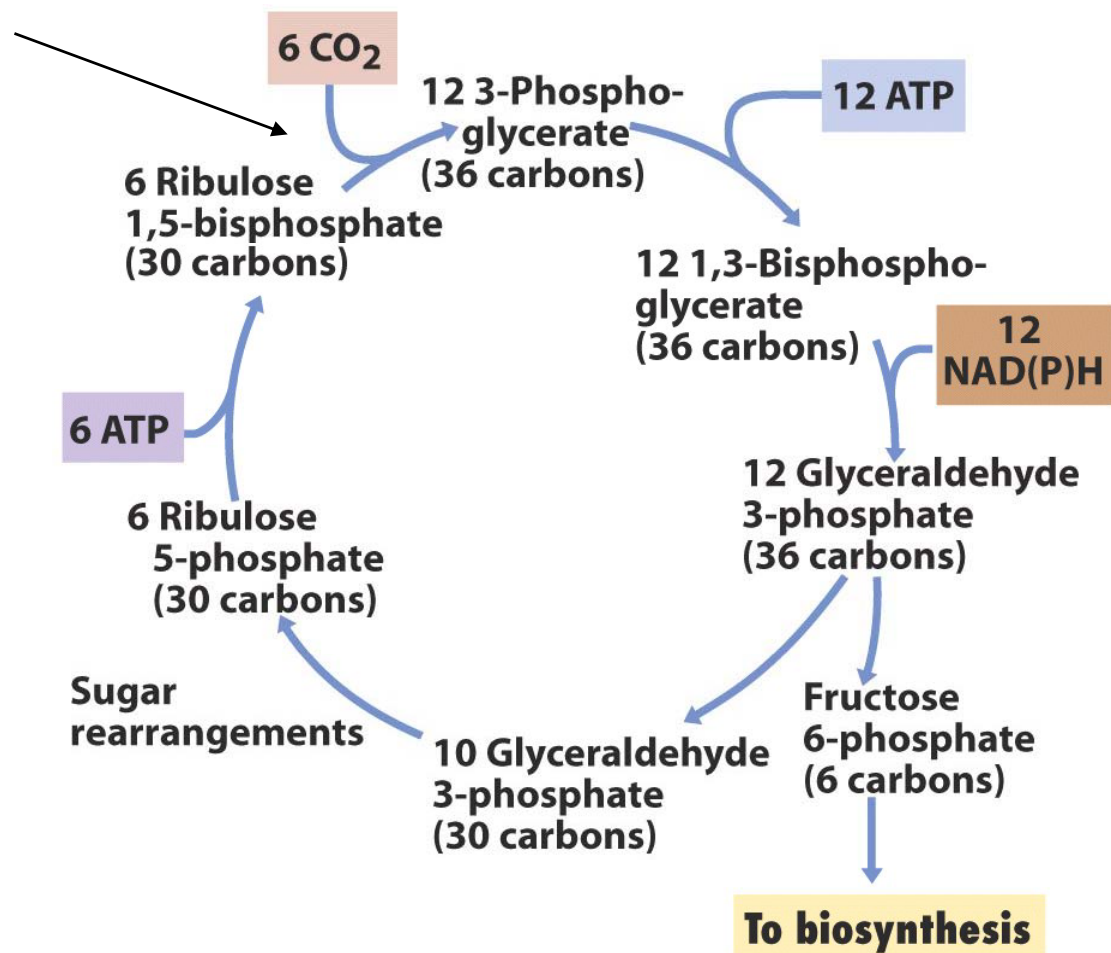


Oxygenic photosynthesis in cyanobacteria

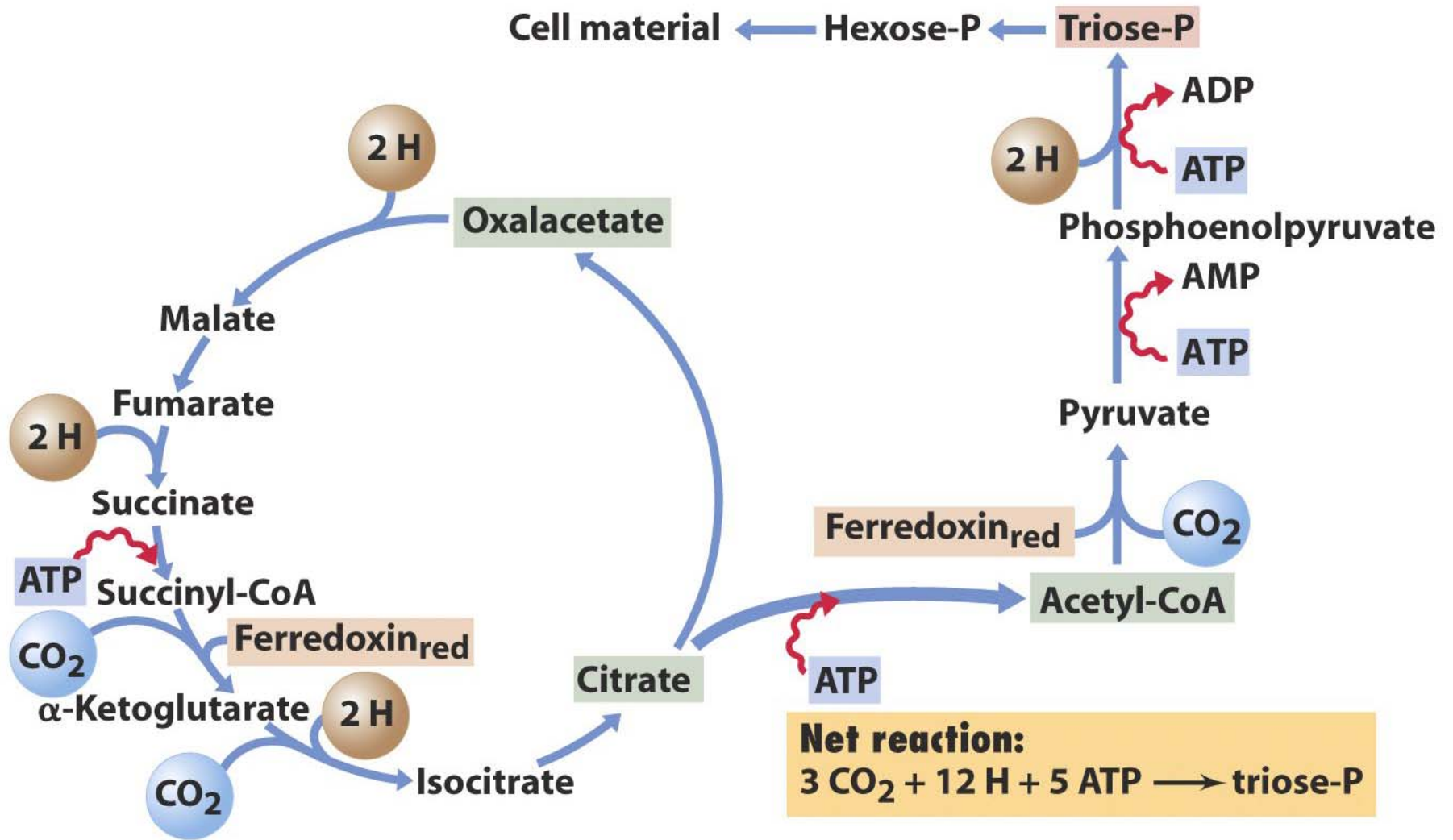
Calvin Cycle



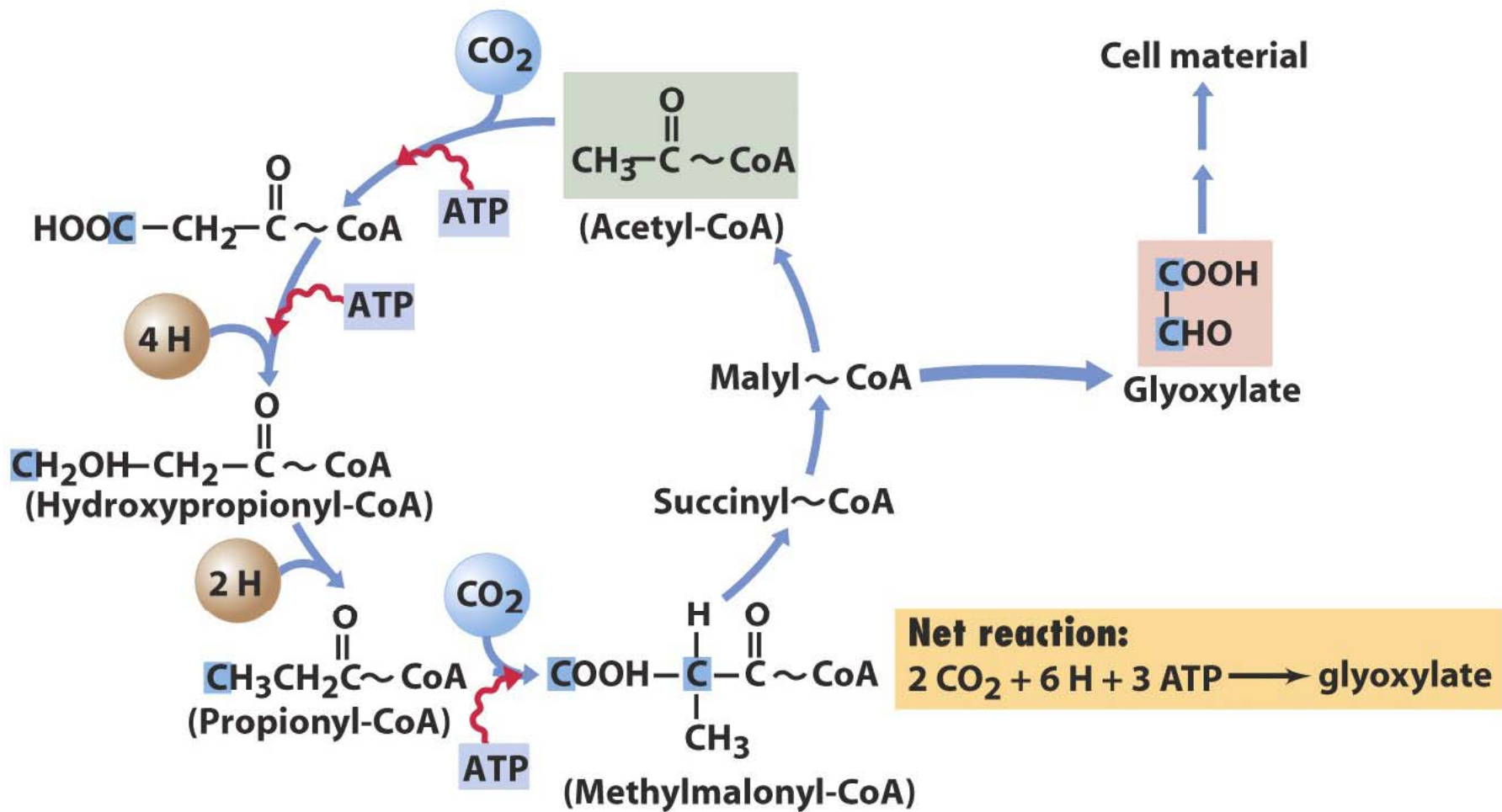
RuBisCo



Reverse TCA in GSBs

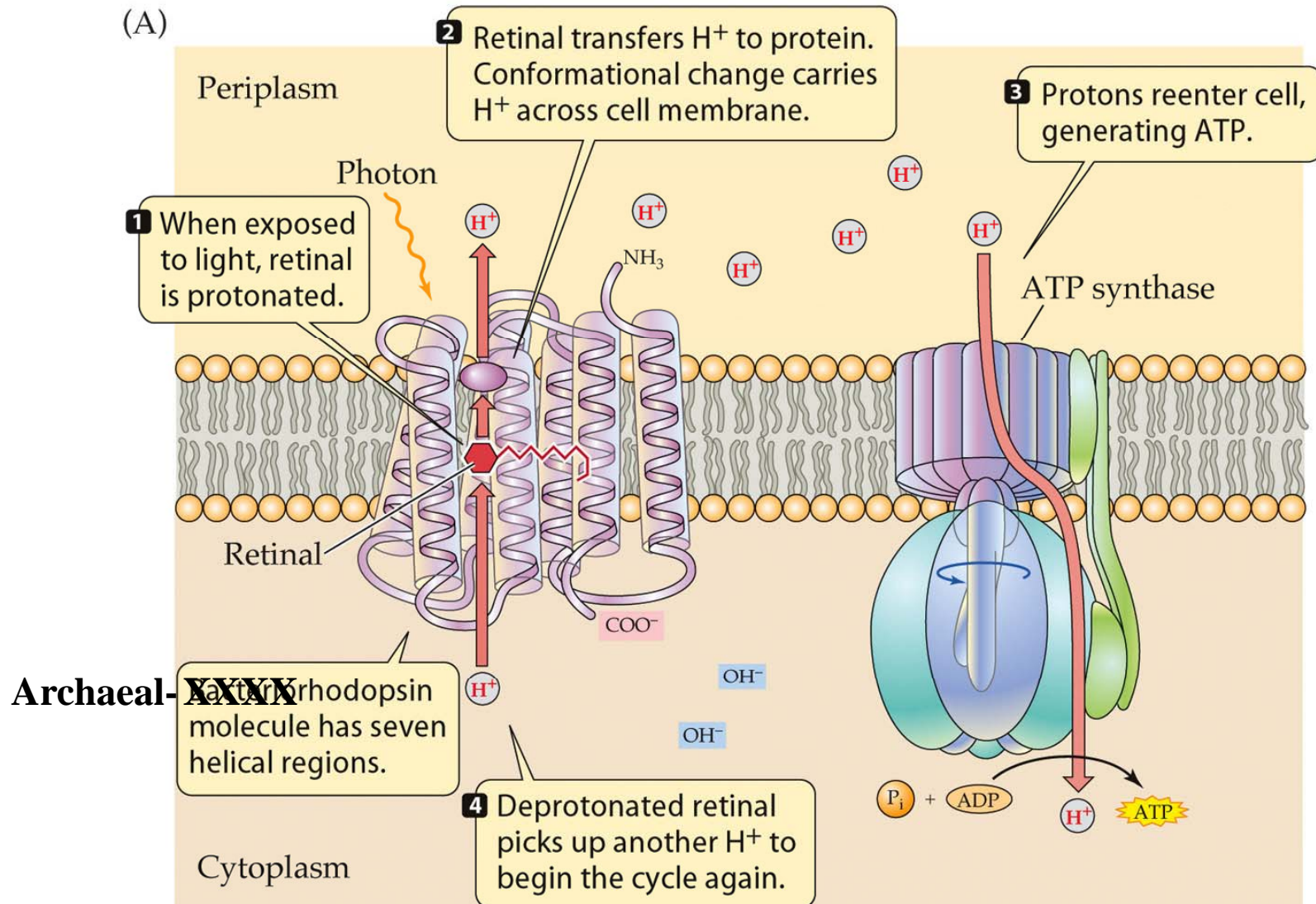


Hydroxypropionate in GNBs

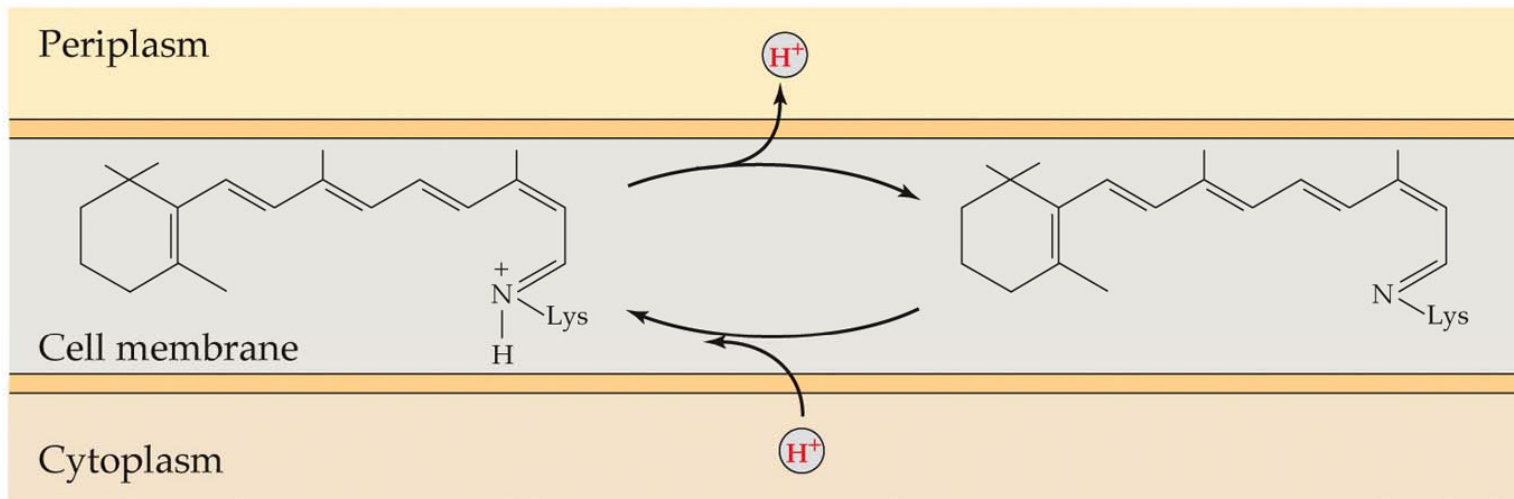




A light-driven proton pump of halophilic archaea



Light-driven proton pump of halophilic archaea



Archaeal rhodopsin: retinal structure



Proteorhodopsin in marine *Bacteria* and *Archaea*

