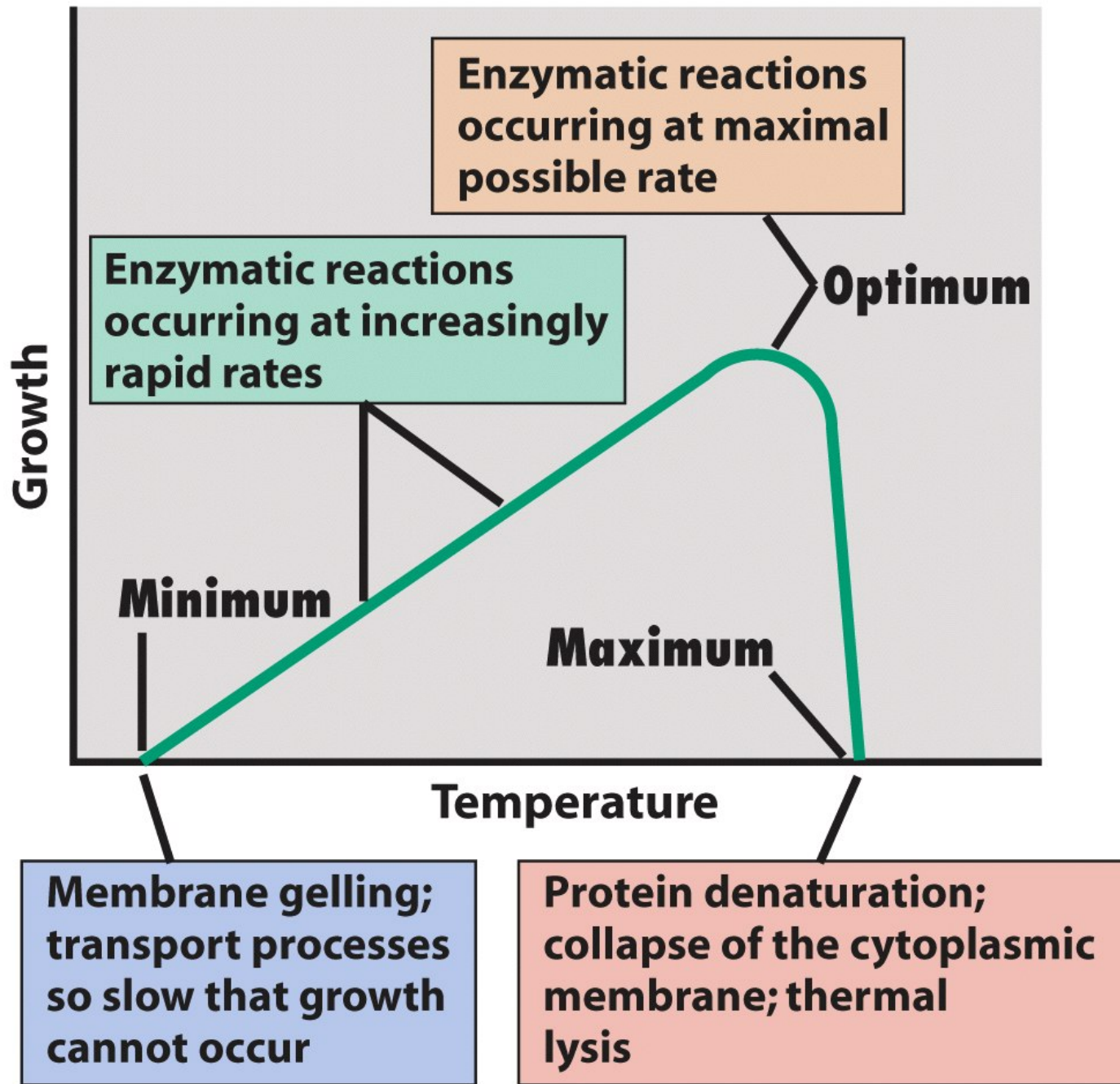


Microbial Growth

Environmental Forcing Functions:

- Temperature: Psychrophile, Mesophile, Thermophile & Hyperthermophile
Cardinal Temps: Min*, Max, & Optimal*
Q₁₀ Rule: 10°C rise will double the growth rate*
- Pressure: Barophiles (Most are also psychrophiles!)
Found only in the deep ocean....so far



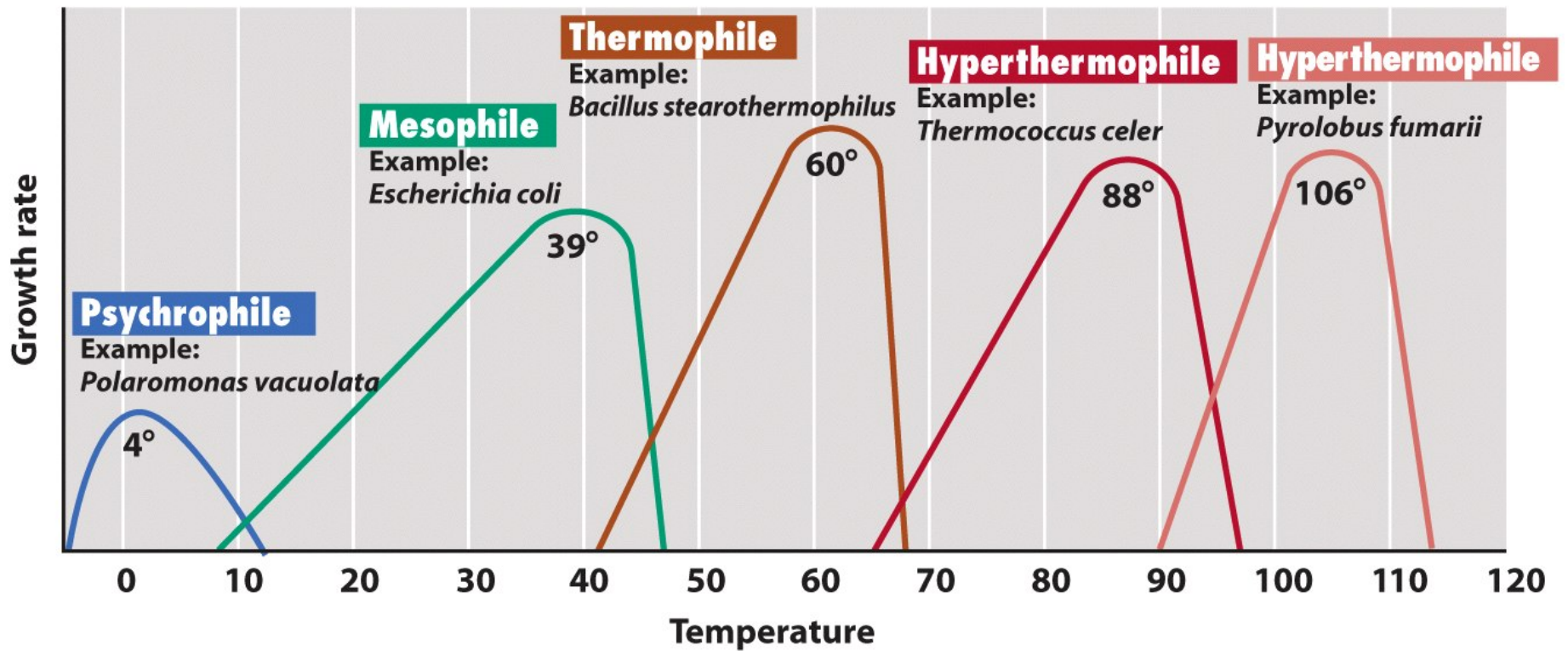
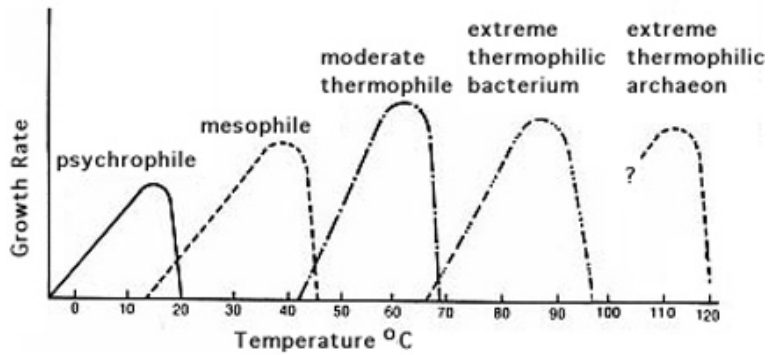
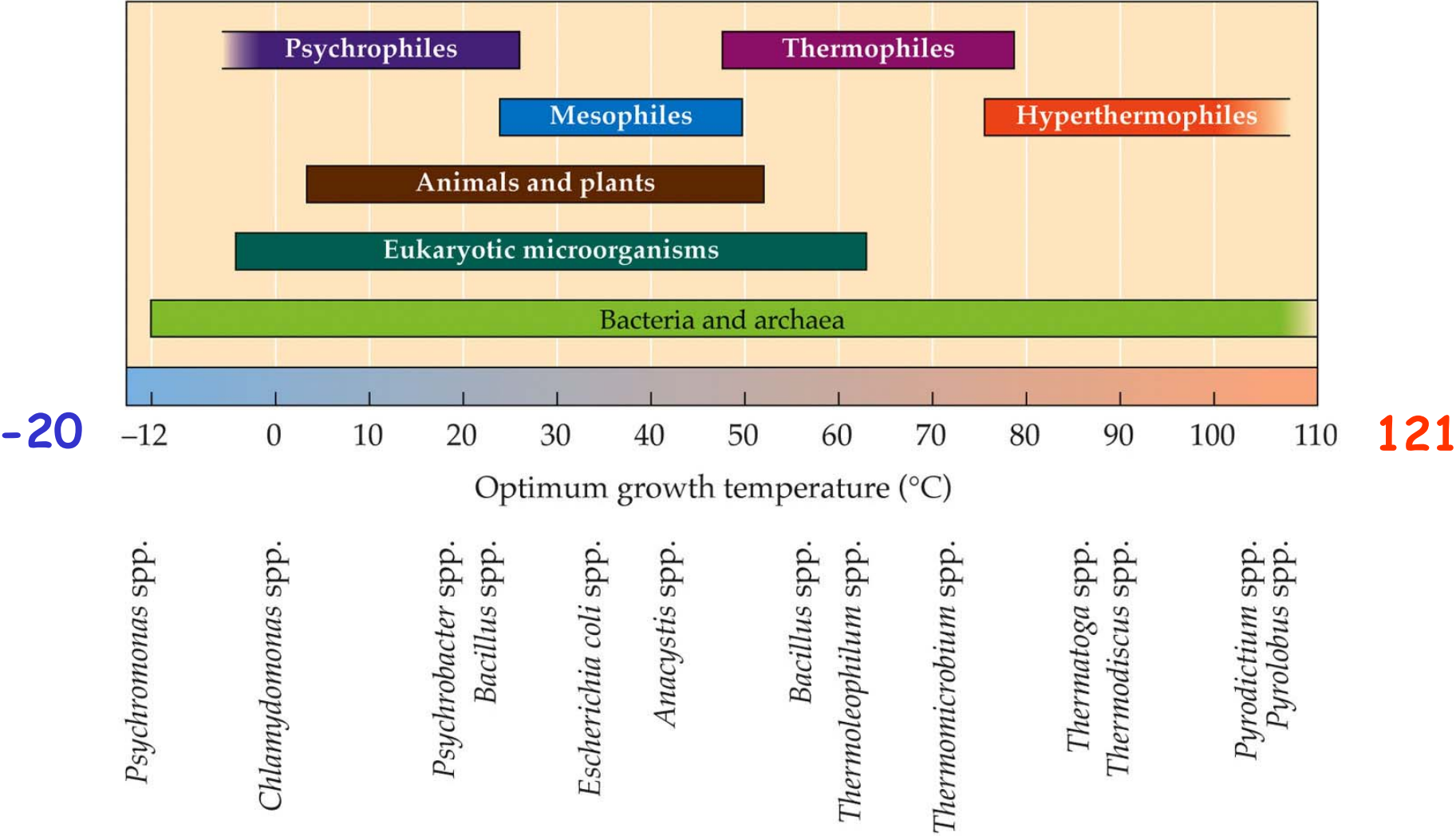


Table 6.3 Temperature ranges for growth of *Bacteria* and *Archaea*

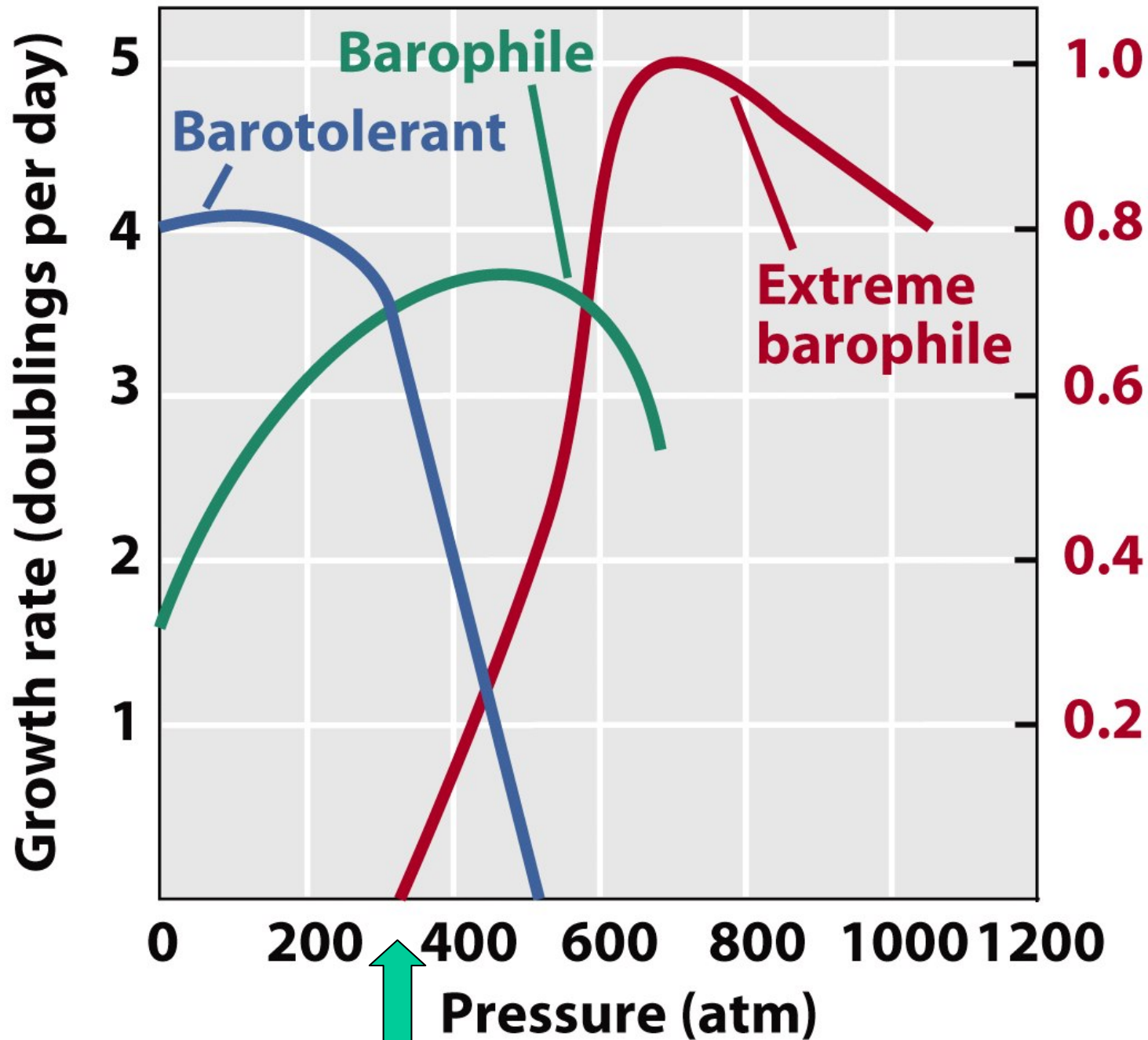
Species	Range (°C)
Psychrophiles	
<i>Cytophaga psychrophila</i>	4–20
<i>Bacillus insolitus</i>	<0–25
<i>Aquaspirillum psychrophilum</i>	2–26
Mesophiles	
<i>Escherichia coli</i>	10–40
<i>Lactobacillus lactis</i>	18–42
<i>Bacillus subtilis</i>	22–40
<i>Pseudomonas fluorescens</i>	4–40
Thermophiles	
<i>Bacillus thermoleovorans</i>	42–75
<i>Thermoleophilum album</i>	45–70
<i>Thermus aquaticus</i>	40–79
<i>Chloroflexus aurantiacus</i>	45–70
Hyperthermophiles (Archaea)	
<i>Hyperthermus butylicus</i>	85–108
<i>Methanothermus fervidus</i>	65–97
<i>Pyrodictium occultum</i>	80–110
<i>Thermococcus celer</i>	70–95

Growth temperature ranges for various life forms









Ave. Ocean Depth

Microbial Growth

Environmental Forcing Functions:

- pH: acidophiles & alkaliphiles
cytoplasm still near neutral
- Eh: available electron donors & terminal electron acceptors
affects the chemistry of the environment

Acidophiles

Increasing acidity

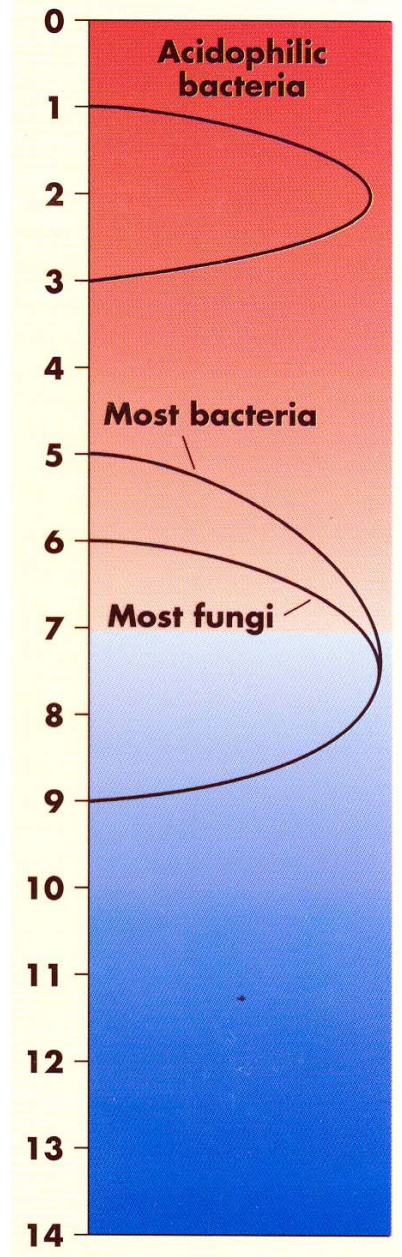
Alkaliphiles

Increasing alkalinity

Neutrality

pH	Example	Moles per liter of:	
		H ⁺	OH ⁻
0		1	10 ⁻¹⁴
1	Volcanic soils, waters Gastric fluids	10 ⁻¹	10 ⁻¹³
2	Lemon juice Acid mine drainage	10 ⁻²	10 ⁻¹²
3	Vinegar Rhubarb Peaches	10 ⁻³	10 ⁻¹¹
4	Acid soil Tomatoes	10 ⁻⁴	10 ⁻¹⁰
5	American cheese Cabbage	10 ⁻⁵	10 ⁻⁹
6	Peas Corn, salmon, shrimp	10 ⁻⁶	10 ⁻⁸
7	Pure water	10 ⁻⁷	10 ⁻⁷
8	Seawater	10 ⁻⁸	10 ⁻⁶
9	Very alkaline natural soil	10 ⁻⁹	10 ⁻⁵
10	Alkaline lakes Soap solutions	10 ⁻¹⁰	10 ⁻⁴
11	Household ammonia Extremely alkaline	10 ⁻¹¹	10 ⁻³
12	soda lakes Lime (saturated solution)	10 ⁻¹²	10 ⁻²
13		10 ⁻¹³	10 ⁻¹
14		10 ⁻¹⁴	1

Bacterial/fungal growth



Growth rate

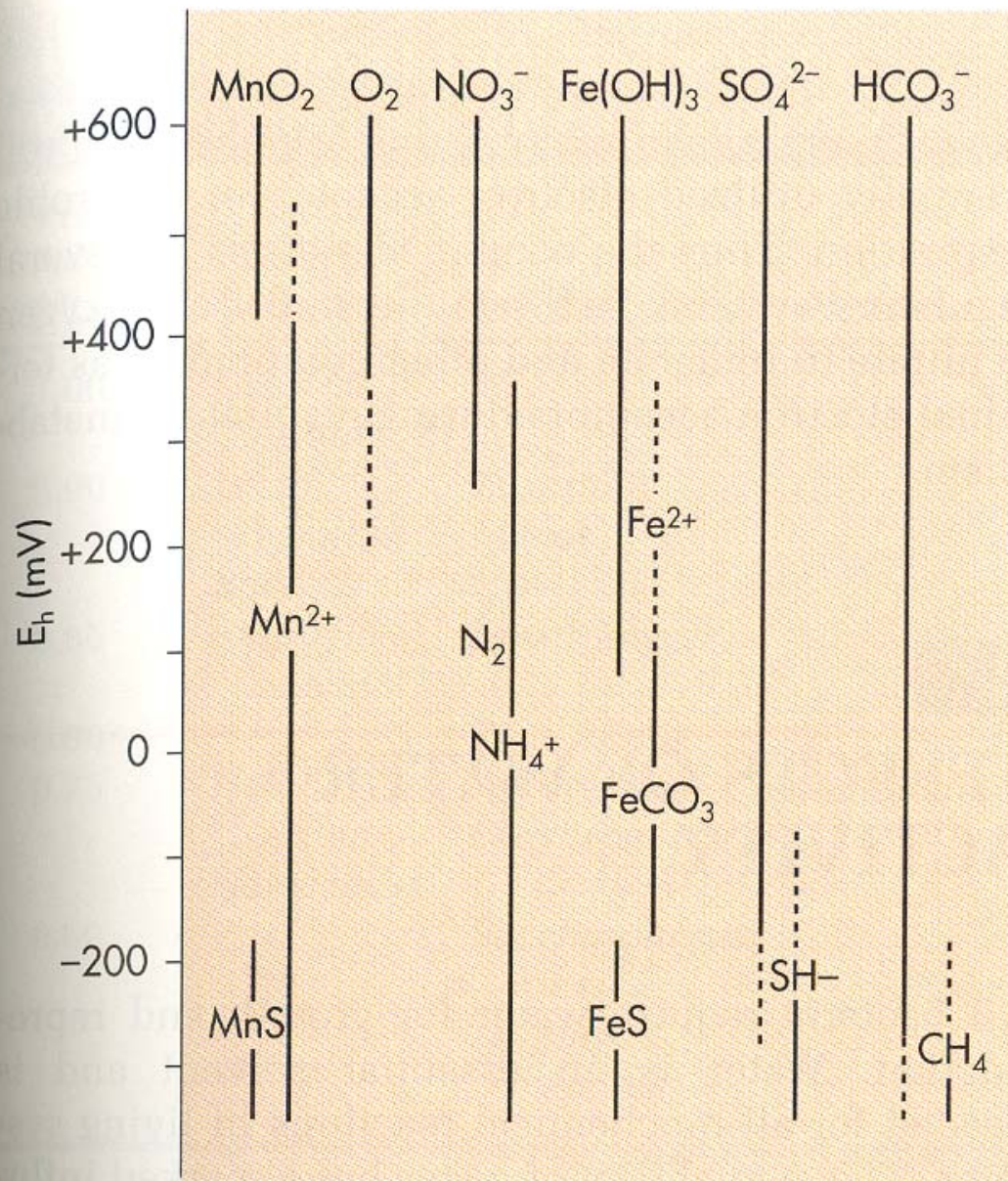
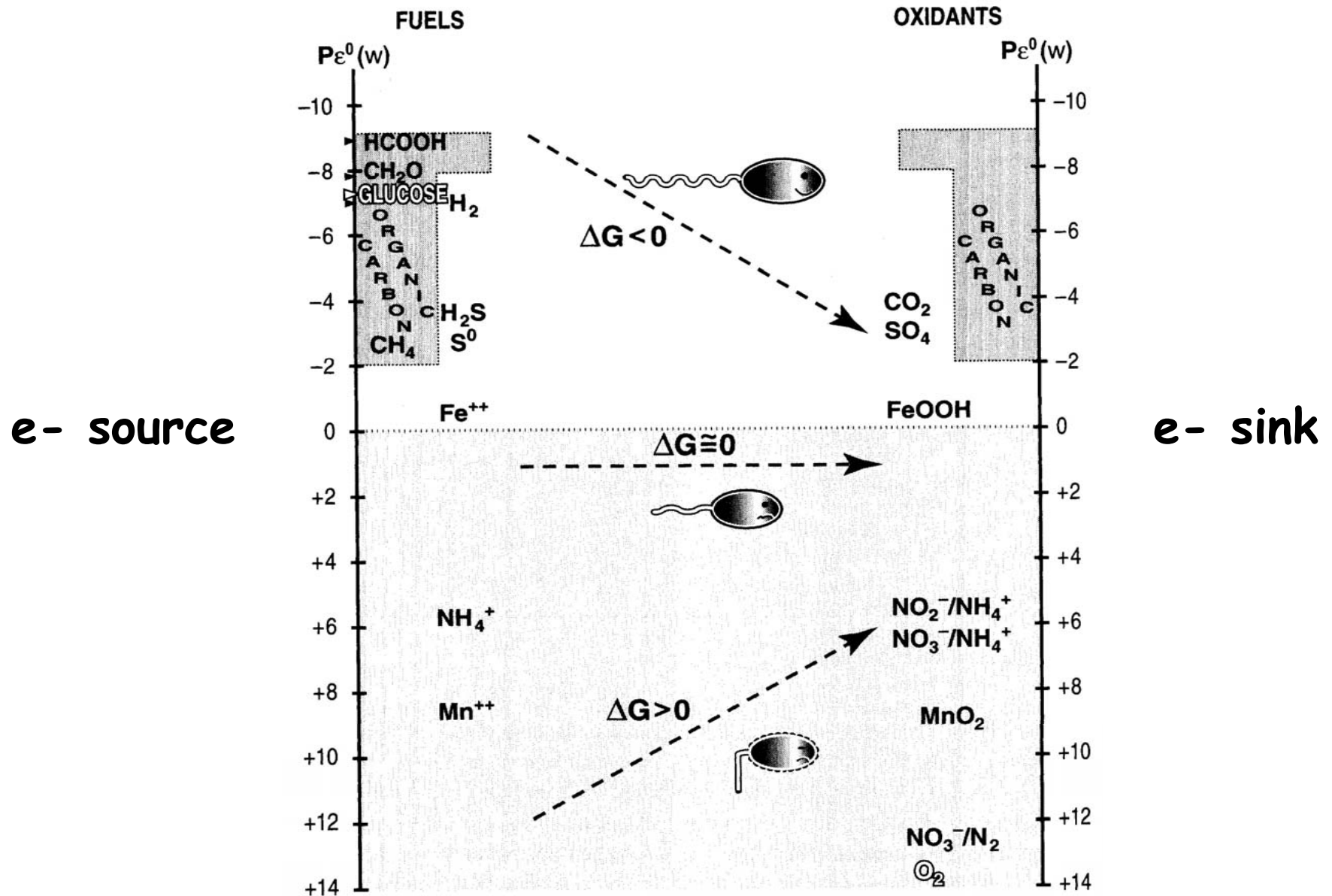


Fig. 9-26 E_h Values. Ranges of E_h values for various substances. In complex systems the reduction potential is influenced by the strongest oxidant, or reductant, in that system.

Thermodynamics: The Chemical Fuels and Oxidants of Life



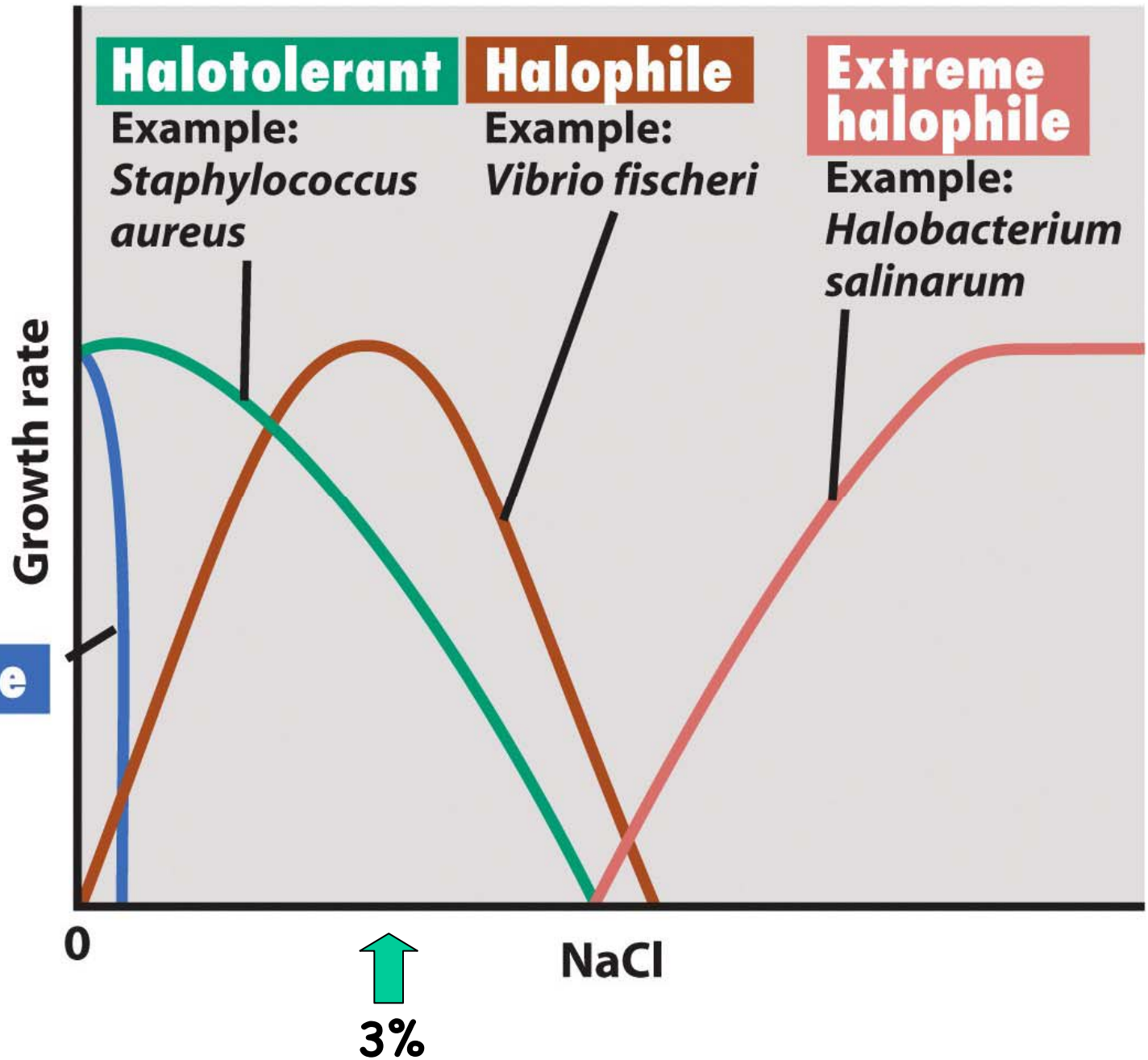
Microbial Growth

Environmental Forcing Functions:

- Salt: Halophiles
Compatible solutes: amino acid derivatives (e.g., proline & glycine), sugars, & alcohols.
- Water Activity: Xerophiles (live in very dry habitats)
Rem: All microbes are **osmotrophs**, must use organic material in solution!
- Oxygen Usage: aerobe, facultative (an)aerobe, microaerophile, obligate anaerobe
DeTox enzymes: Catalase, Peroxidase, SOD

Nonhalophile

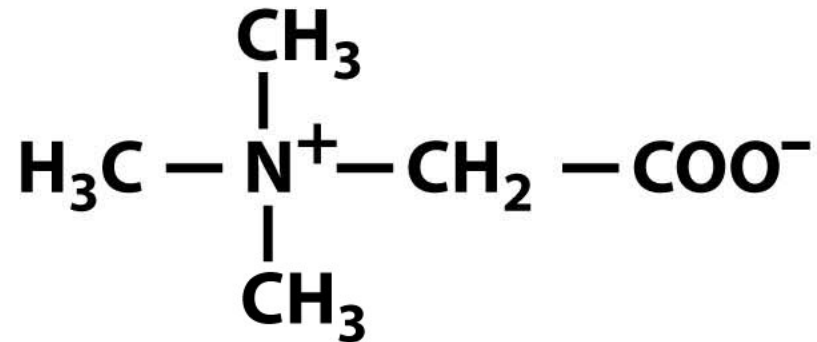
Example:
Escherichia coli



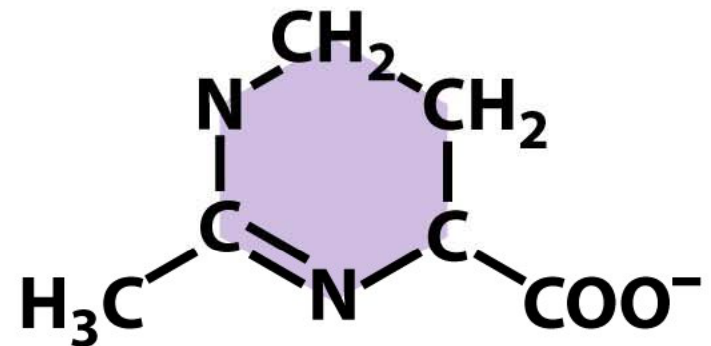
Compatible solutes

1. Amino acid-type and related solutes:

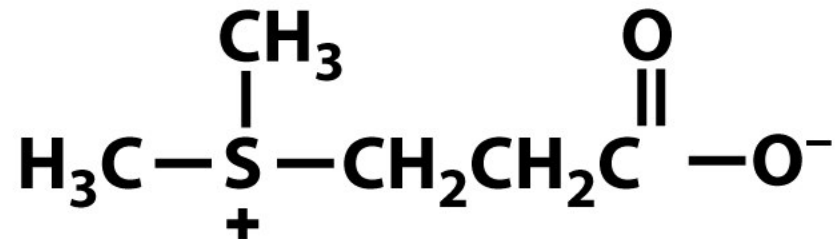
Glycine betaine



Ectoine



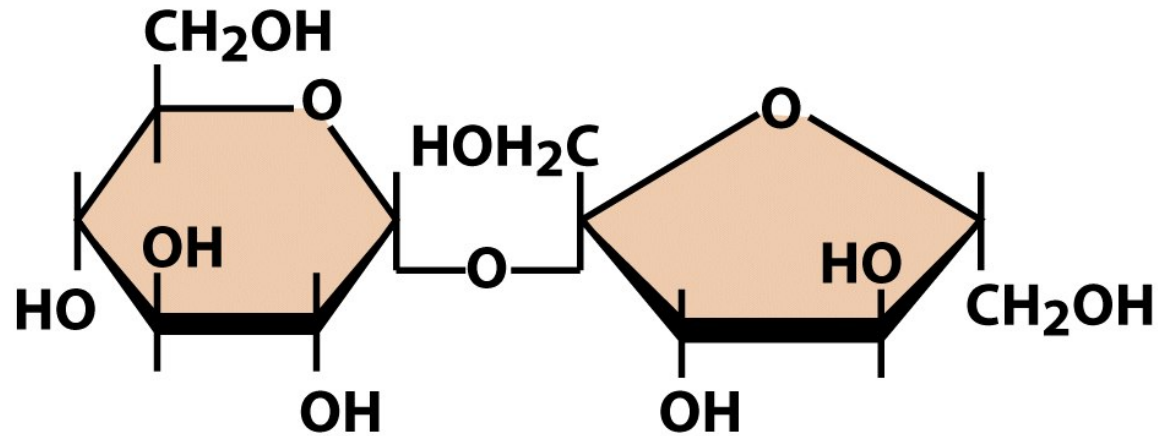
Dimethylsulfoniopropionate



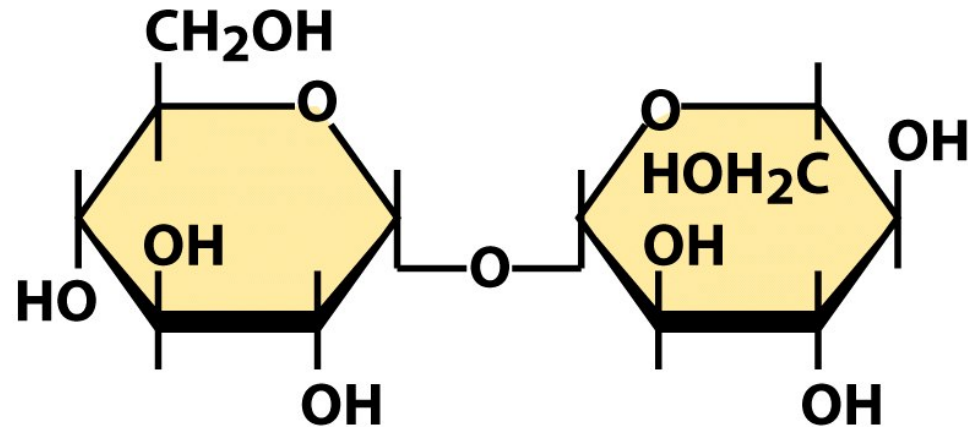
Compatible solutes

2. Carbohydrate-type solutes:

Sucrose



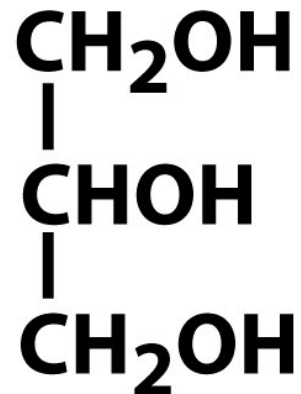
Trehalose



Compatible solutes

3. Alcohol-type solutes:

Glycerol



Mannitol

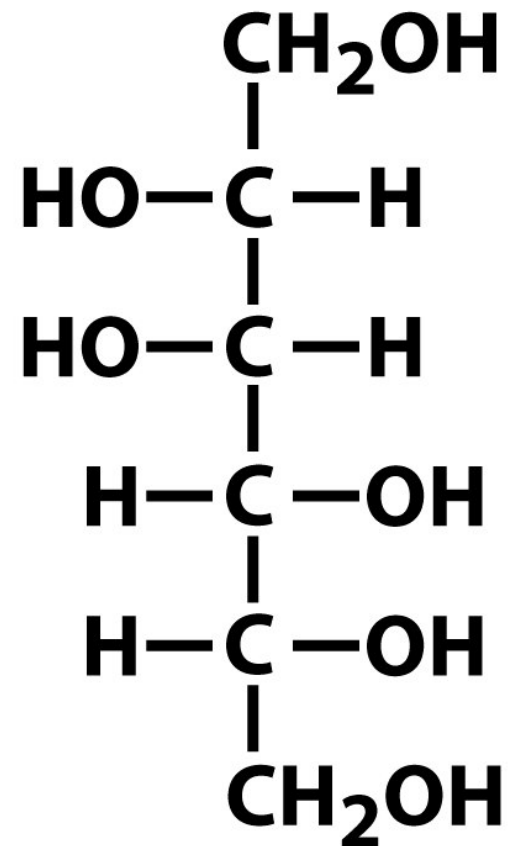
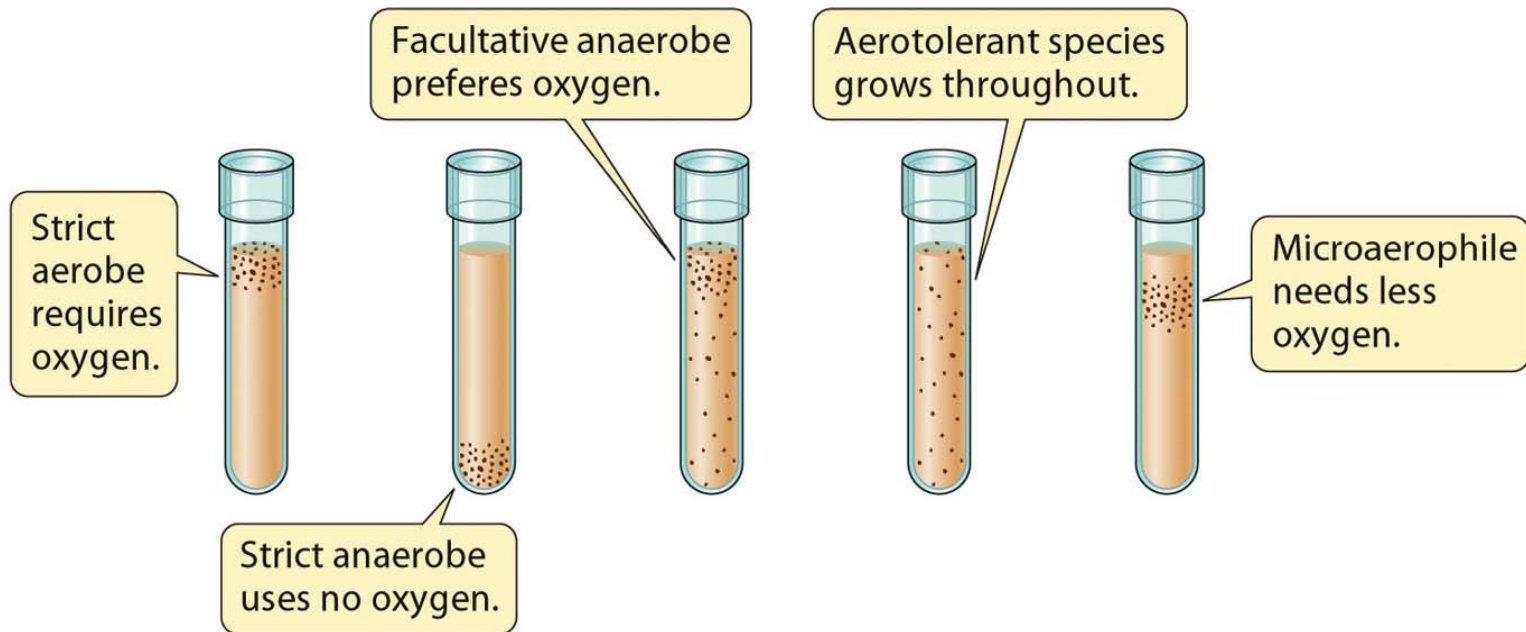


Table 6.4**Tolerance of selected *Bacteria* and *Archaea* for decreased water activity a_w**

Type	Organisms	a_w
Nonhalophiles	<i>Aquaspirillum</i> and <i>Caulobacter</i>	1.00
Marine forms	Pseudomonads and <i>Alteromonas</i>	0.98
Moderate halophiles	<i>Vibrio</i> species and gram-positive cocci	0.91
Extreme halophiles	<i>Halobacterium</i> and <i>Halococcus</i>	0.75

Response of bacterial growth to oxygen availability



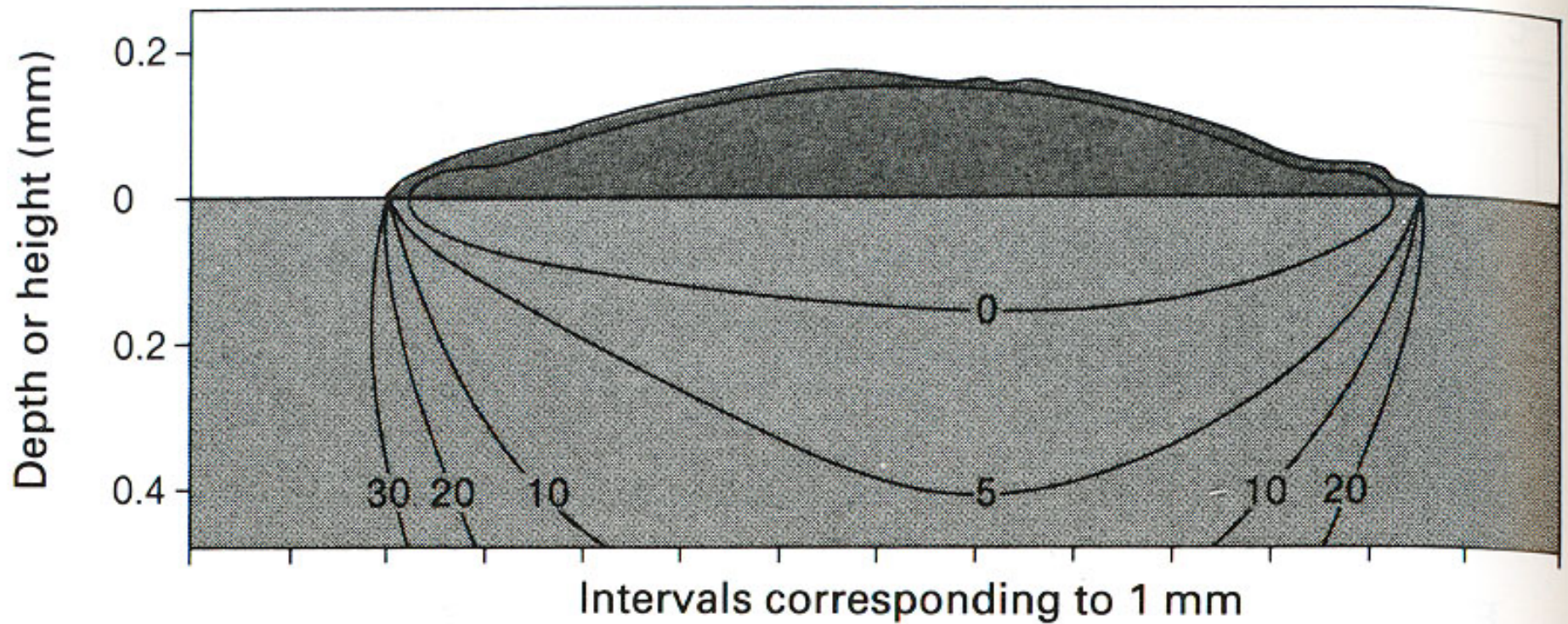



Fig. 6.3. *Oxygen distribution under and inside a colony of Escherichia coli after growth on complex agar.*



Table 9-5 Electronic States of Oxygen

Form	Formula	Simplified Electronic Structure	Spin of Outer Electrons	
Triplet oxygen (normal atmospheric form)	${}^3\text{O}_2$	$\dot{\text{O}}-\dot{\text{O}}$	\uparrow	\uparrow
Singlet oxygen	${}^1\text{O}_2$	$\dot{\text{O}}-\dot{\text{O}}$	$\downarrow\uparrow$	$\uparrow\downarrow$
 Nasty!			\uparrow	\downarrow
Superoxide free radical	O_2^-	$\ddot{\text{O}}-\dot{\text{O}}$	$\downarrow\uparrow$	\uparrow
Peroxide	O_2^{2-}	$\ddot{\text{O}}-\ddot{\text{O}}$	$\downarrow\uparrow$	$\downarrow\uparrow$

4 electron reduction of O₂ to water

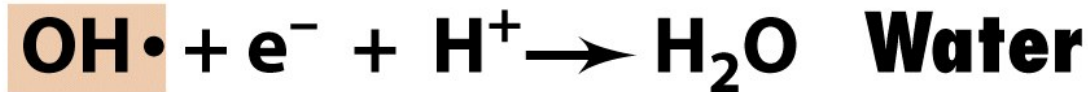
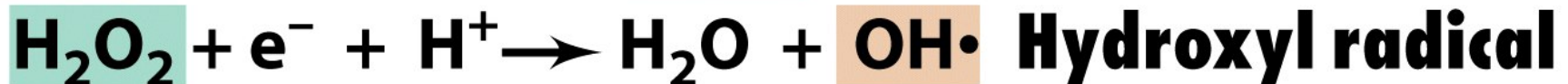
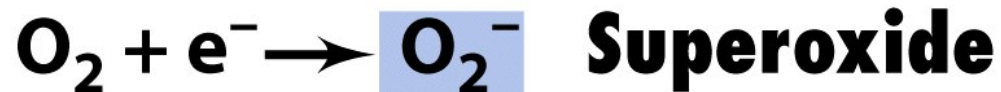
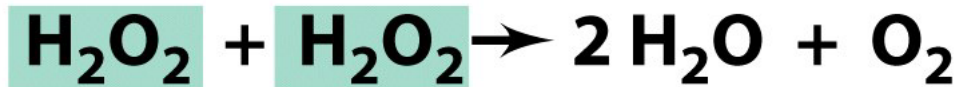


Table 9-6 Bacterial Enzymes that Protect the Cell Against Toxic Forms of Oxygen

Microorganism	Catalase	Superoxide Dismutase
Aerobe	+	+
Facultative anaerobe	+	+
Microaerophile	-	+
Obligate anaerobe	-	-

(a) Catalase:



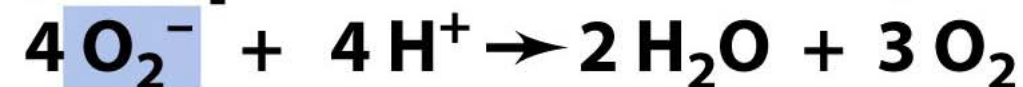
(b) Peroxidase:



(c) Superoxide dismutase:



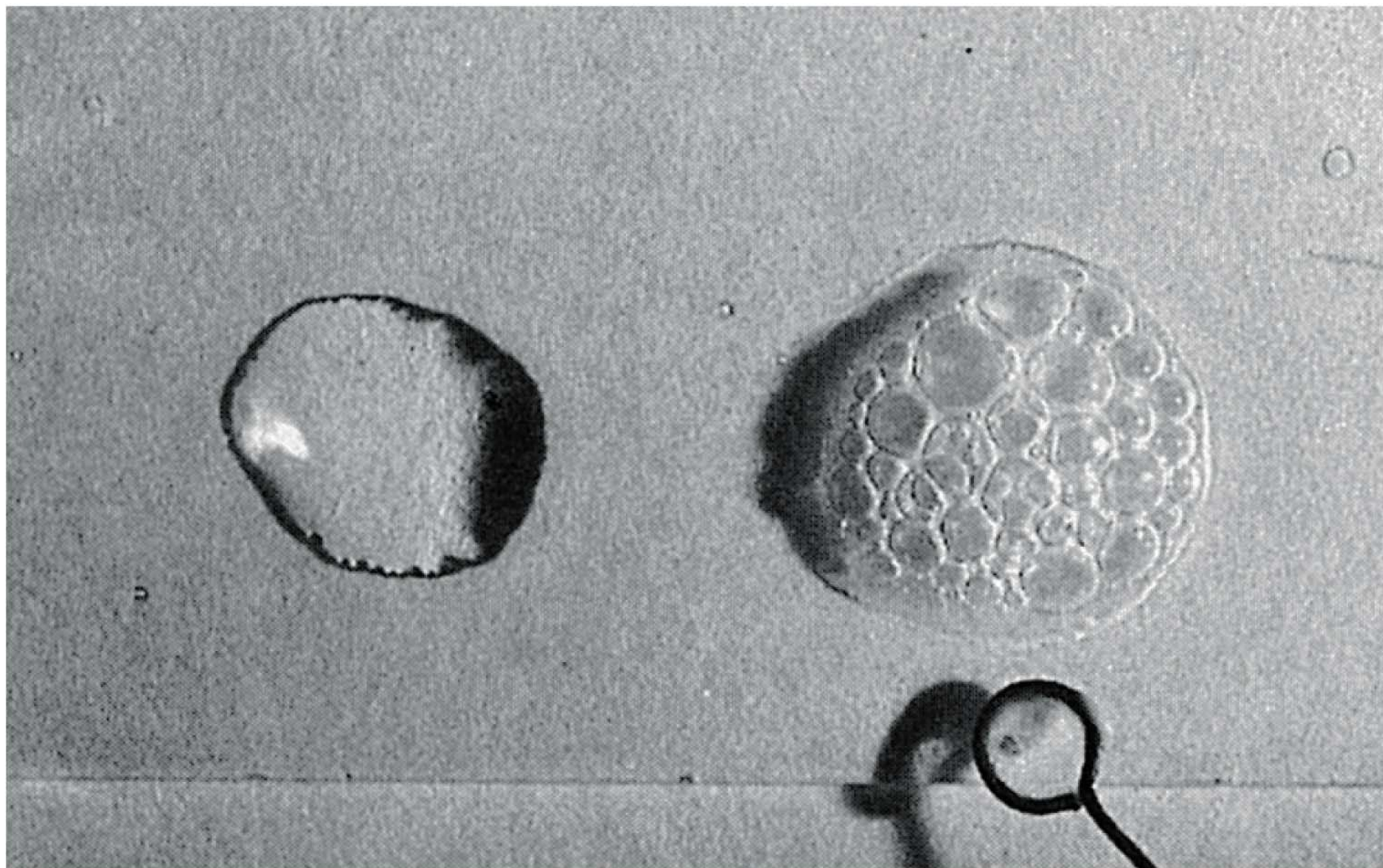
(d) Superoxide dismutase/catalase in combination:



(e) Superoxide reductase:



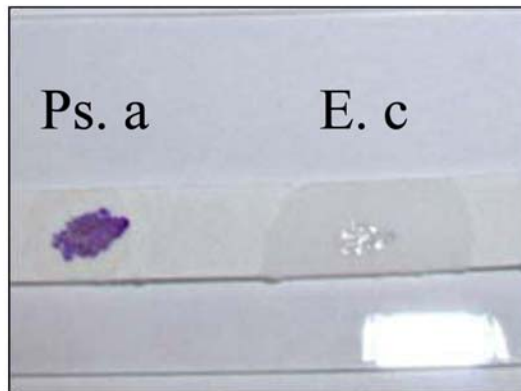
Catalase Test



Cytochrome Oxidase Test



An important diagnostic indicator for the id of *Pseudomonas* and *Neisseria* spp.



Oxidase Test

